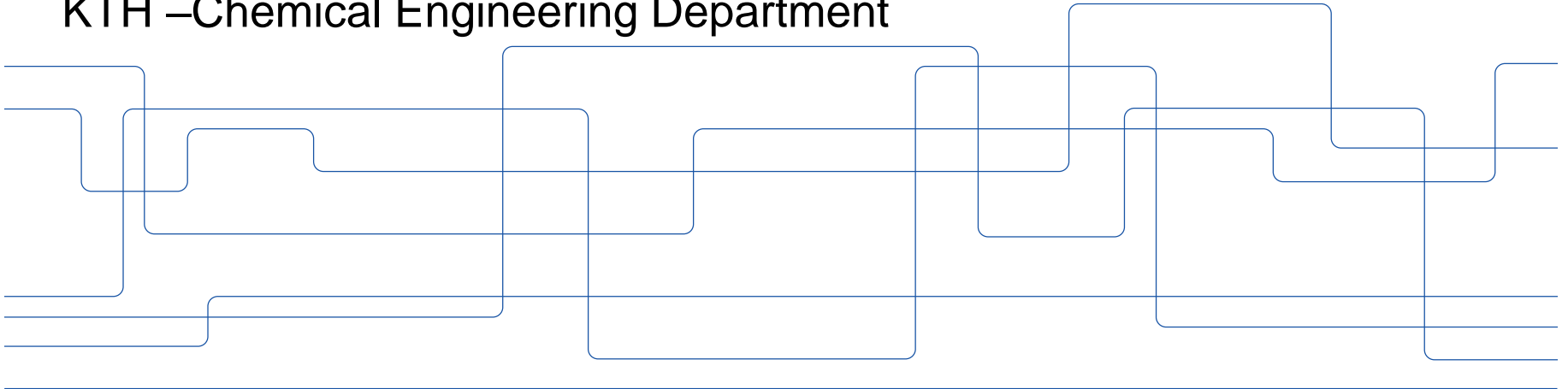


Effects of gas phase K and H₂S on catalytic tar reforming using commercial Ni catalysts downstream a biomass gasifier

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It is teamwork...



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- K. Andersson (Haldor Topsoe)
- P. Moud (KTH)
- J. Pettersson (Gothenburg University)
- R. Lanza (KTH)
- A. Hernandez (KTH- La Sapienza)
- Y. Ge (Gothenburg University)
- E. Kantarelis (KTH)



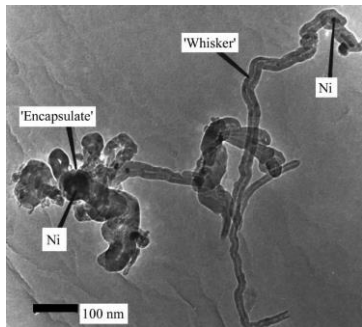
HALDOR TOPSØE 



Coking of tar reforming catalysts

Coking can be controlled by controlling:

- Increased steam/carbon ratios in the feed.
- Ensemble size control on available nickel surfaces (e.g. doping with traces of sulfur or with small amounts of metals that concentrate on the nickel surface)

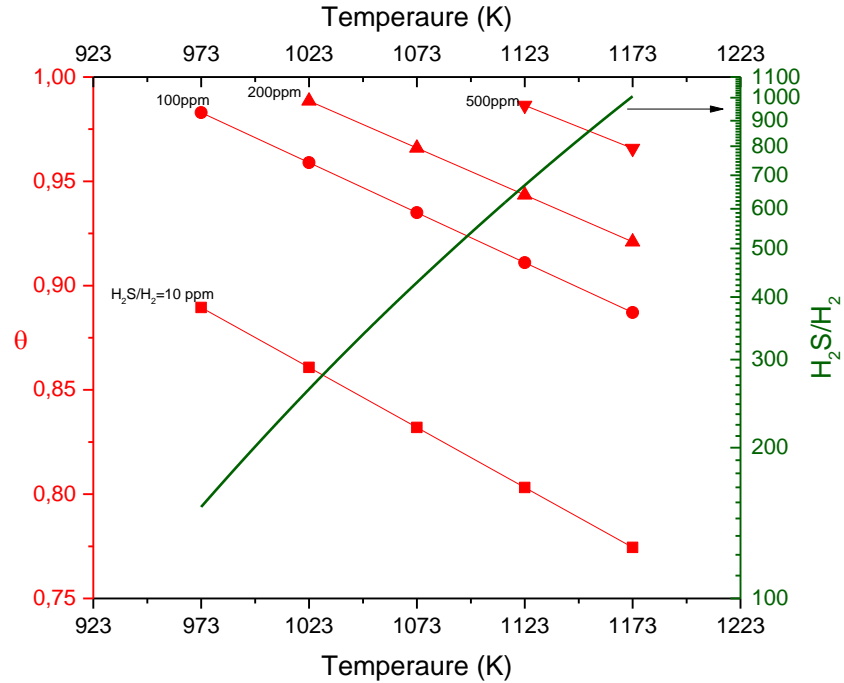


Catalysis Reviews 49(4):511-560

Research Question(s)

- How could we control carbon formation and catalyst deactivation by utilizing biomass gasification impurities (H_2S & K) ?

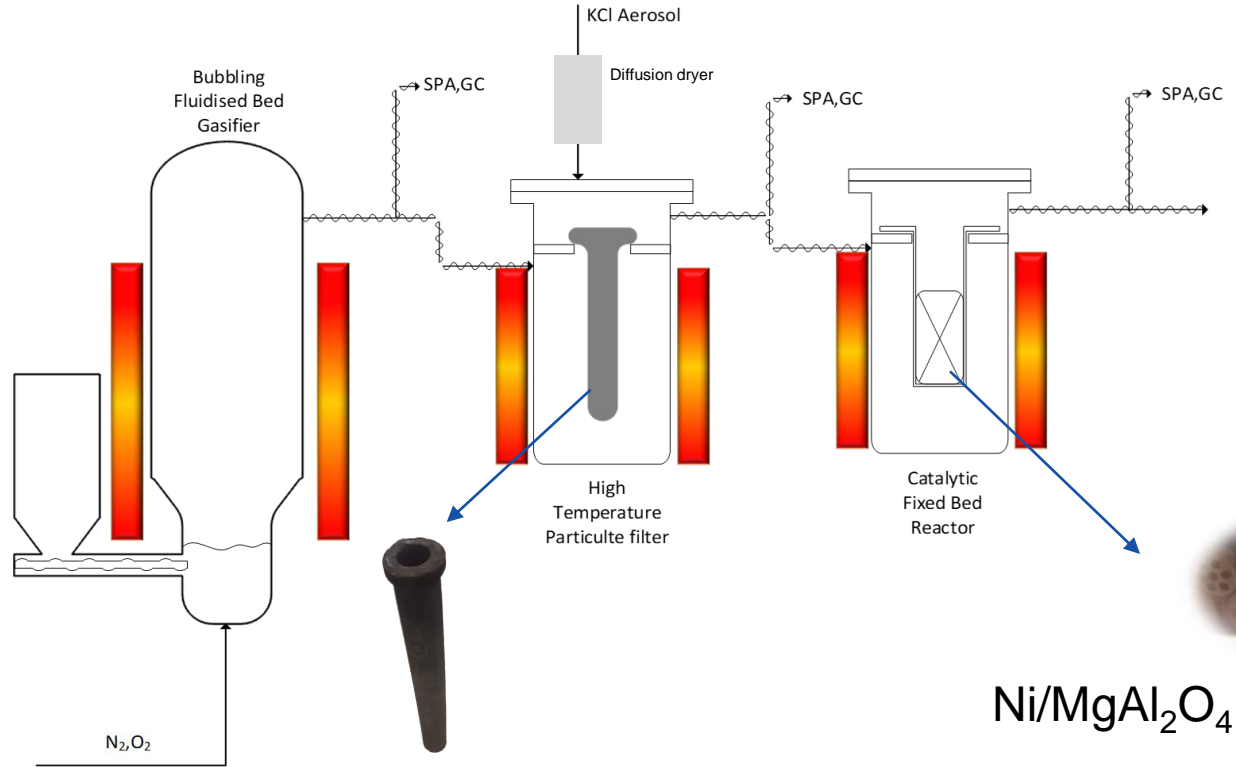
H₂S sorption on Ni



Potassium effects

- Contradictory literature results of gas phase K effects on methane and tar reforming activity
- Deposition mechanisms (impregnation) not realistic, not similar and sometimes not reproducible
- Transient phenomena have not been screened out

Experimental Setup



Experimental conditions and materials

Pine Pellets 3mm

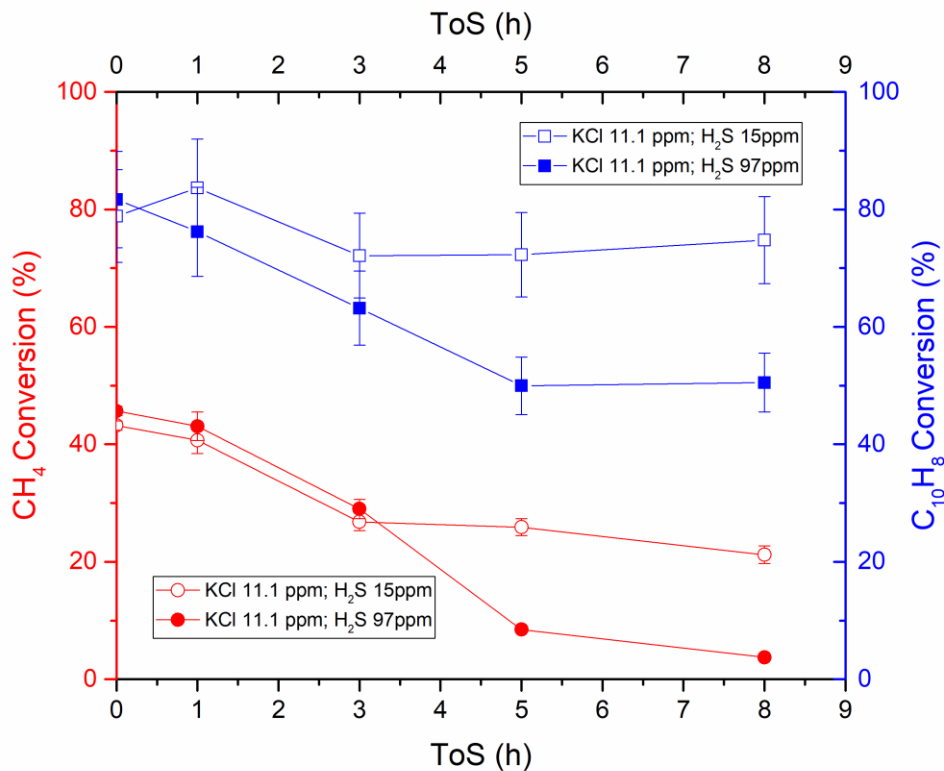
Ash(wt% db)	0.5
VM(wt% db)	76.5
C (wt% db)	47.7
H (wt% db)	6.3
N (wt% db)	0.16
O (wt% db)	45.3
S(wt% db)	<0.012
Cl(wt% db)	0.03
K (mg/kg db)	639
Na (mg/kg db)	57.3

Operating Conditions

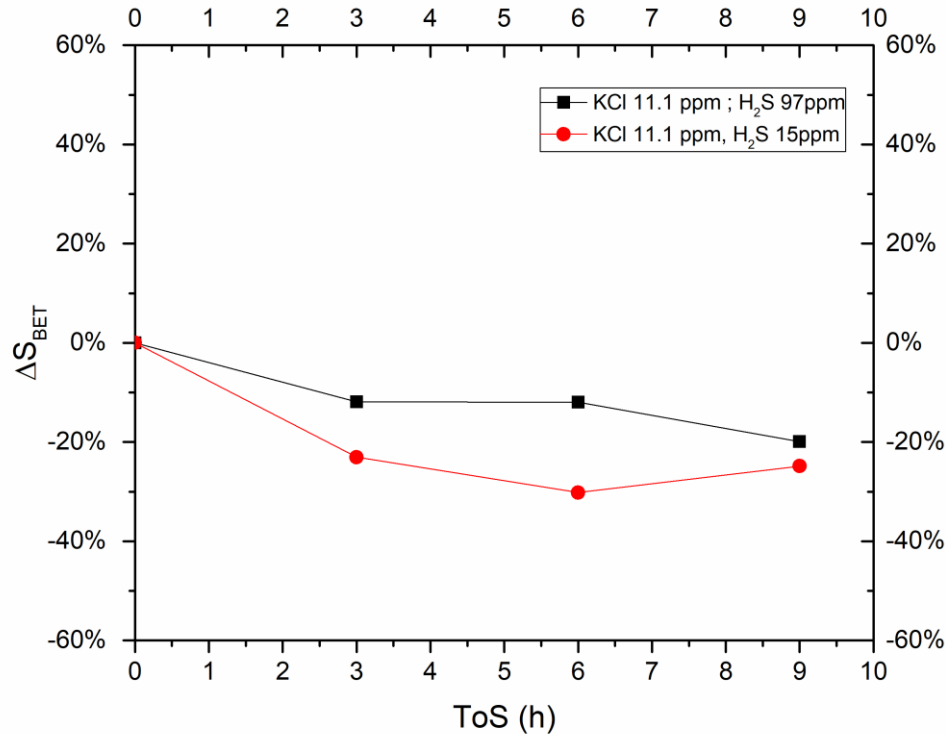
Biomass Feed rate(g/h)	210-225
Gasification Temperature (°C)	825-850
λ	0.23-0.25
Catalytic Bed Temperature [°C]	800-850

Results- 850°C Reduced Ni-catalyst

Gradual sulfur coverage affects the rate limiting step of conversion of different molecules



Results- 850°C Reduced Ni-catalyst

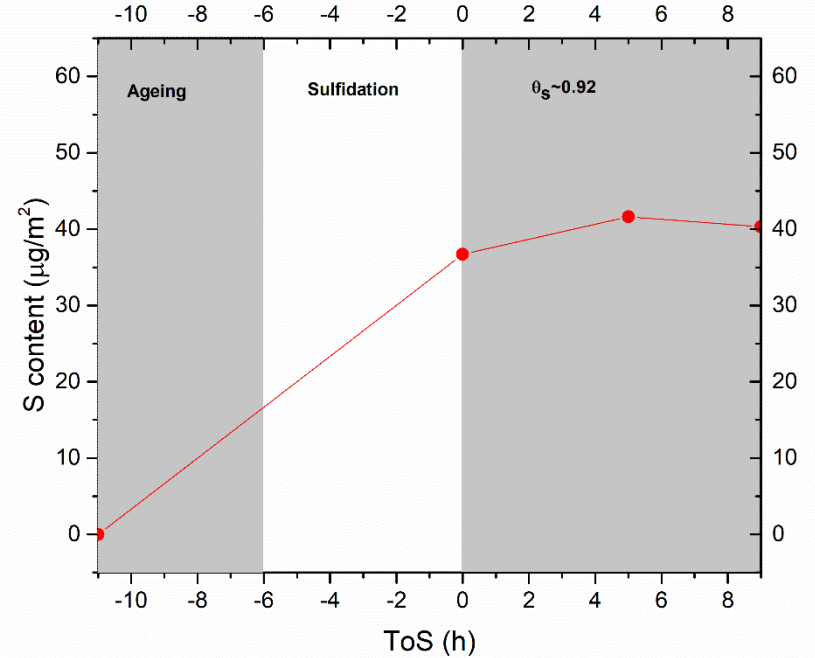
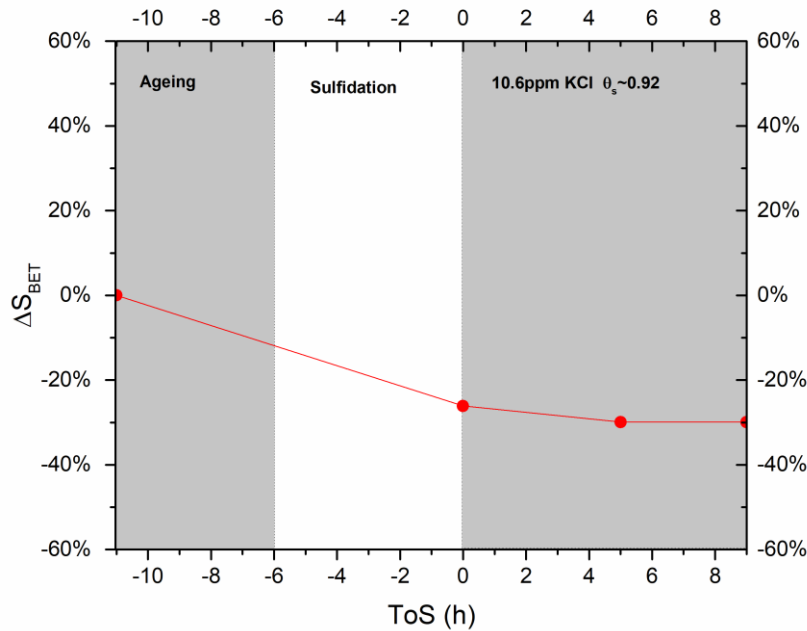


Sintering affects
observed conversion
changes

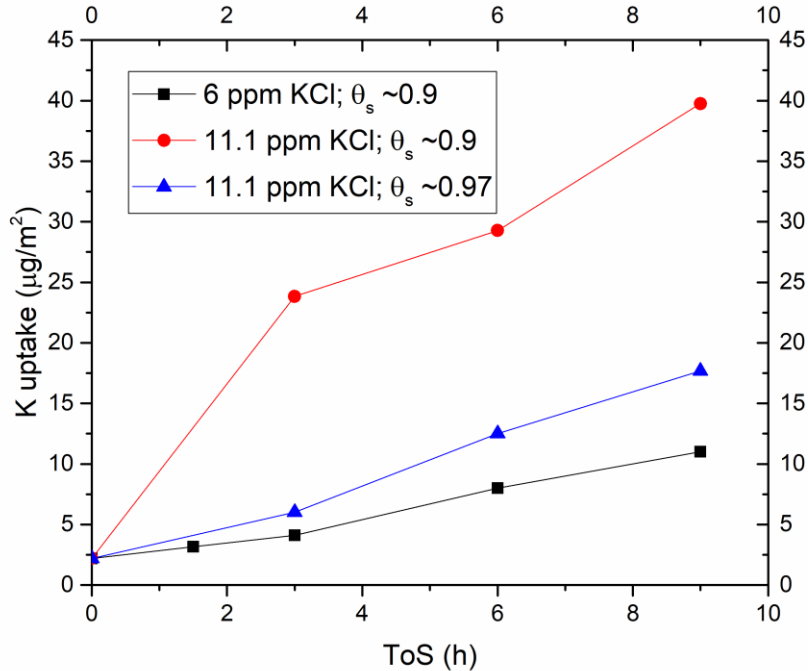
Methodology development- Stabilized & S passivated catalyst

- Ageing @900+ °C in steam ($\text{H}_2\text{O}:\text{H}_2 = 10$) for 5hrs - minimization of activity changes due to sintering
- Pre-sulfidation (passivation) at certain θ_s at relevant operating conditions- elimination of activity changes due to transient phenomena related to sulfur adsorption and equilibration on the surface.

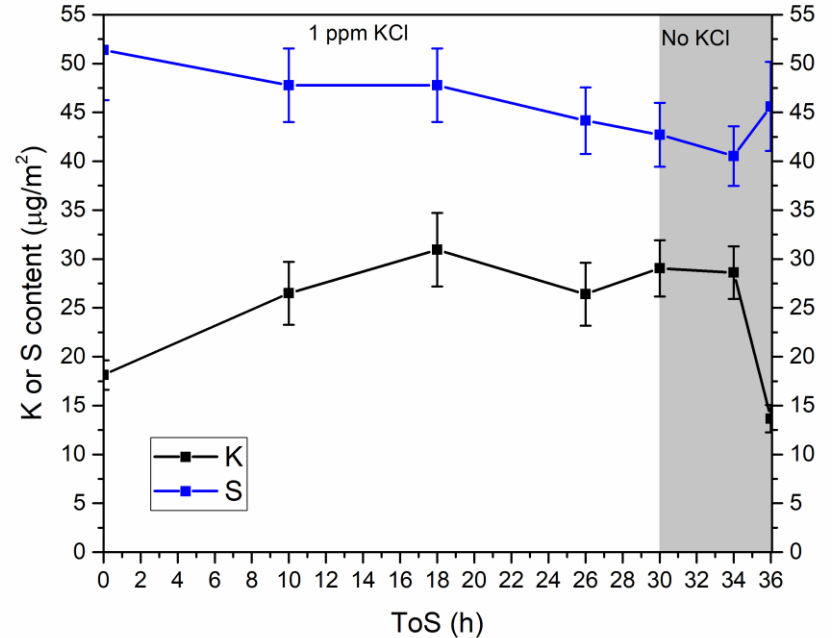
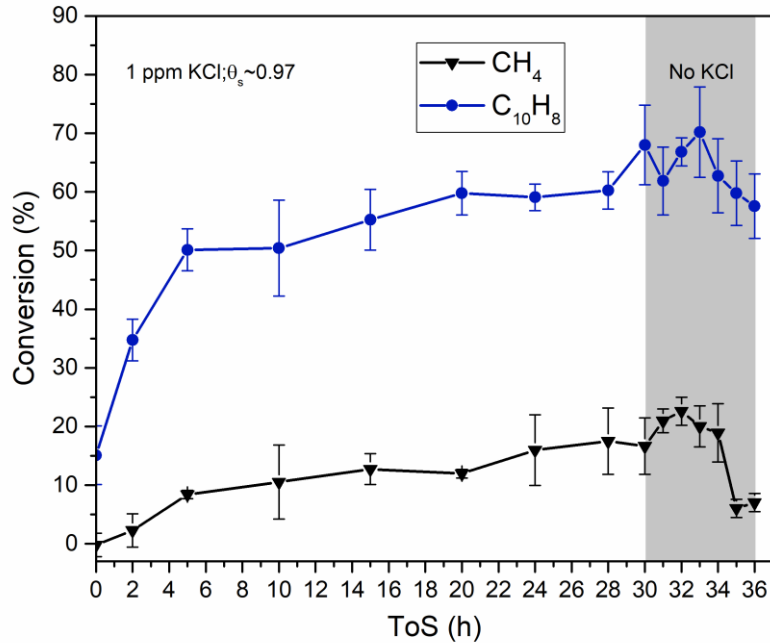
Results- 850°C Aged & passivated Ni-catalyst

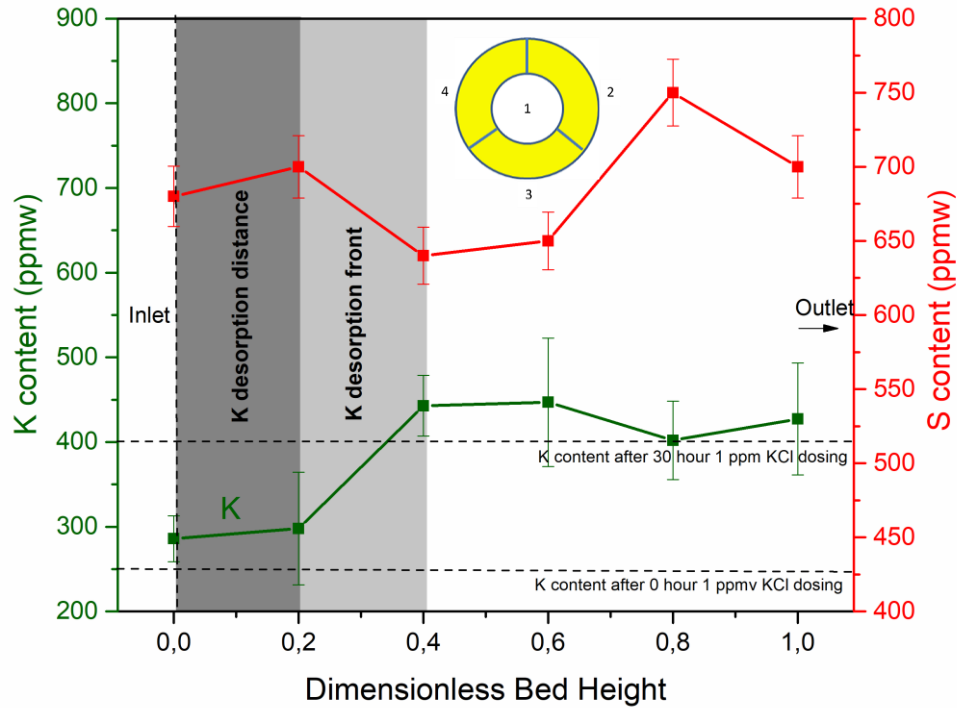


Results- 850°C Aged & passivated Ni-catalyst



Tar reforming at 800°C $\theta_s \sim 0.97$

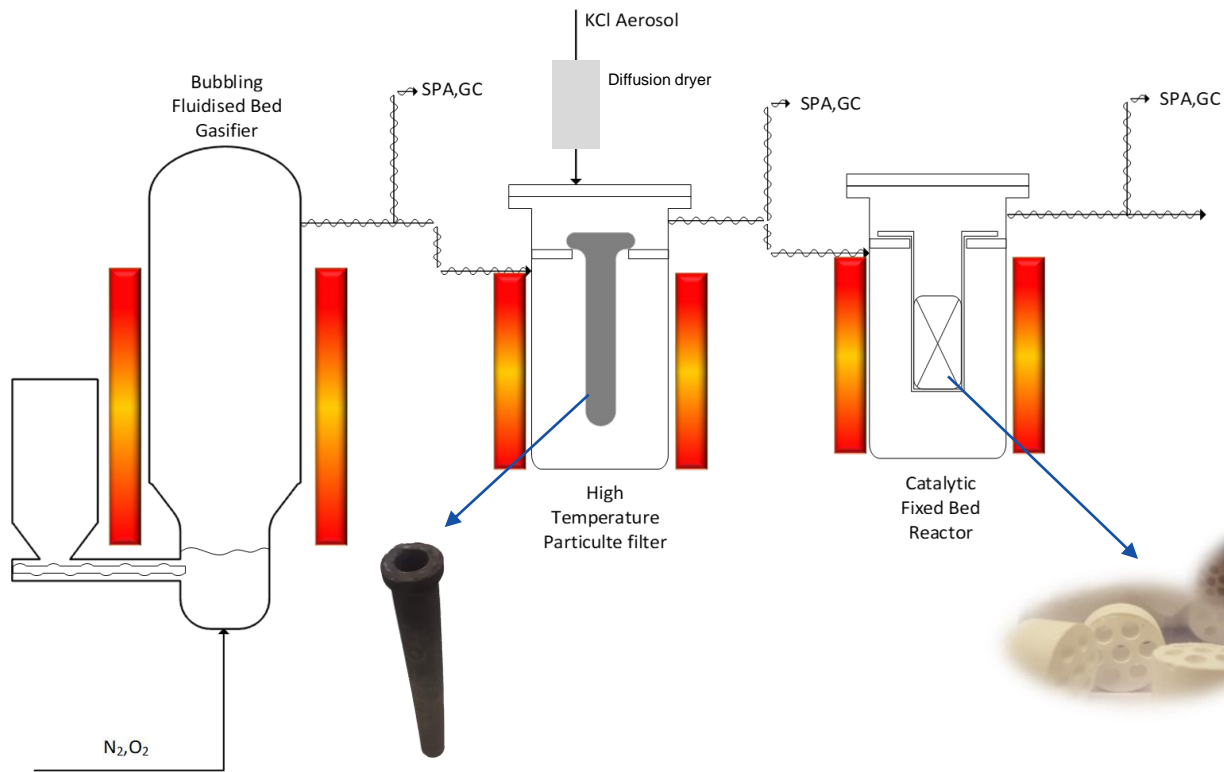


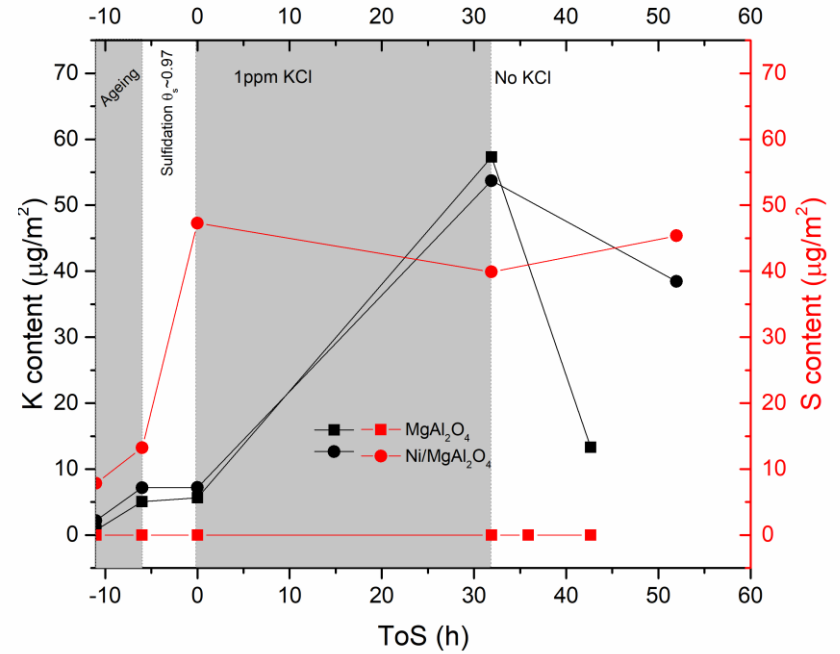
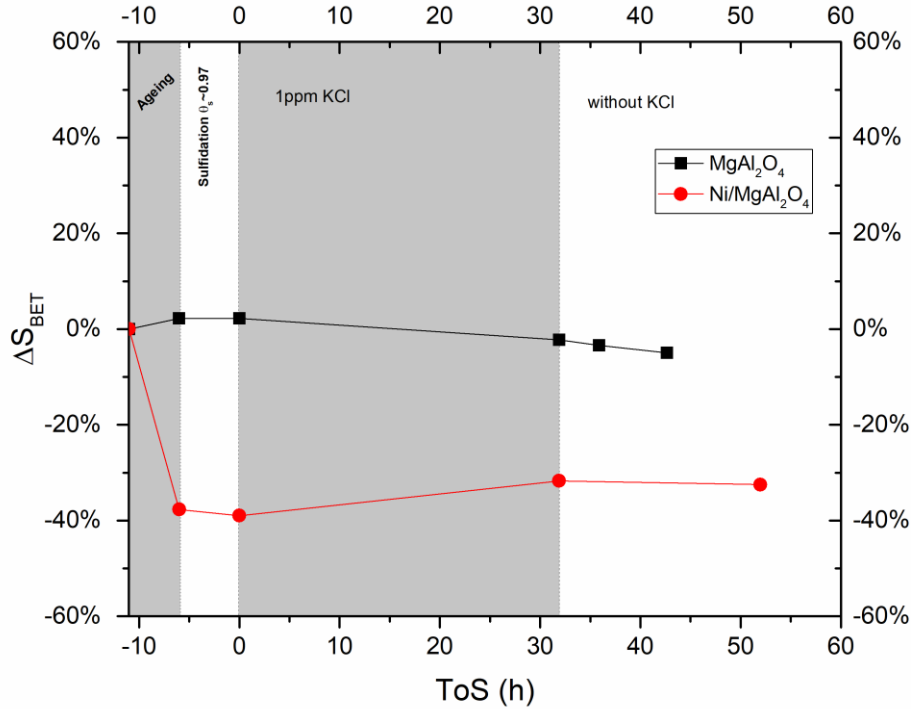


Question

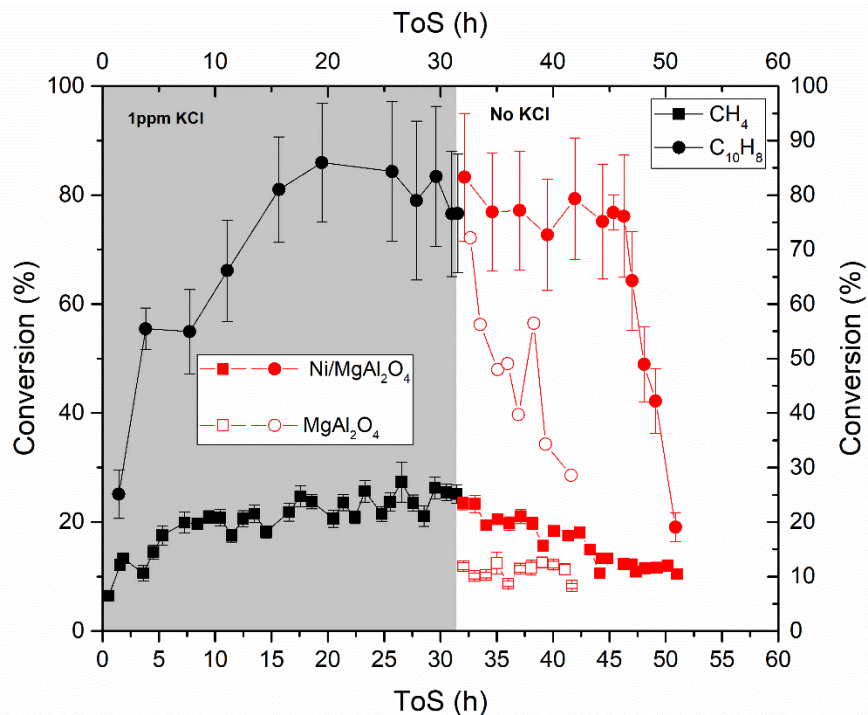
- What is the role of the support ?

Experimental Setup



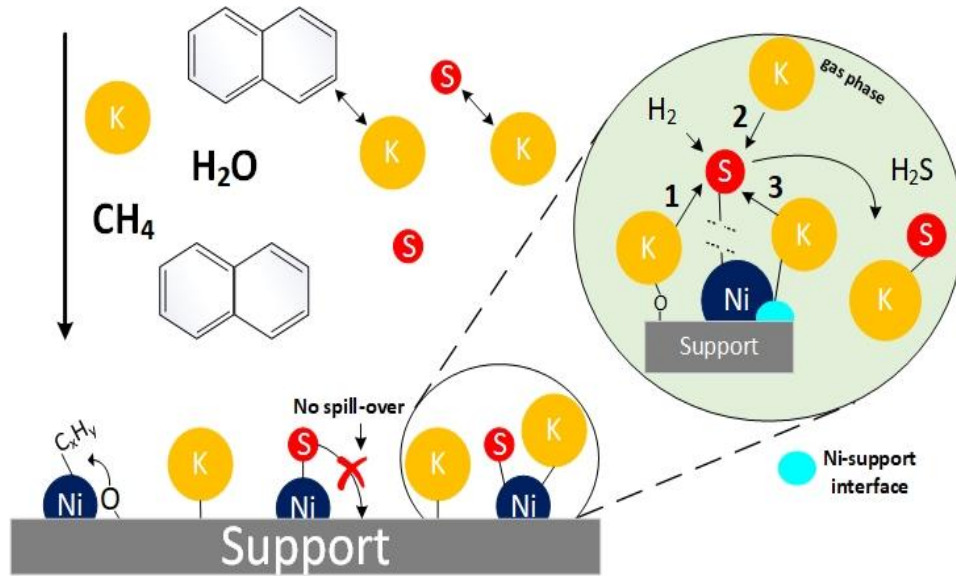


Effects of support-sorbed K

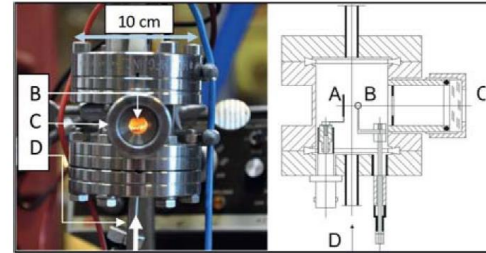
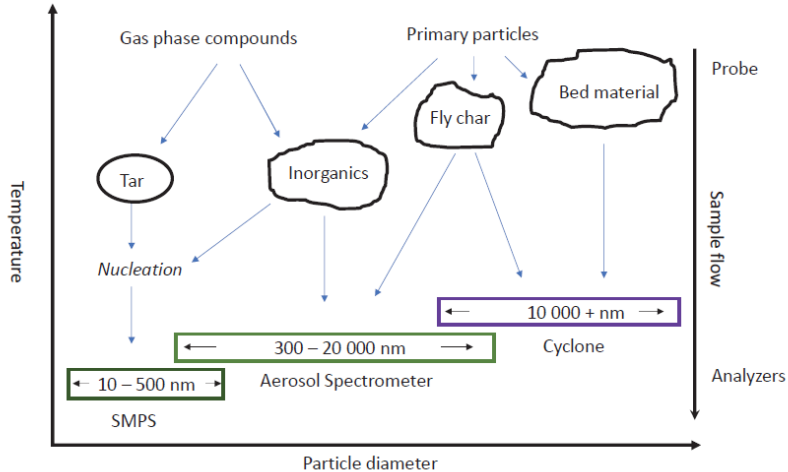


C on catalyst at the end of saturation: 0,56 mg/g_{cat}

C on catalyst after decay: 0,67 mg/g_{cat}



On Line K measurement using Surface Ionization Detector



Correlate tar particle size distribution to its content

H₂S trace detection (0-1ppm) using colorimetric analyzer

Conclusions

- Reproducible dosing of gas-phase K and elimination of transient effects catalyst sintering and sulfur passivation allow for investigation of K/S effects using real gasification gas
- K uptakes of $<100 \mu\text{g}/\text{m}^2$ catalyst under reforming conditions
- As θ_K increase on the catalyst the θ_S on Ni sites decreases improving conversions (Plausible K-induced Ni-S bond weakening)
- Support-sorbed K has some effect on conversion of bigger tar molecules and almost no effect on CH_4
- Tailoring K and H_2S concentrations in the gas allows for virtually carbon-free operation

Acknowledgements

