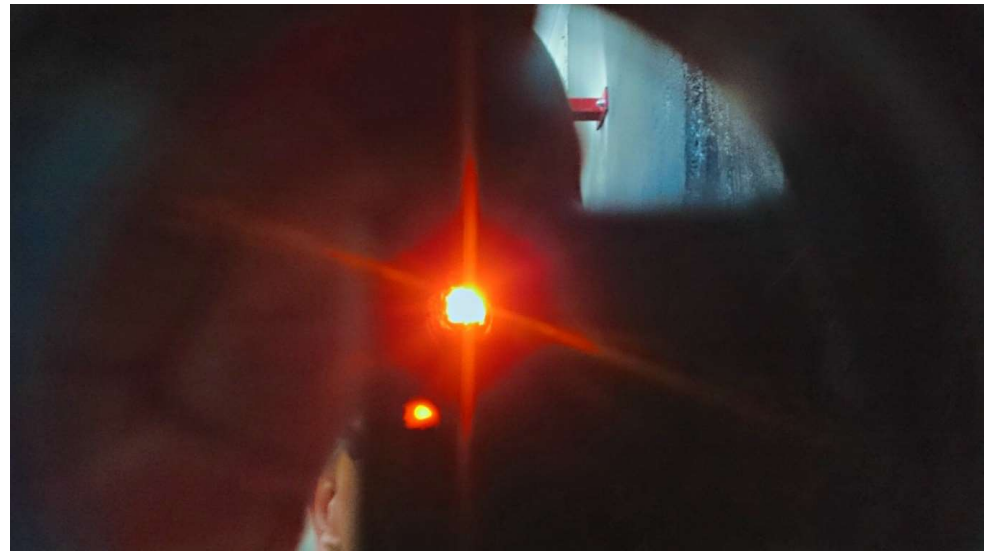


KIT, Karlsruhe, 06-06-2019| Workshop: Gas cleaning, experiences,
new developments, analytics and diagnostics

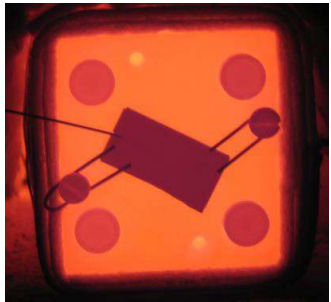


On line UV/IR measurements of tars and other gas compounds

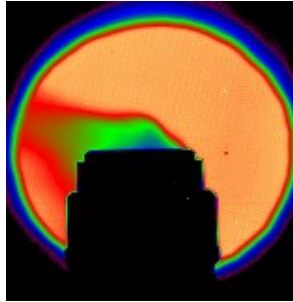
Alexander Fateev, Senior Scientist
DTU Chemical Engineering
e-mail: alfa@kt.dtu.dk



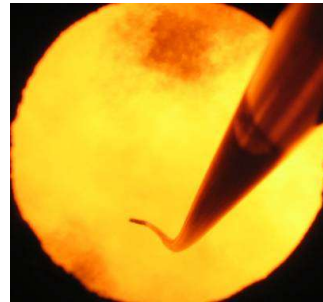
Activities



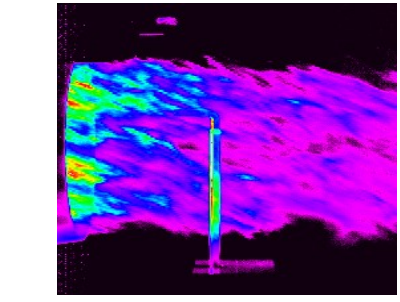
Emission from surfaces
Heat transfer, energy balance in solids



Leak valve bottle
IR gas/particles fast imaging with LED



Extractive/*in situ*/*on line*
Gas composition, (non-contact) gas/particle temperature, RHT modeling in gases



Exhaust aircraft engine
Mixing, flows and properties

At one glance:

- ❑ Two men show: Alexander Fateev and Sønnik Clausen, both Senior Scientists (Risø National Lab, then DTU)
- ❑ A track record in **harsh** industrial measurements: power plants, waste incinerators, ship/aircraft engines
- ❑ A track record in collaboration with industry in DK, SE, DE (energy, bio-medical, environment)
- ❑ Participation in various national and EU-funded projects
- ❑ Instrumentation: gas cells (<1500C), spectrometers: from 120 nm (far UV) to 20 cm⁻¹ (far IR)
- ❑ Spectroscopic (HITRAN/HITEMP/ExoMol) databases validation/development/spectra modeling
- ❑ Sensor development (NG quality (C1-C5), wood stoves (soot, PAH), fast particle imaging with LED's)

Outline

Not fancy at all: broadband Optical Absorption Spectroscopy/DOAS

- ❑ Choice of spectral range: IR vs UV/far-UV
- ❑ Re-born of oldies: VUV as far-UV without vacuum: a practical approach

In the lab:

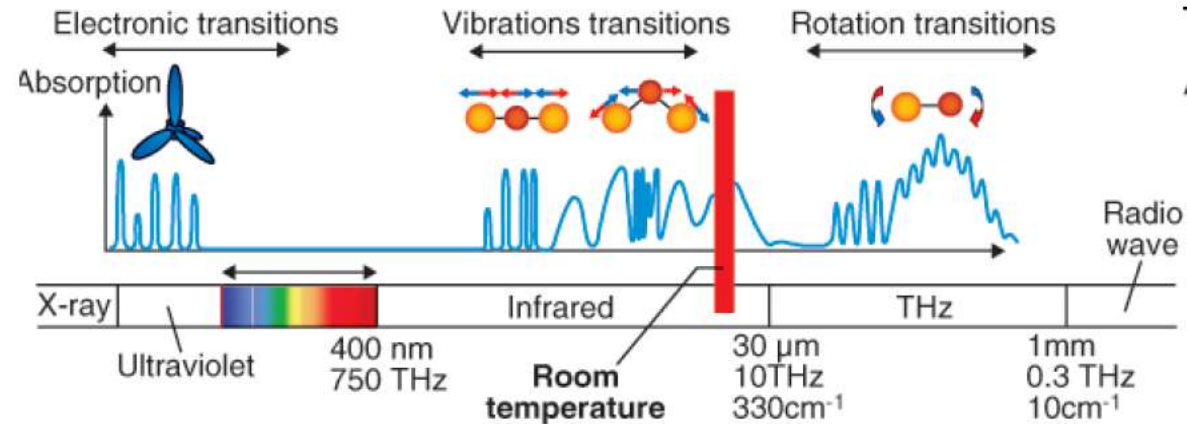
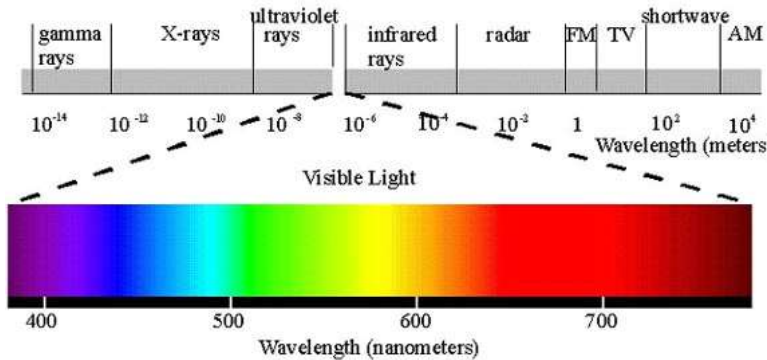
- ❑ gas cells for wide-range of P/T's: 20-1500C/1-200bar
- ❑ temperature-dependent Absorption Cross-Sections Databases or line-lists databases

In the field:

- ❑ From 100kW to 6MW gasification
- ❑ Measurements in stack gases (combustion)

Conclusions

Choice of spectral range: depends on...



IR (MIR: 600-8000 cm^{-1}):

- Classic tool for H₂O, CO₂ and HxCy+
- Databases available (HITRAN, PNNL)
- in situ or on-line measurements

UV (200 nm < λ):

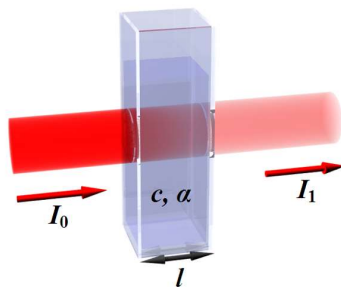
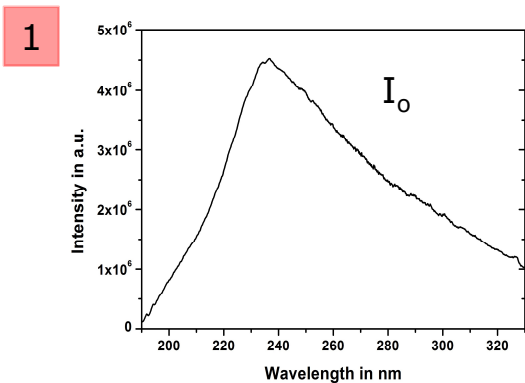
- superb sensitivity for (complex) organics
- (very) strong light absorption
- in situ or on-line measurements

Special for gasification: no O₂

- possibility to go further down ($120\text{nm} < \lambda$): far-UV
- superb sensitivity for major/minor gas components
- compact system
- in situ or on-line measurements

Optical Absorption Spectroscopy/DOAS

Spectrum w/o absorption

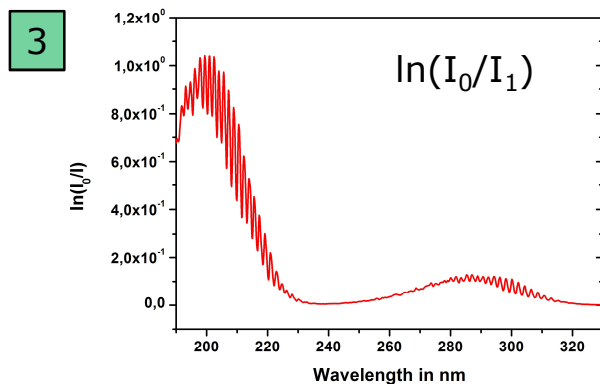
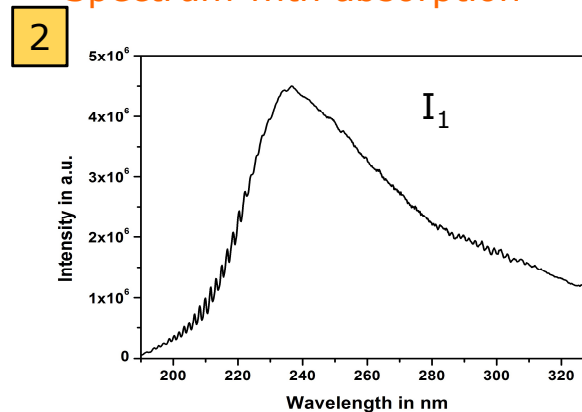


Lambert Beer Law:

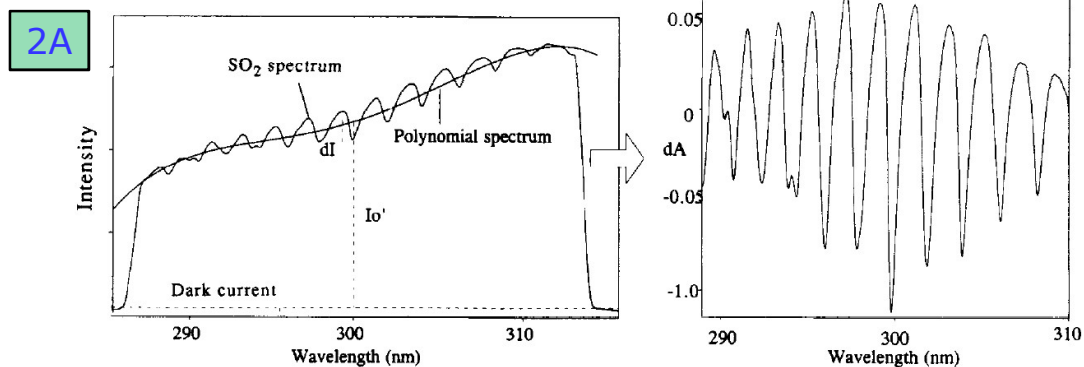
$$\frac{I_1}{I_0} = e^{-\alpha \cdot l} = e^{-\sigma \cdot N \cdot l}$$

$$\ln \frac{I_0}{I_1} = \sigma \cdot N \cdot l$$

Spectrum with absorption

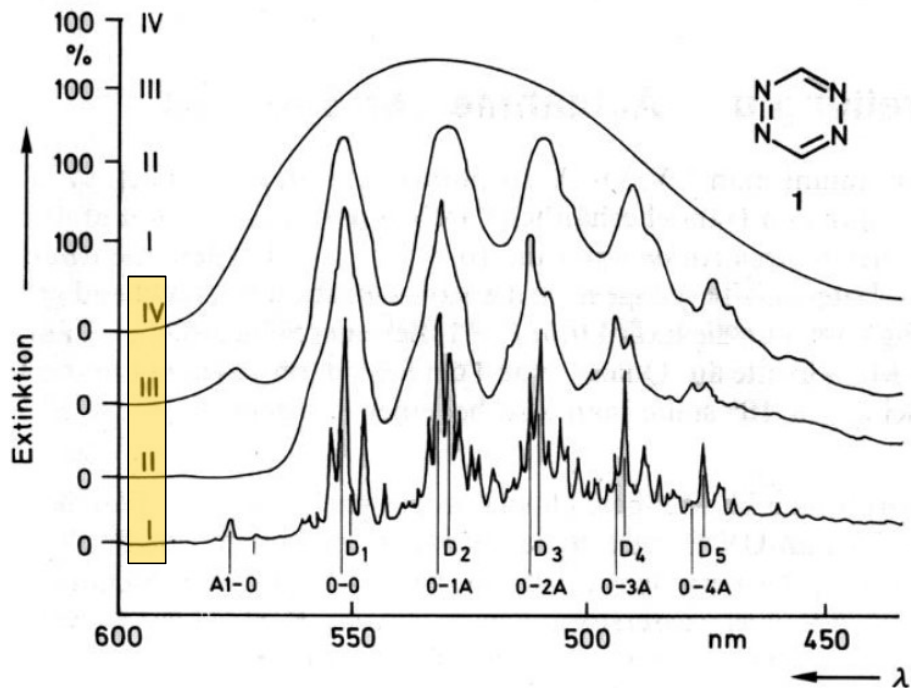


DOAS



Choice of medium

Example: 1,2,4,5-tetrazine



- I Gas phase, room temperature
- II In isopentane-methylcyclohexane matrix, 77K
- III In cyclohexane, room temperature
- IV In water, room temperature

- ❑ Molecules have their own "fingerprints" in IR/UV
- ❑ Vibrational fine structure disappears in solutions but not in the gas phase
- ❑ Fine structure degrades with temperature

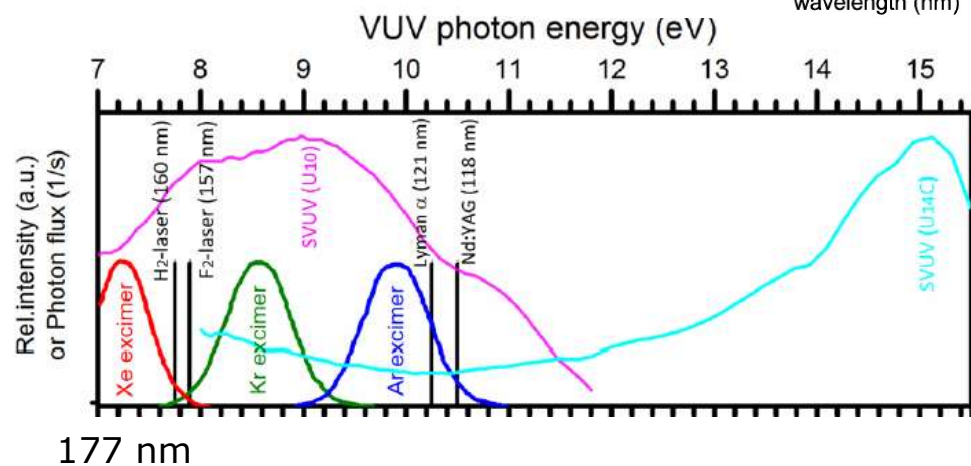
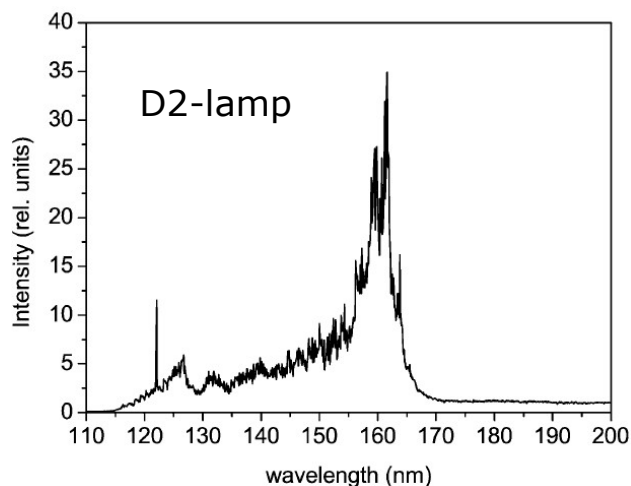
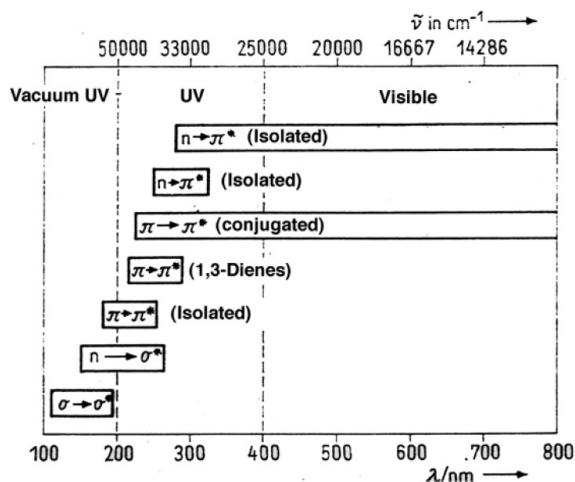
Re-born of oldies: VUV as far-UV

- Classical VUV definition: $\lambda < 200$ nm (when O₂ absorption matters)
 - 4 main “-”:
 - forced use of HV-pump
 - light source availability (synchrotron the best)
 - spectrometer/optics performance drop at $\lambda < 110$ nm
 - windowless system design (i.e. coupled system)

- Far UV definition: 110 nm $< \lambda$ (defined by MgF₂ cut off)
 - 4 main “+”:
 - MgF₂: good robust optical material (H₂O/T) (i.e. de-coupled system)
 - No need for use HV-pumps: N₂ or Ar purge is enough
 - VUV D₂-lamp: affordable light-source with good *costs:performance* ratio
 - transportable system (flight-case scenario) for lab/field measurements

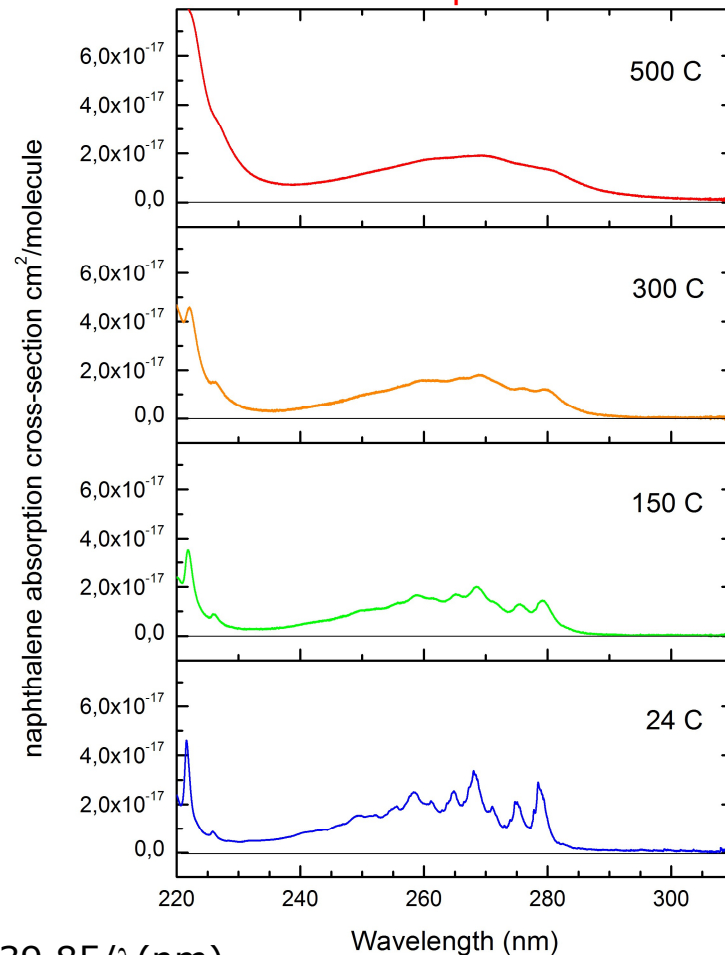
- Rydberg state spectroscopy below ionization limit
 - large absorption cs (i.e. short absorption pathlengths)

Re-born of oldies: VUV as far-UV in pictures



8

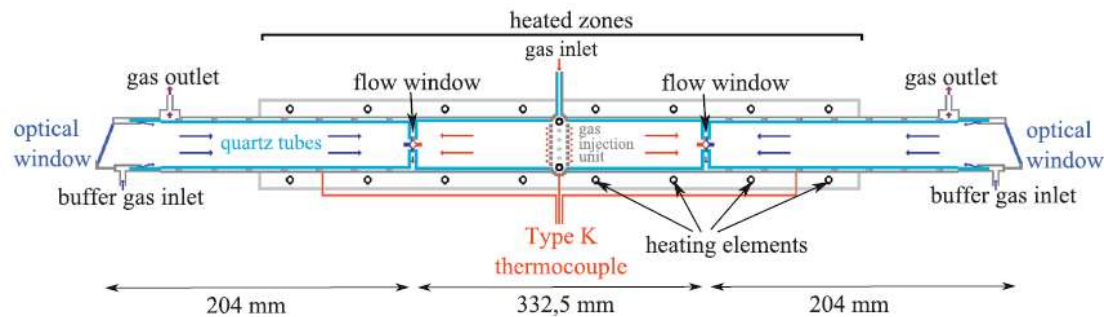
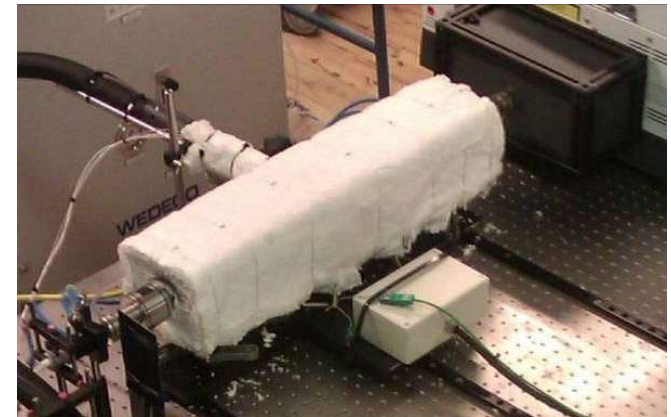
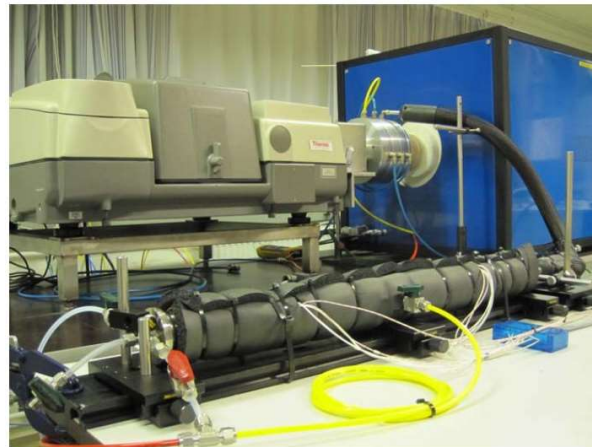
T-effects on spectra



$$E(\text{eV}) = 1239.85/\lambda(\text{nm})$$

Experimental facilities

- ✓ High Temperature Gas Cell (max 1600C): CO, O₂, NO, CO₂, H₂O, CH₄,...
- ✓ Quartz Gas Cell for reactive gases (max 525C): NH₃, SO₃, SO₂, PAH (phenol, naphthalene),...
- ✓ 0.39 cm, 5.1 cm (max 300C), 50cm, 1m, 5.7 m (max 200C) gas cells for lab/field measurements

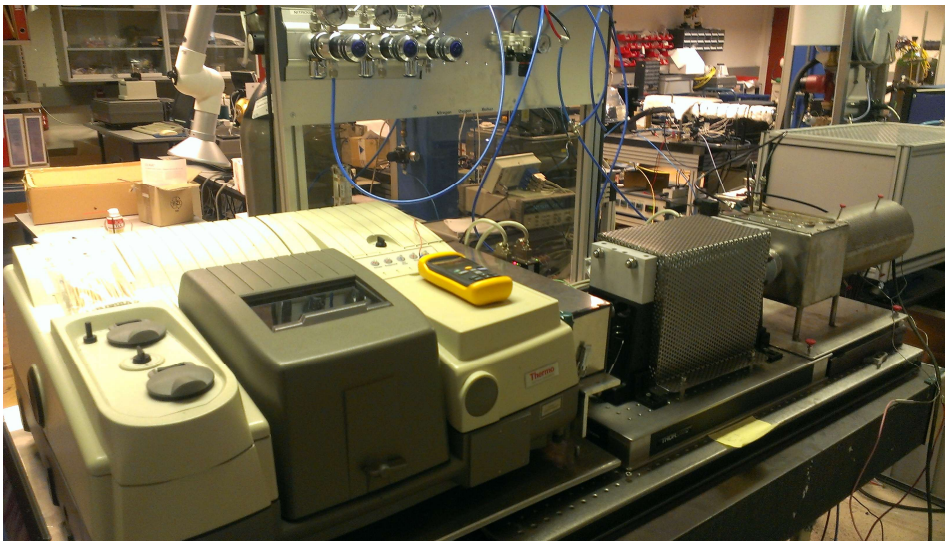


Replaceable outer windows:

- ❑ no reactive gas contact with windows
- ❑ far UV to μw coverage

Experimental facilities: special case

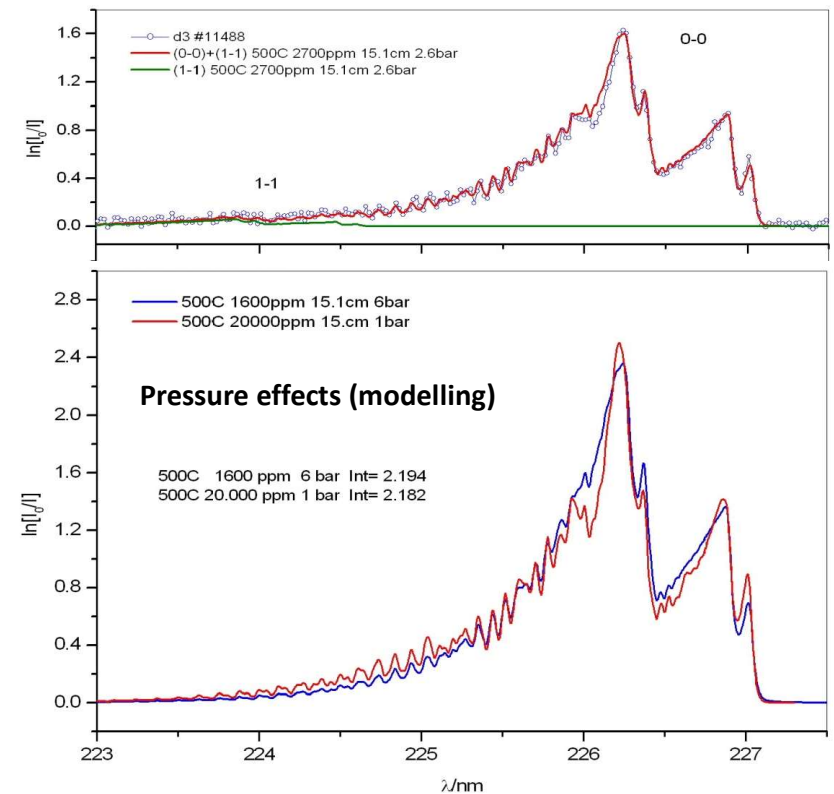
- high-pressure (up to 1000°C)/high pressure (up to 200 bar) flow gas cell



- ❑ sapphire windows:
- ❑ far UV to mid IR: CO₂, CO, O₂, NO, H₂O

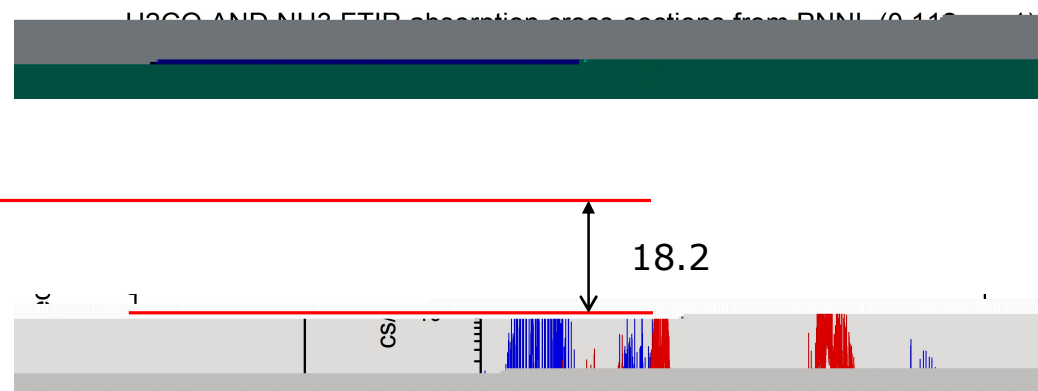
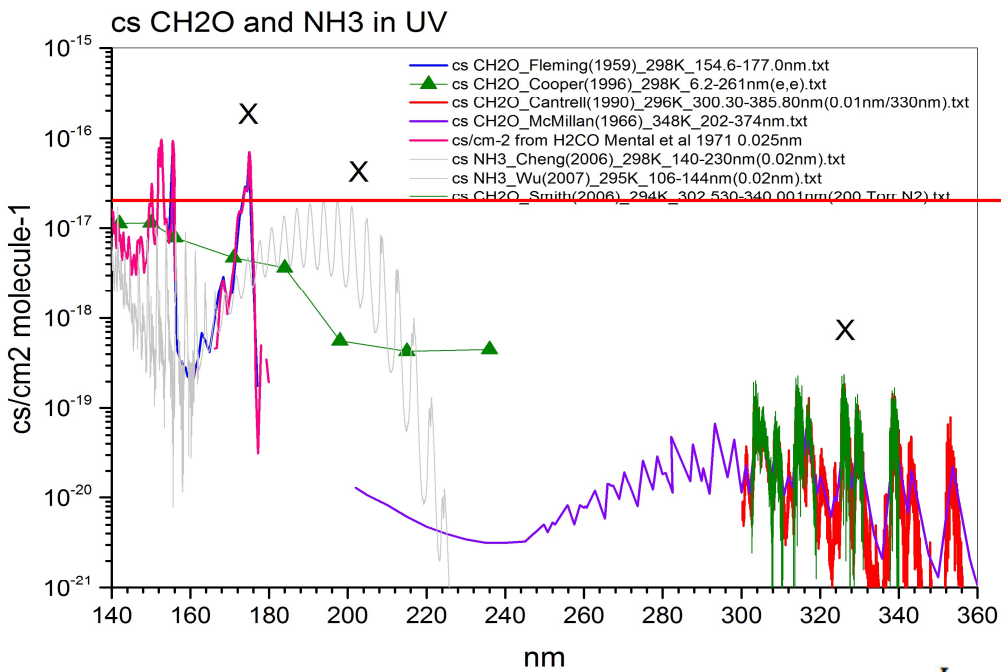
UV NO band:

fast *in situ* measurements (blue) vs. modelling (red)



Spectroscopy of NH3 and H2CO

What shall I choose: far UV, UV or IR?



$$\ln \frac{I_0}{I_1} = \sigma \cdot N \cdot l$$

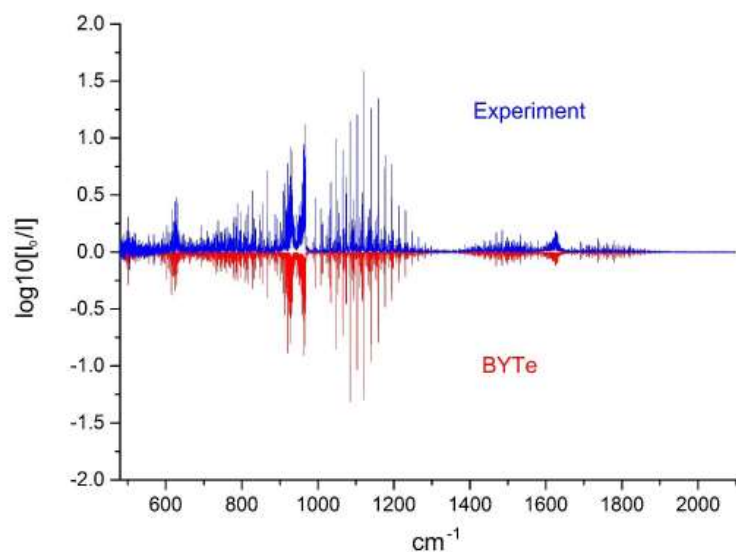
l in cm/mm or meters?

IR spectroscopy of NH₃

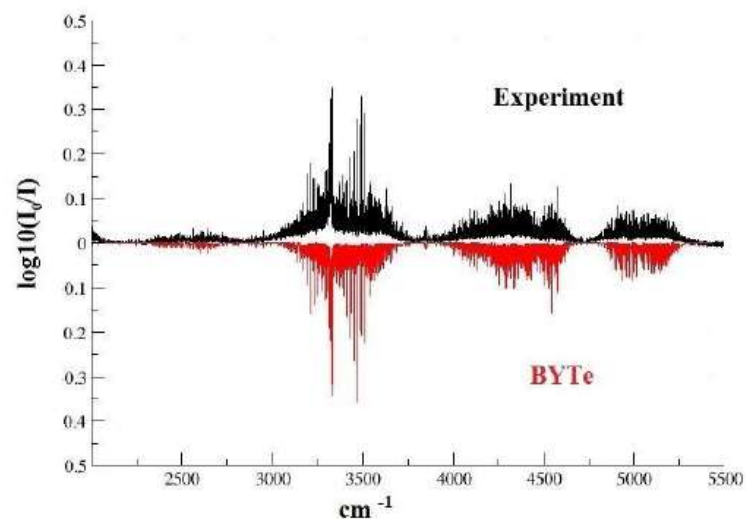
ExoMol Modelling High Resolution Absorption Spectra
with ExoMol Line Lists: NH₃ and CH₄
E. J. Barton¹, S. N. Yurchenko¹, J. Tennyson¹, S. Clausen² and A. Fateev²
¹University College London, Gower Street, London WC1E 6BT
²Technical University of Denmark (DTU), Frederiksborgvej 399, 4000 Roskilde, Denmark



500 - 2100 cm⁻¹ 500⁰C NH₃ = 1%

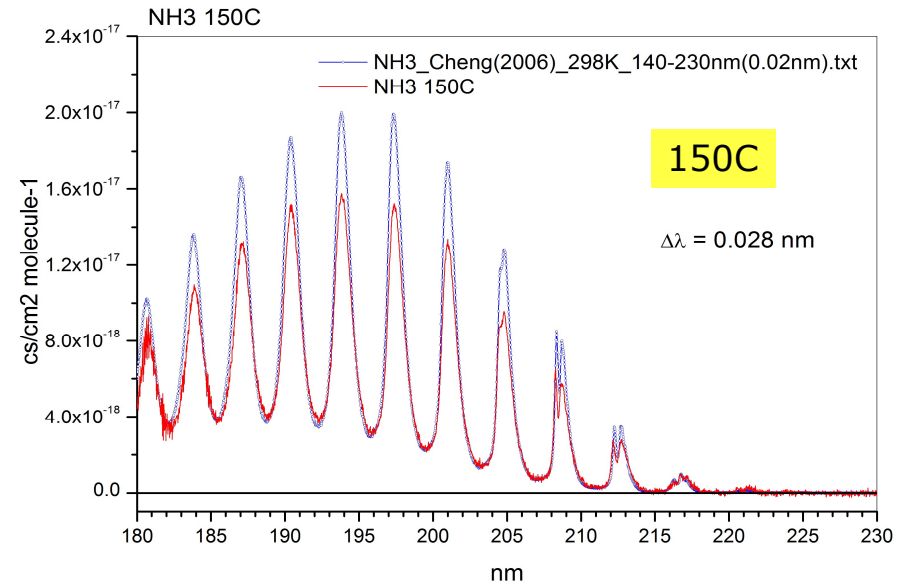
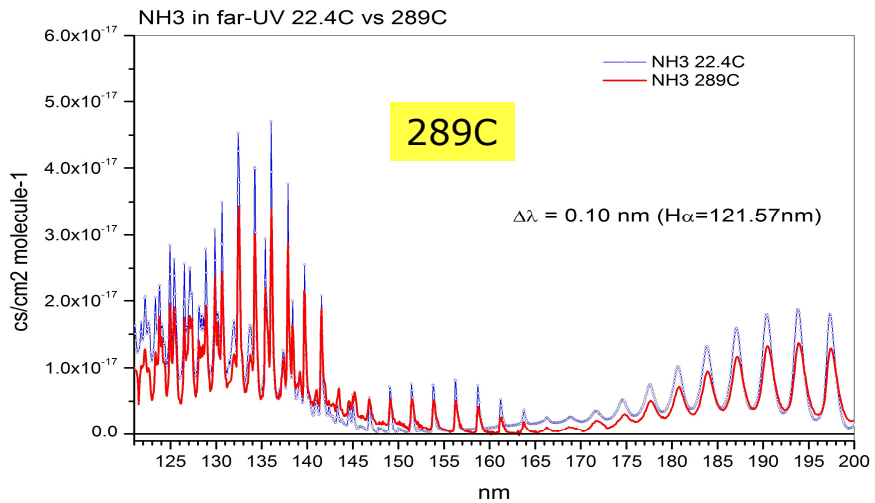
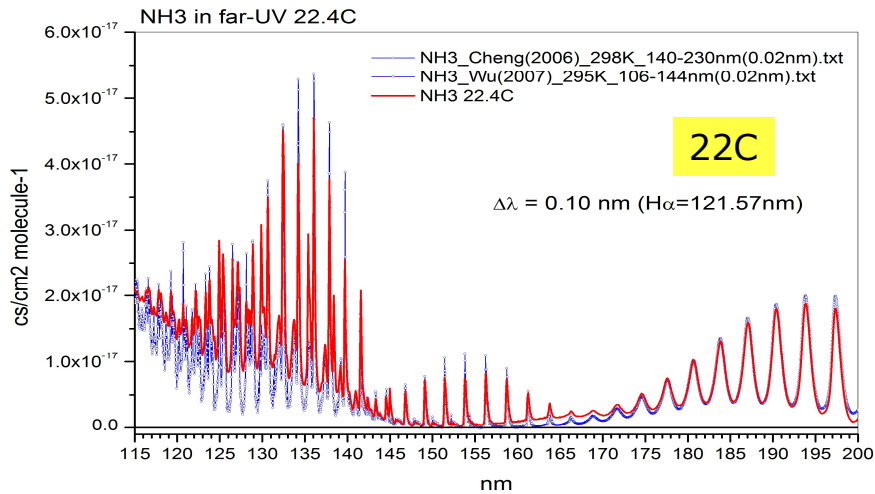


2100 - 5500 cm⁻¹ 1027⁰C NH₃ = 10%



- ❑ Updated NH₃ line-list database available at ExoMol web site
- ❑ Can be used for NH₃ spectra modeling up to 1500C

Far-UV spectroscopy of NH3



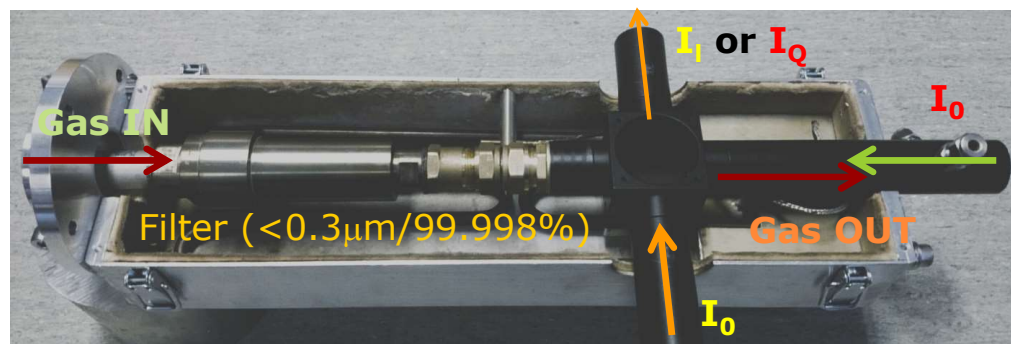
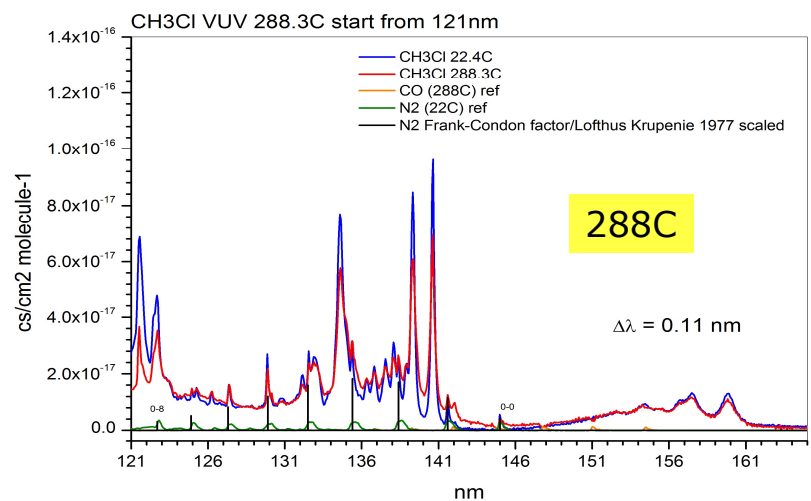
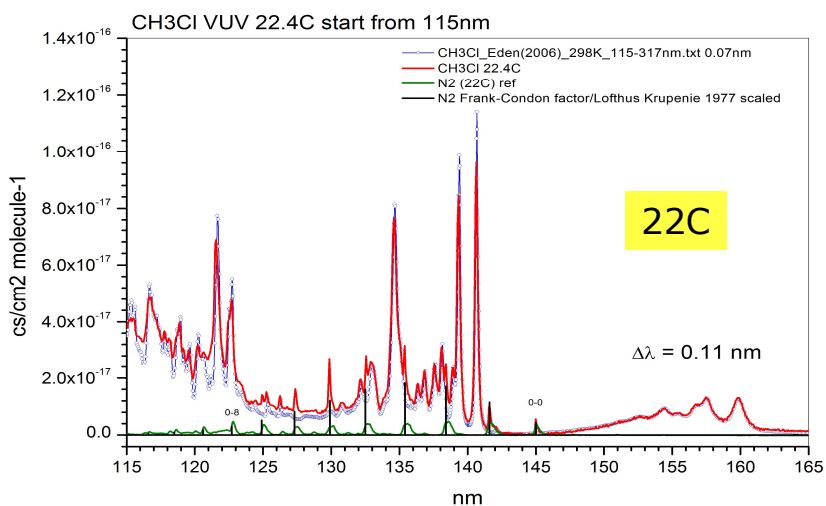
- ❑ Difficult molecule to deal with
- ❑ Most reliable data: $180 < \lambda$

NH3 spectra analysis:

Wo et al (2007): 110-144 nm (synchrotron)

Cheng et al (2006): 140-220 nm (synchrotron)

Far-UV spectroscopy of CH₃Cl and C₂H₄



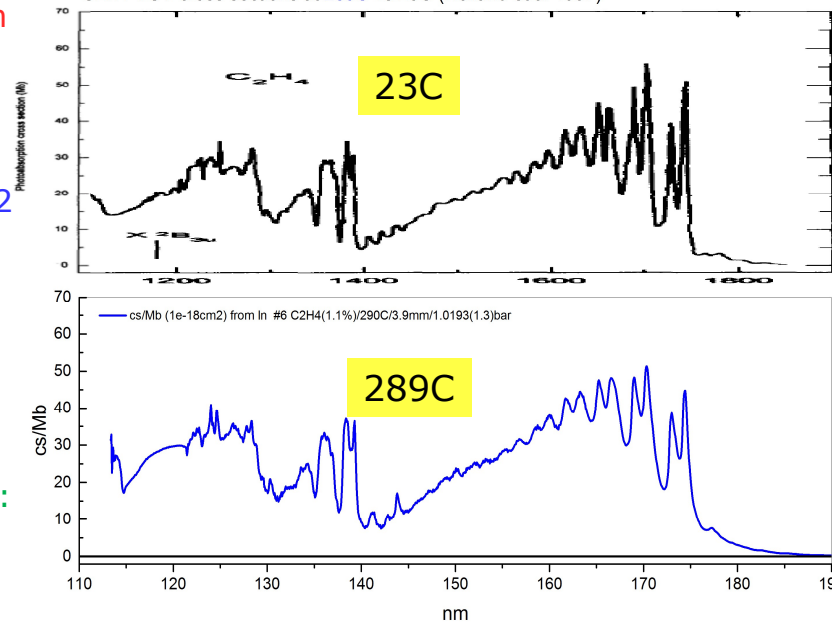
Measurements on 0.39 cm gas cell (up to 300C)
(λ -calibration with NH₃ spectrum)

Lower end limited by MgF₂ cut off (temperature depended)

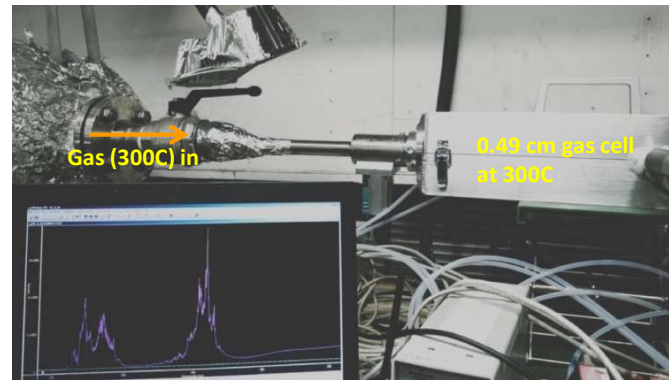
Detailed CH₃Cl spectra analysis (320-115 nm):
Eden et al (2007)
(synchrotron)

Ethylene spectra analysis :
Holland et al (1997)
(synchrotron)

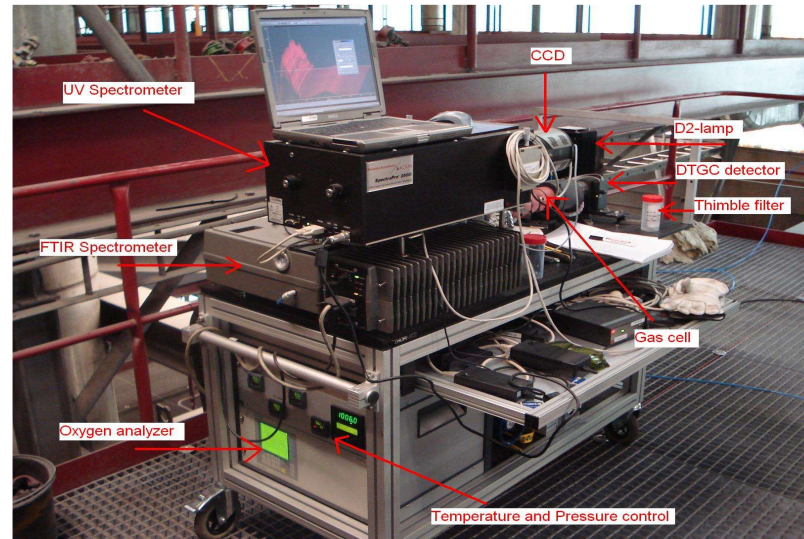
C₂H₄ VUV cross sections at 289C vs 23C (Holland et al 1997)



Field measurements: **on-line** vs **in situ**

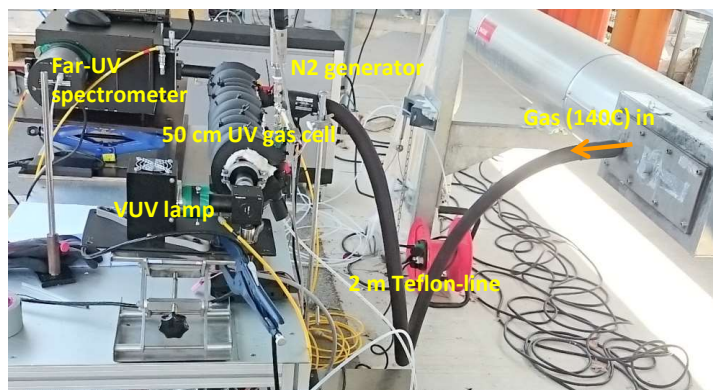


- ❑ UV/IR Light through
- ❑ Ti-probe at 300C, gas cell at 300C (far-UV)
- ❑ No tar condensation issues



- ❑ Removing heavy tar with tar trapper
- ❑ UV/IR dedicated measurements at 150C: only limited by optics (windows)

Field measurements: NH3 in stack gas



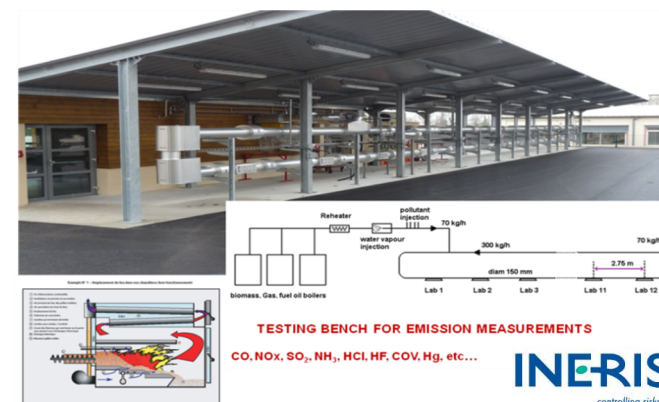
Combustion cases:

- natural gas (EL, power)
- biomass (wood pellets, heating)
- diesel (cars)

On-line measurements at 150C

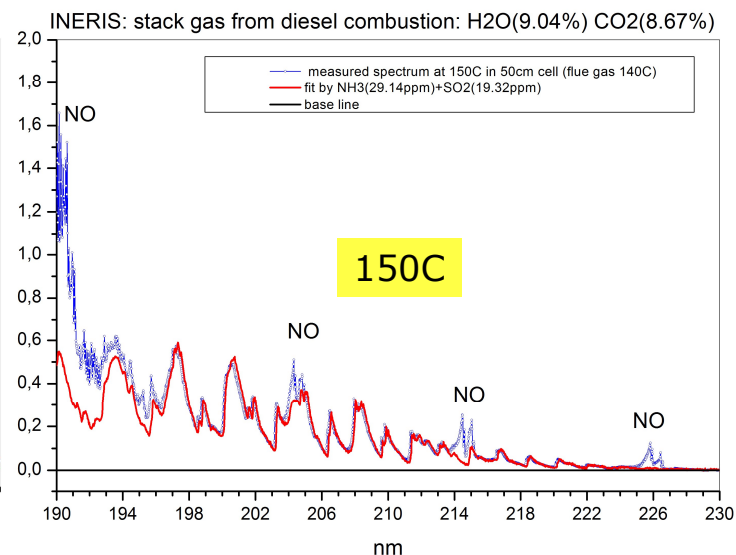
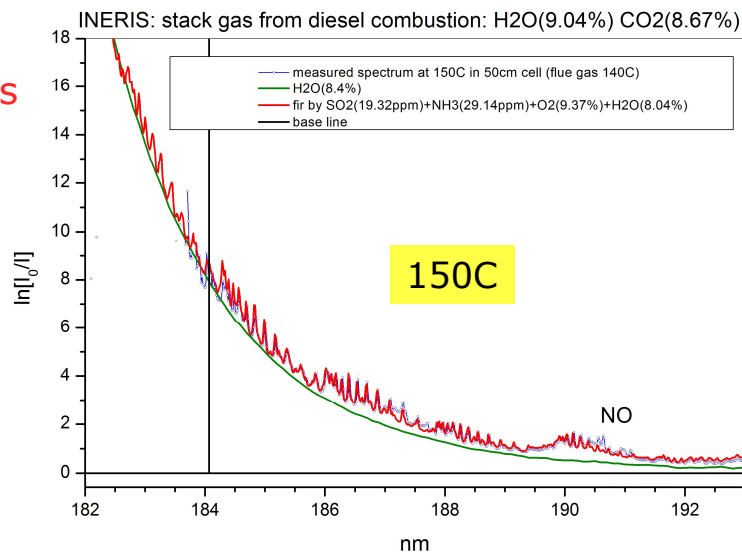
- above stack gas temperature 130-140C

Relevant for gas cleaning process

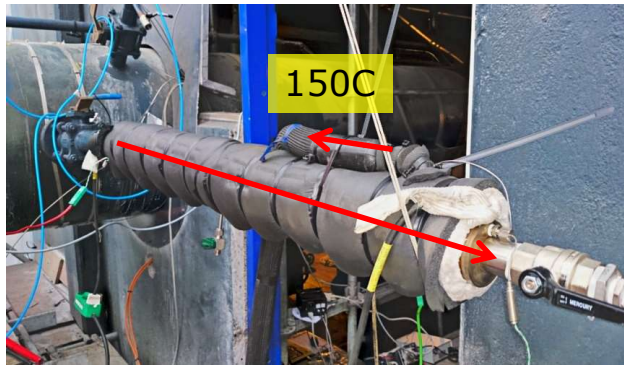


- ✓ In situ/on line measurements
- ✓ Limited by H2O (50cm)
- ✓ Excellent sensitivity to **NH3, SO2, O2, H2O, NO**

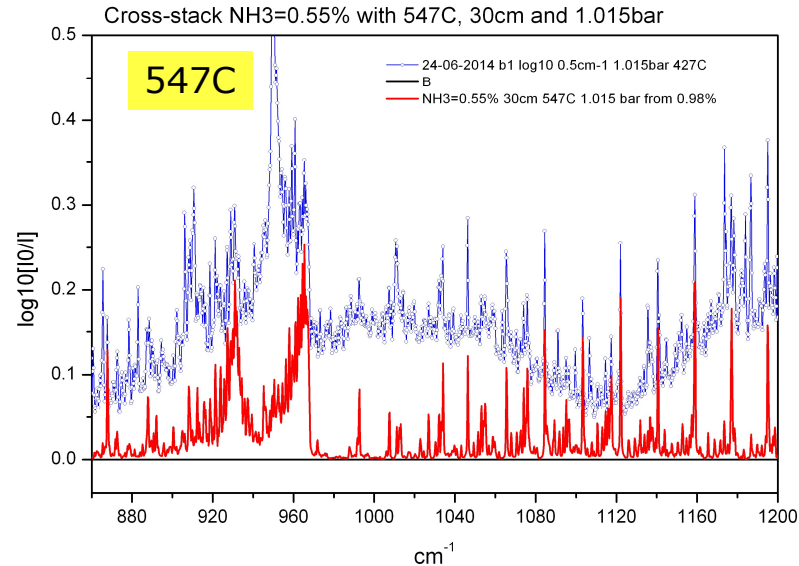
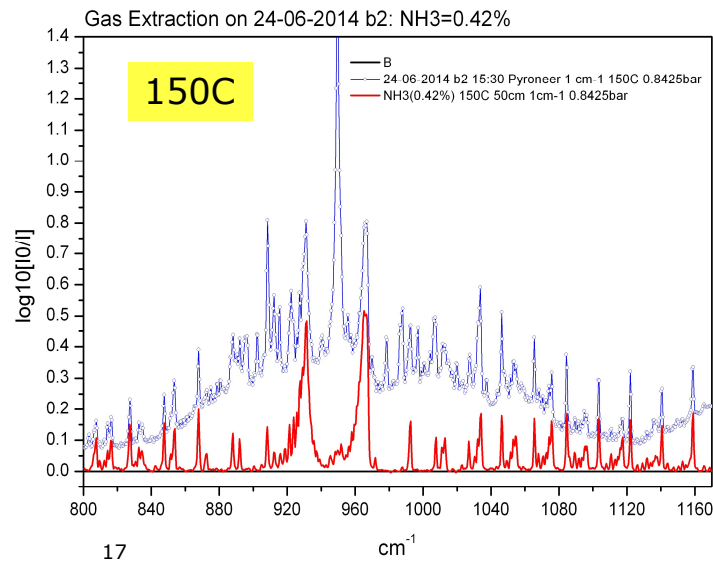
IMPRESS 2: Metrology for Air Pollutant Emissions



Field measurements: 6MW demonstration plant



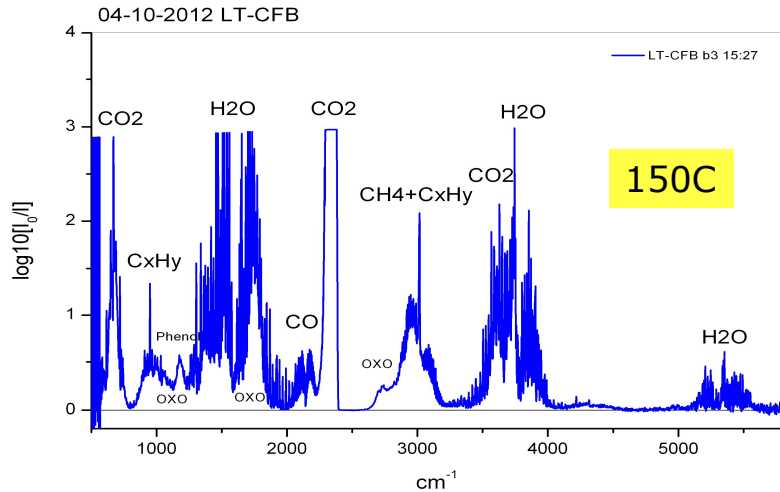
- Straw gasification
- Gas feed into burner
- Target: NH₃, HCN, NO and H₂O
- In situ (cross-stack) and gas extraction: same H₂O



- NH₃ extraction: ca. 24% less than in situ
- NH₃ trapping by condensing acids/tars

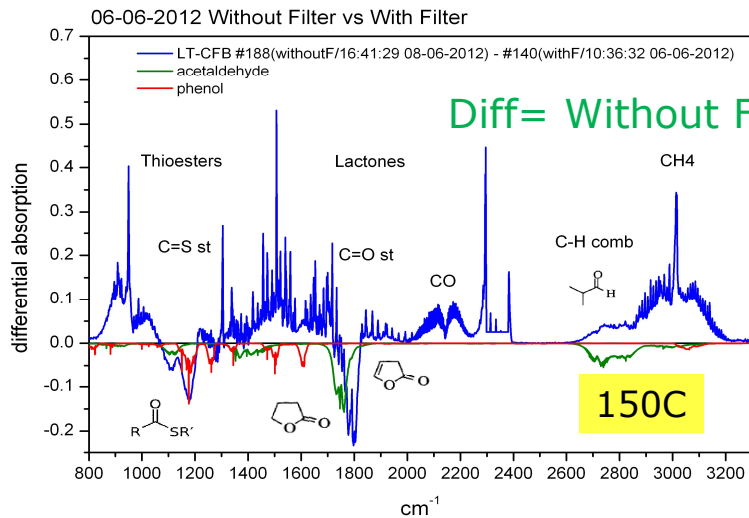


Field measurements: LT-CFB (LT-gasification, 100kW) DTU



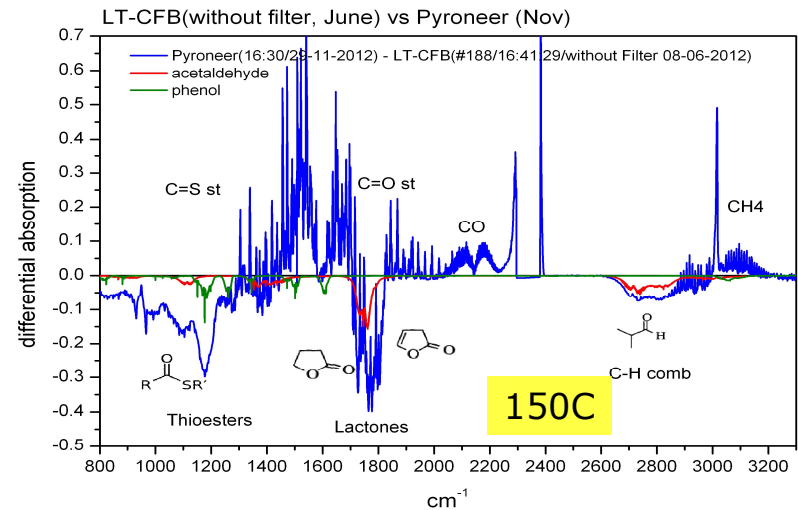
- ❑ Straw gasification
- ❑ Gasifier with (330C) and without (504C) particle filter: "base line" check (=process monitoring)
- ❑ thioesters and lactones: precursors of more simple HC
- ❑ Higher T → cleaner gas

Diff= 6MW(547C) -100kW(504C)

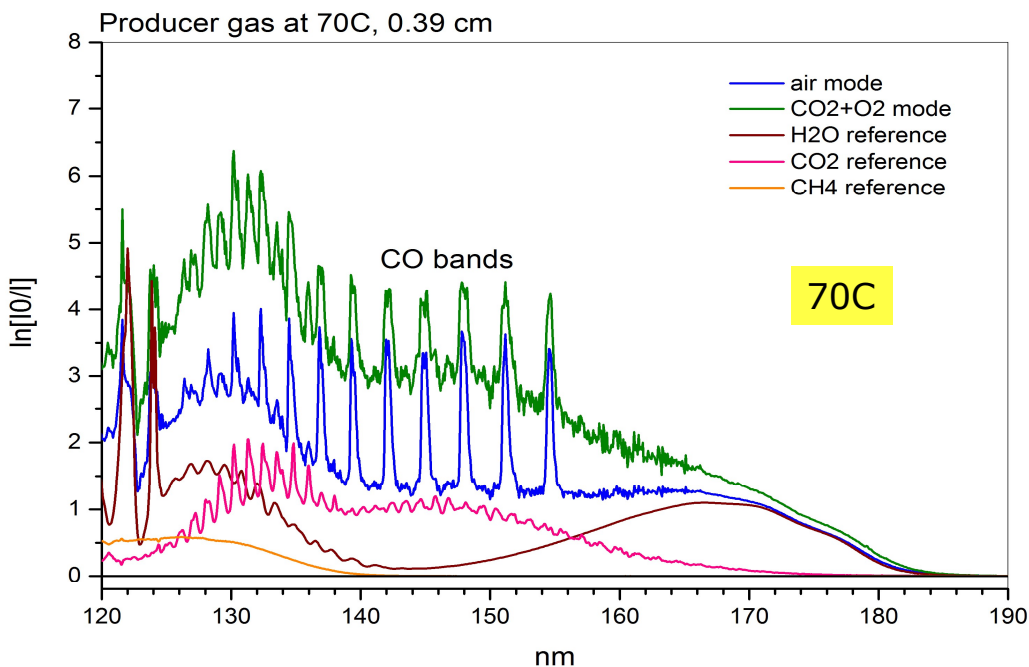


Diff= Without F(504C) -with F(310C)

Process monitoring: IR



Field measurements: Viking (HT-gasification)



Very clean gas

Traces of NH₃ and other organics

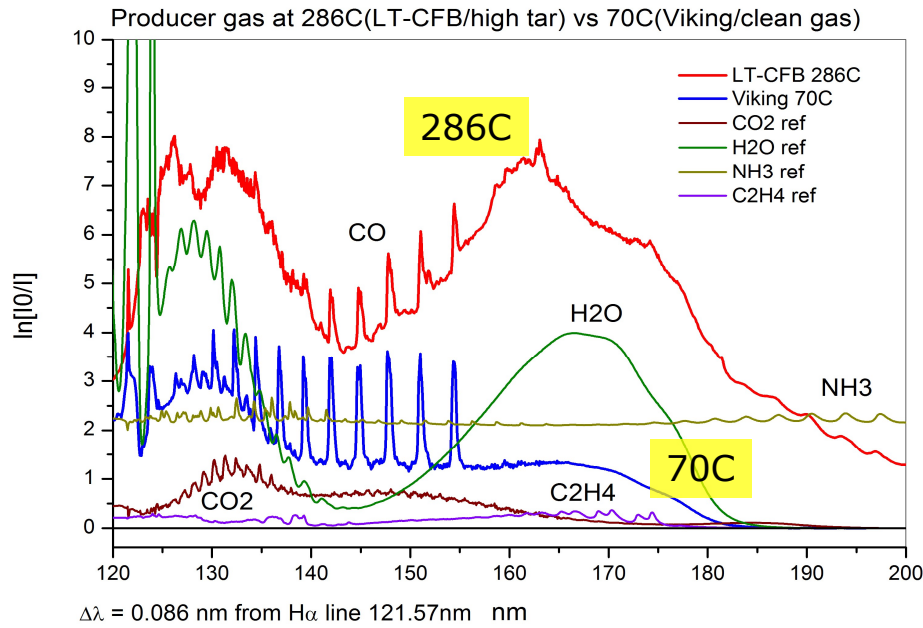
*) Aldehydes in CH₃CHO equivalent

***) below detection limit

19 ***) concentrations calculated from spectra measured over 10 min measurement time

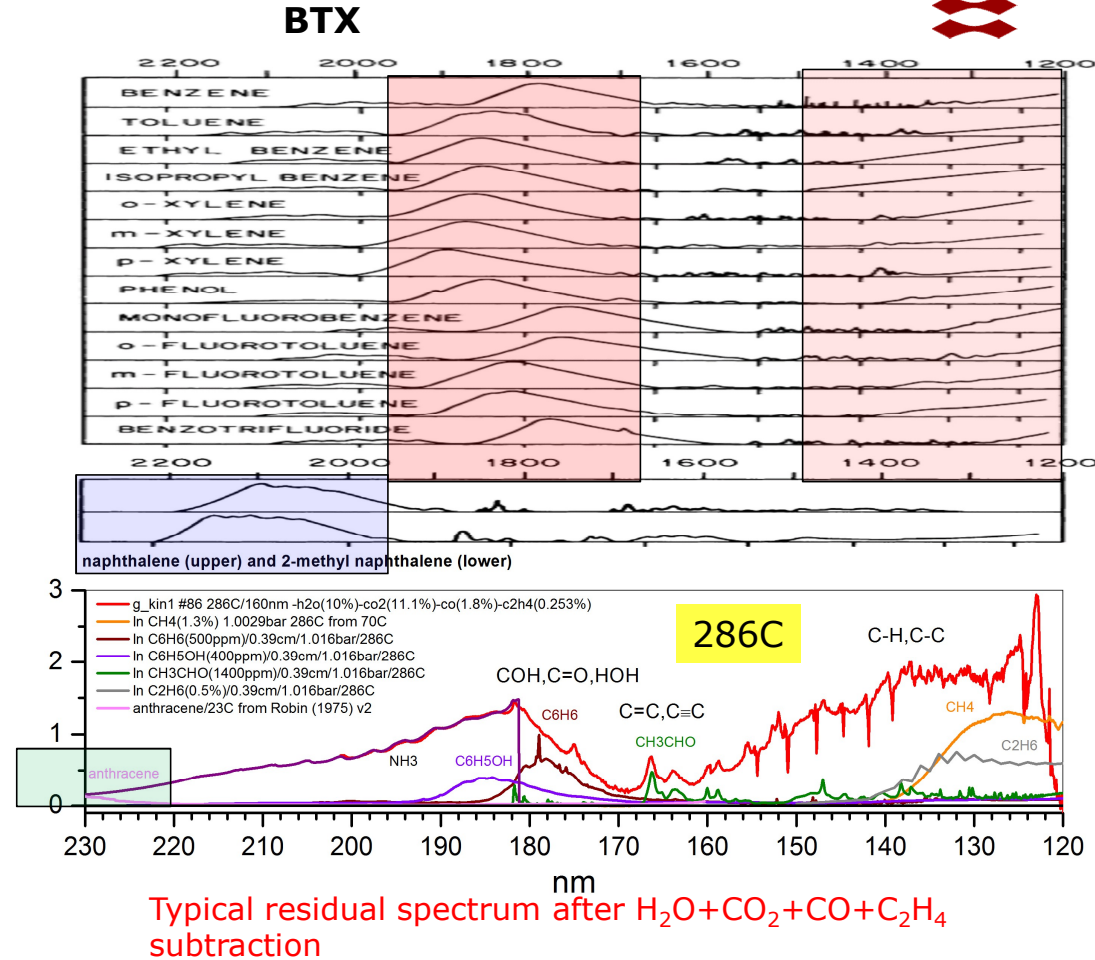
Viking	Air ***	O ₂ -CO ₂ 1st run ***	O ₂ -CO ₂ 2nd run***
CH ₄	0.433%	0.866%	1.028%
CO ₂	12.2%	31.08%	24.42%
H ₂ O	2.74%	2.82%	2.8%
O ₂	0.354%	0.885%	0.955%
CO	8%	14%	14%
N ₂	77%	0%**	0%**
NH ₃	33ppm	0ppm**	0ppm**
C ₆ H ₆	0ppm**	22ppm	22ppm
CH ₃ CHO*	0ppm**	100ppm	100ppm
OCS, CH ₃ Cl, HCl	0ppm**	0ppm**	0ppm**

Field measurements: LT-CFB (LT-gasification, 100kW)

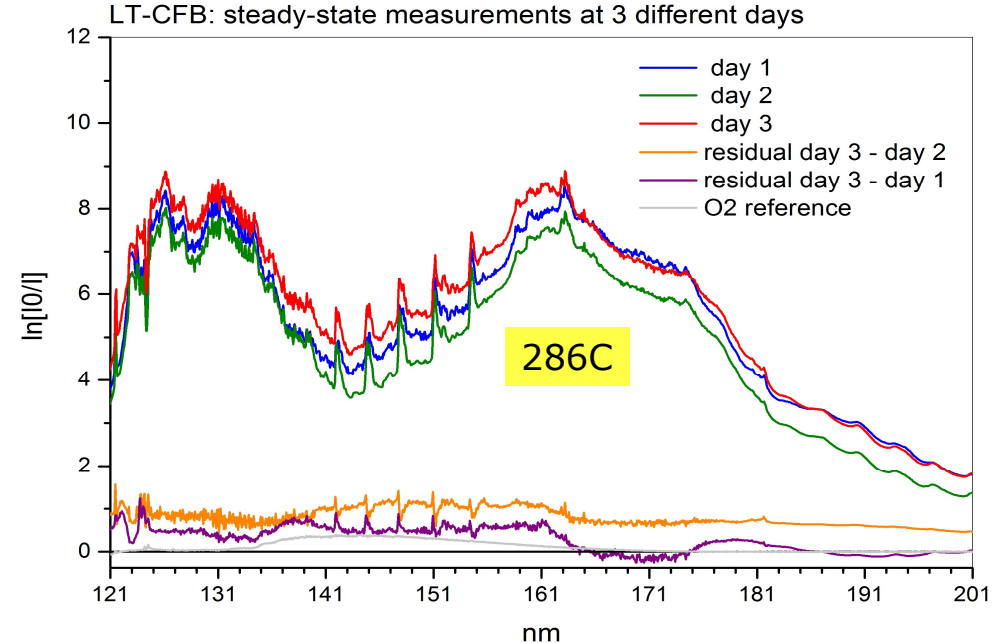
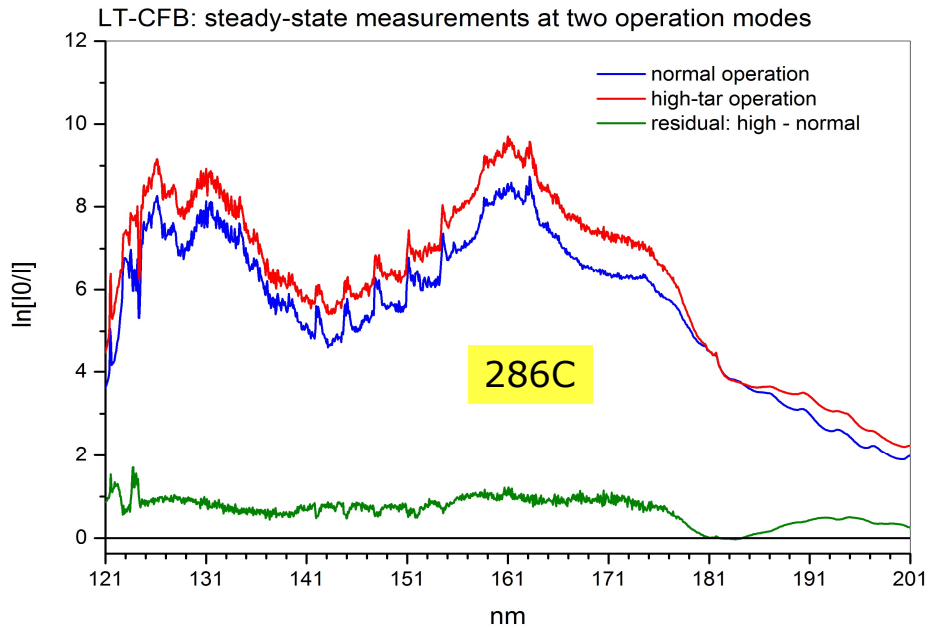


Tar (PAH) contribution to the total absorption:

- High: BTX (170-190nm)
- Medium: "light tar" (naphthalene+) (190-220nm)
- Low: "heavy tar" (anthracene+) ($\lambda > 220 \text{ nm}$)



Field measurements: LT-CFB (LT-gasification, 100kW)



Process monitoring: far UV

- "high-tar" operation mode: less BTX(181 nm), more tar (<170nm) and less NH3(185-200nm)
- "normal" operation mode: no change in BTX and minor O₂ changes in the gas over few days

Conclusions

- Broad-band absorption spectroscopy is a powerful and robust tool for tar and major/minor gas components *in situ* and *on line* measurements and process monitoring: **buy all for 1 price concept**
- T-dependent cross section databases can be generated on request (support instrument/sensor development)
- **Successful demonstration of in situ/on line approaches in measurements in various environments (low/high temperature gasification and combustion)**
- Far-UV (160nm<) can be used for tar/BTX in situ measurements (absolute or relative) by simple weighting of the 195-230 nm and 170-200 nm areas under an absorption spectrum
- **Ability unexpensive far UV small spectrometers opens possibility for a new in situ tar/BTX sensor development when a complex tar/BTX sampling can be avoided.**

Acknowledgments



- Energinet.dk: projects No. 2013-12027 and 2011-1-10622
- The work partly has received funding from the EMPIR programme co-financed by the Participating States and from the European Union's Horizon 2020 research and innovation programme

IMPRESS 2: **Metrology for
Air Pollutant Emissions**

Thank You

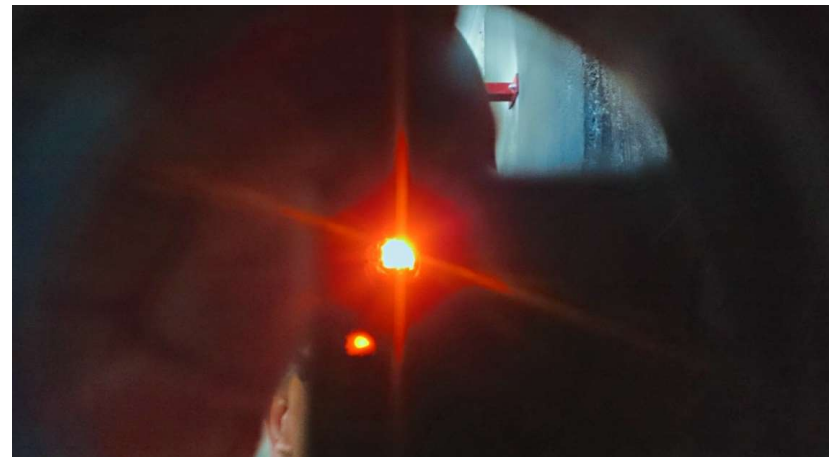
Questions? Comments?

alfa@kt.dtu.dk

+45 23 65 2906



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