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Z L W K) 89X R U H V F H Q F H

YorkNeubauer

TCKON Engineering Services Dr.-Ing. York Neubauer

Berlin, Germany

IEA Bioenergy, Task 33: Thermal Gasification of Biomass and Waste
WORKSHOP "GAS CLEANING, EXPERIENCES, NEW DEVELOPMENTS, ANALYTICS AND DIAGNOSTICS"
June 6th 2019, KIT, Karlsruhe, Germany

outline

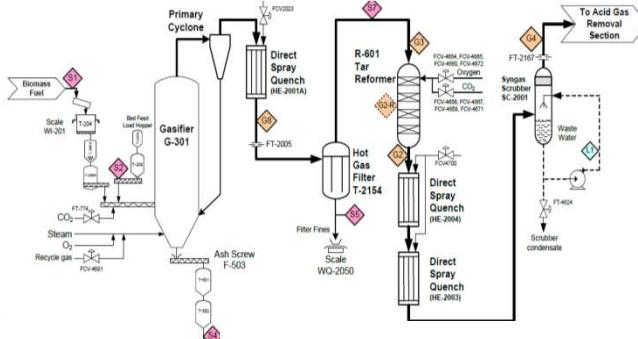
- (on-line) ,tar'-monitoring and ,tar'-analysis – what is needed?
- The dream of on-line tools for perfect results at neglectible operating and maintenance efforts
- Approaching the needs and wishes with UV-fluorescence
- outlook

perspectives

science

Researcher

- reaction engineering (gasifier)
- material science
- catalyst development and testing
- chemical analysis
- ...



instrument manufacturer

- Standardized analytical procedures (emissions)

accredited analytical lab

- Standardized analytical procedures (emissions)

plant operator

- stable and trouble free operation
- maintenance
- working safety for operating & maintenance personnel

plant owner

- ROI

industry

plant manufacturer

- commissioning
- optimization
- product development
- general contractor

unit manufacturer

- gasifier
- filter
- scrubber
- adsorber

consumables supplier

- catalysts
- bed materials
- ash disposal

overall gas composition as quality measure

permanent gases

desired contaminants hydrocarbon gases
C1-C5

condensables

partides, aerosols, metals

O	H ₂ S
H	COS
O ₂	NH ₃
N ₂	HCN
	HCl
	...

methane
ethane
ethene
propane
propene
...

steam/water

benzene
1-ring aromatics
polycyclic aromatic
compounds (PAC)



,tar' – when condensed and
in each other dissolved

What gases are the troublemakers?

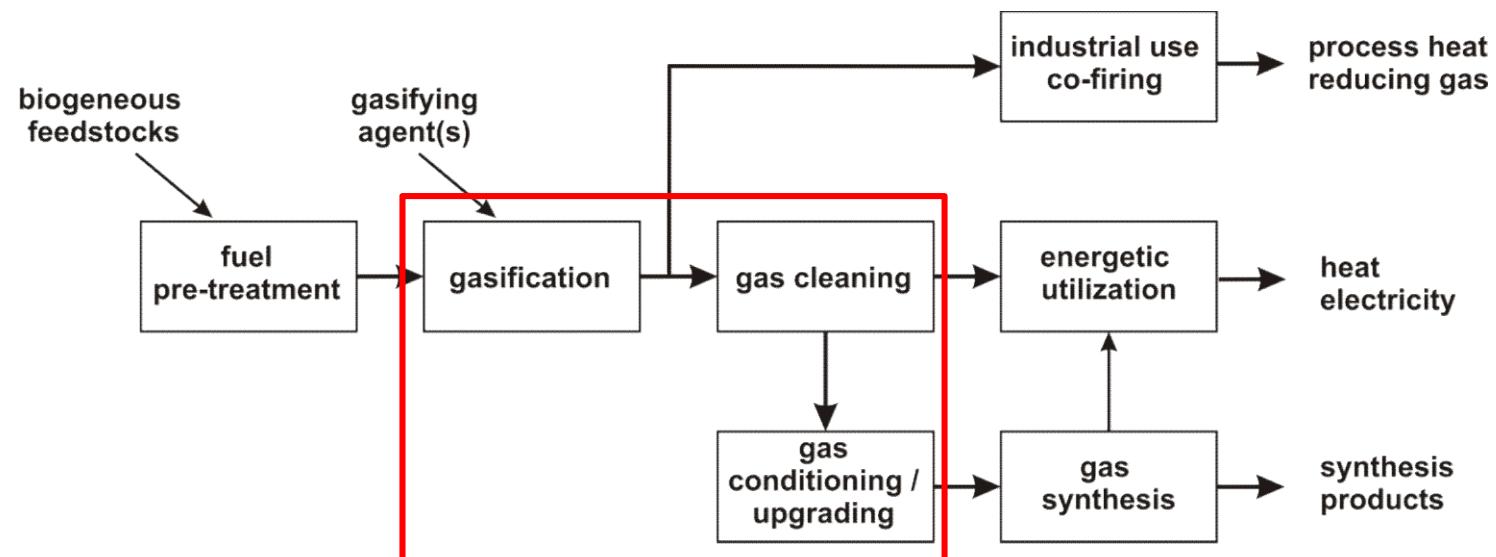
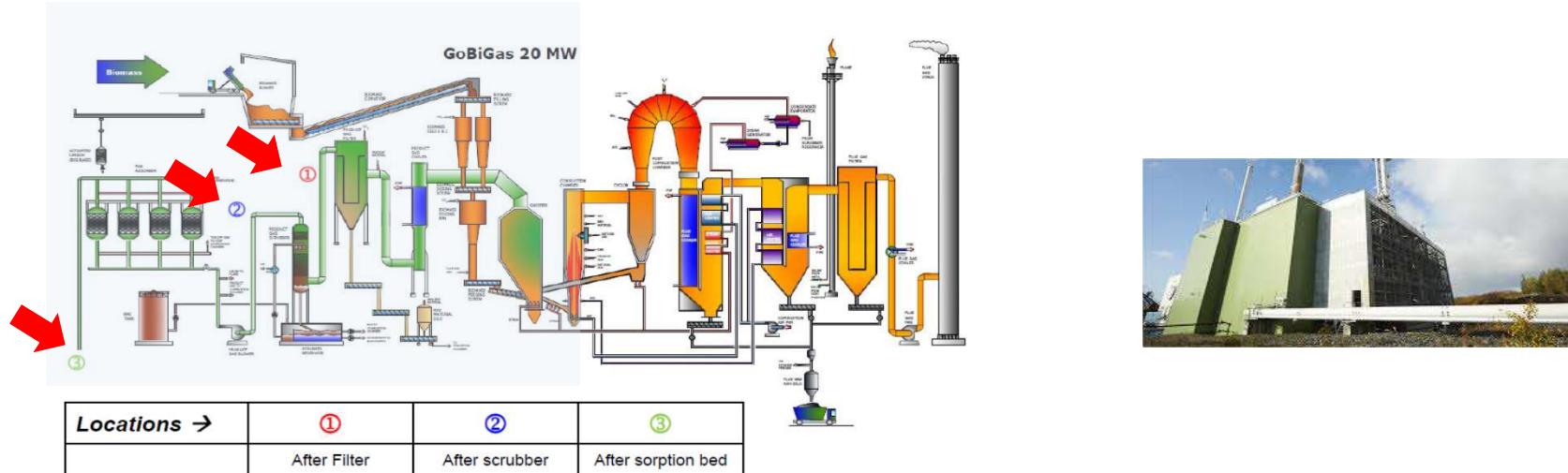
Needs and wishes

Industryneeds

Scienewarts

- Fast and detailed results
 - Preferred L Q O L Q H P H D V X U H P H Q W V revealing reactor insights on time
 - continuous monitoring
 - validated, spatial resolved data for modelling and simulation
- A reliable, robust, easy to operate and to maintain measurement or monitoring instrument.
- A ‘translation’ of the physical tar species properties and amounts into a useful value, characterizing the current status of the respective plant section.
- A device that can be taken care of by a technician - rather than a scientist !
- A tool that handles tar loaded gas which stays longer in operation than the monitored part of the plant itself!
- Such a device needs to be applicable in the industrial environment (temperature, vibrations, dust, ex proof zones). To assure a long operating and life time, the direct contact of the measuring unit with the tar species especially in ‘raw’ gases needs to be avoided.
- 2 S W L F D O W H U C K & L T V - X fluorescence spectroscopy offer a number of benefits for this purpose.

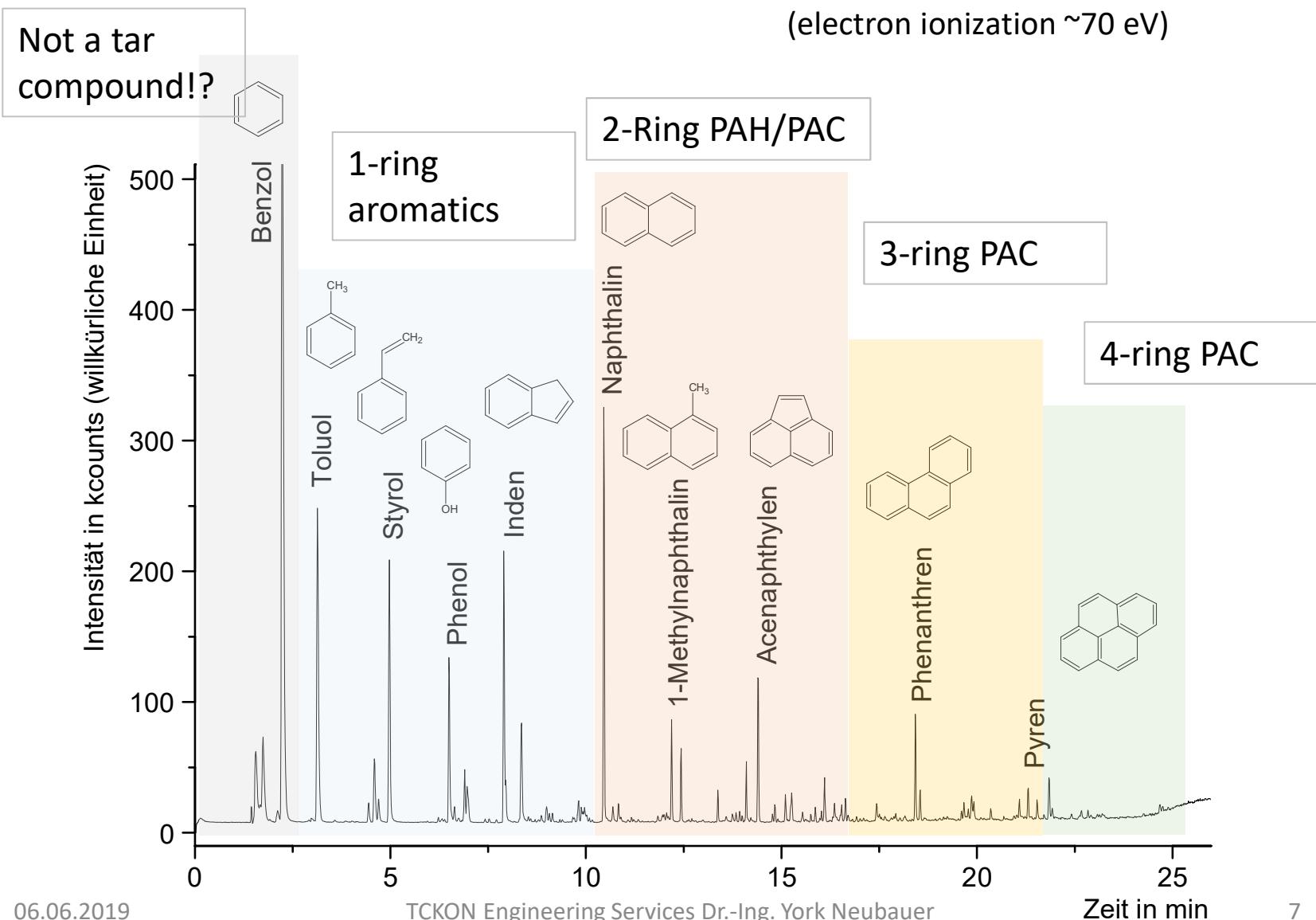
J D V I D P E Depending on process, fuel and location within the plant



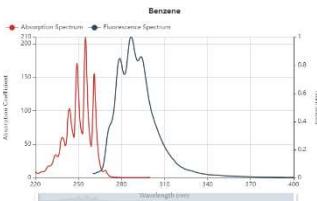
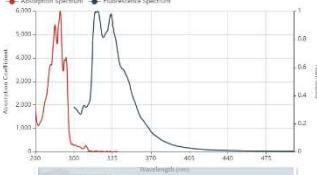
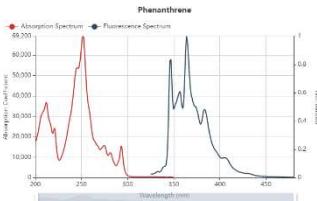
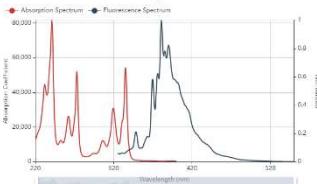
tar composition

Onlinetaranalysis with GCMS

gasification of woody biomass with air
(electron ionization ~70 eV)



Properties of selected tar ,representatives'

1 D P H	& \$ 6	1 R 0 R O D I F T Q D P O R L C 2 Q W L F D O 3 U R S H U W L H V \$ E V R U S W L R Q D Q G (P	Z H L J K W S R L Q f & f &		
Benzene 	71-43-2	78,11	6 80	Absorption coefficient: 210 at 254.75 nm Fluorescence quantum yield: 0.053	
Naphthalene 	91-20-3	128,17	80 218	Absorption coefficient: 6000 at 275 nm Fluorescence quantum yield: 0.23	
Phenanthrene 	85-01-8	178,23	99 336	Absorption coefficient: 69200 at 252 nm Fluorescence quantum yield: 0.125	
Pyrene 	129-00-0	202,25	156 340	Absorption coefficient: 81700 at 241 nm Fluorescence quantum yield: 0.32	

Taniguchi, M.; Du, H.; Lindsey, J. S., PhotochemCAD 3: Diverse Modules for Photophysical Calculations with Access to Multiple Spectral Databases, *Photochem. Photobiol.* 2018, 94, 277–289.

Taniguchi, M.; Lindsey, J. S., Database of Absorption and Fluorescence Spectra of >300 Common Compounds for Use in PhotochemCAD," *Photochem. Photobiol.* 2018, 94, 290–327.

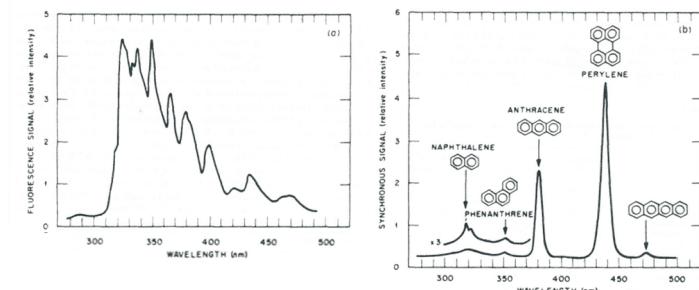
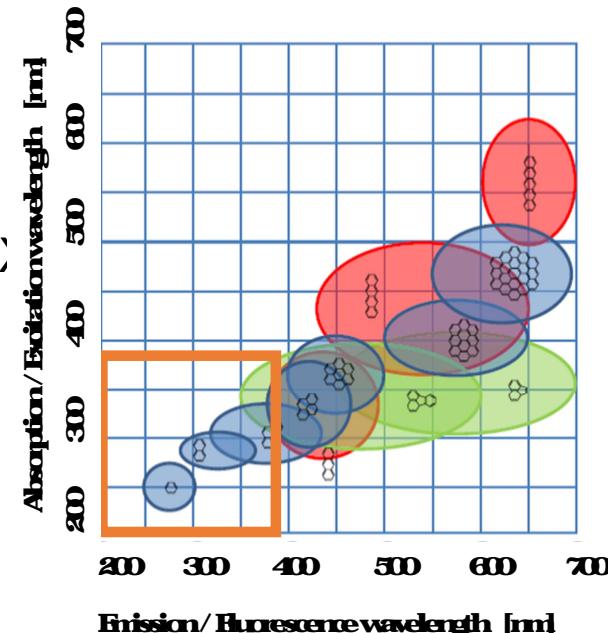
Attempting to access aromatics contents in the process gases continuously on-line

Goal of fundamental research work:

2 E W D L Q L Q J L Q I R U P D W L R Q
F R P S R X Q G V R U R Q F R P S R)
Z L W K 8 9 I O X R U H V F H Q F H V

examination of:

- Excitation-Emission matrices - EEM
 - Synchronous fluorescence
 - Fluorescence lifetime
-
- Comparisons between liquid solvent phase with tar species and high temperature gas phase

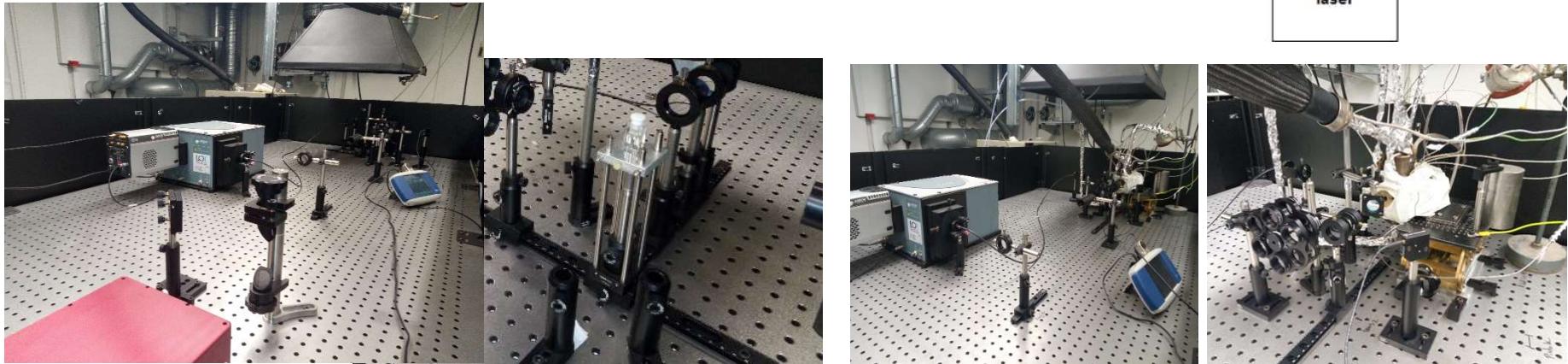
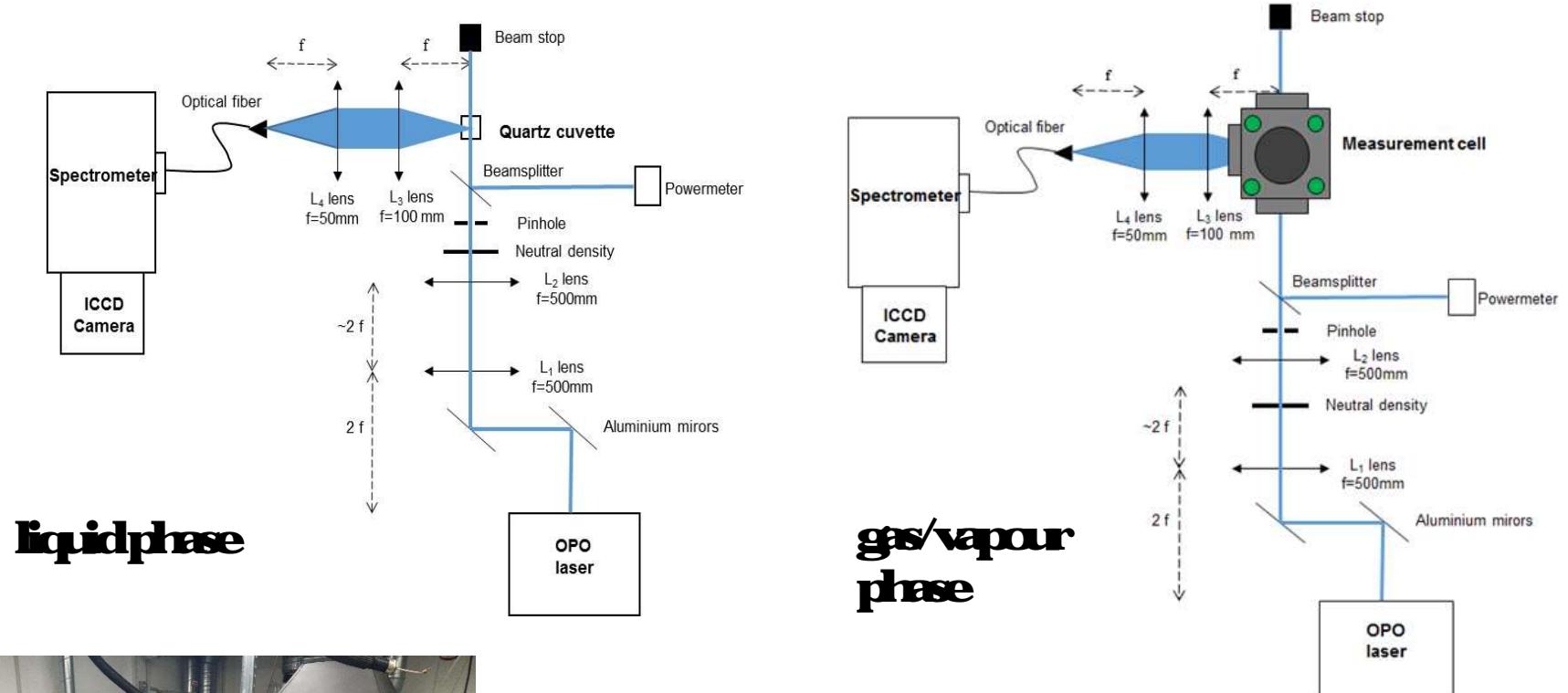


synchronous luminescence spectroscopy [Vo Dinh 1978]

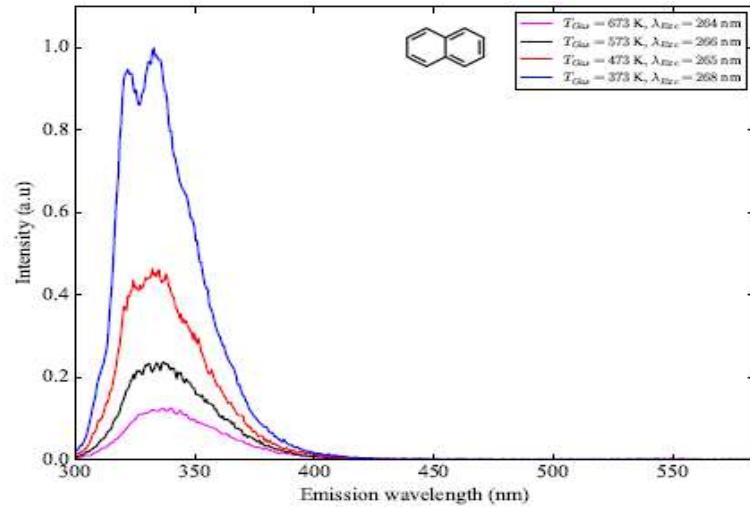
Picture on the upper right taken and adapted from:

S. Bejaoui, X. Mercier, P. Desgroux, E. Therssen: Laser induced fluorescence spectroscopy of aromatic species produced in atmospheric sooty flames using UV and visible excitation wavelengths. Combustion and Flame 161 (2014) 2479–2491

Principal setups for fundamental examinations

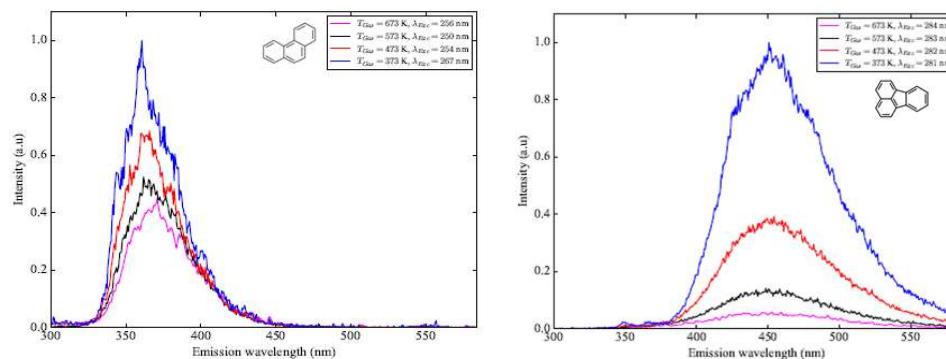


Temperature dependence of the fluorescence signal



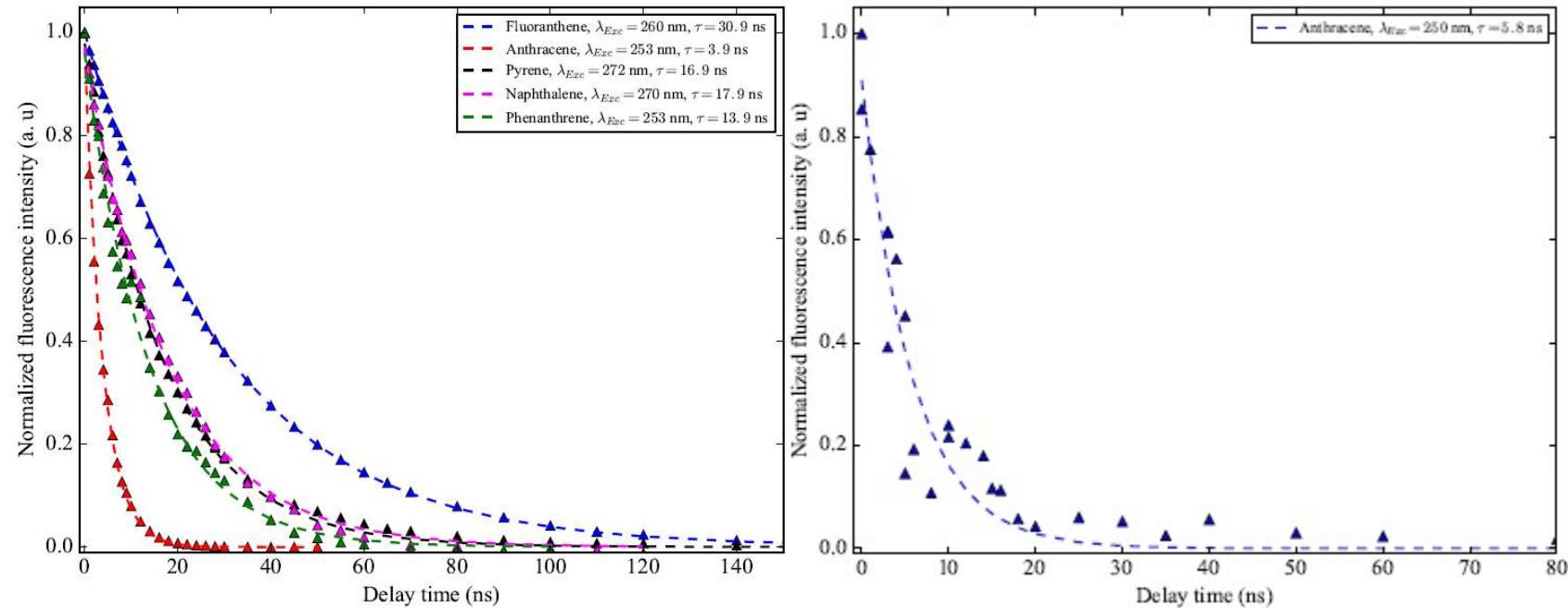
Fluorescence intensities drop with increasing temperature

Consequence high temperatures of 350 – 400°C for surely avoiding tar deposits are in contrast to spectral quality



Diluting could be a solution but represents a challenge on its own – see work of gti or VTT!

Separation by fluorescence lifetime



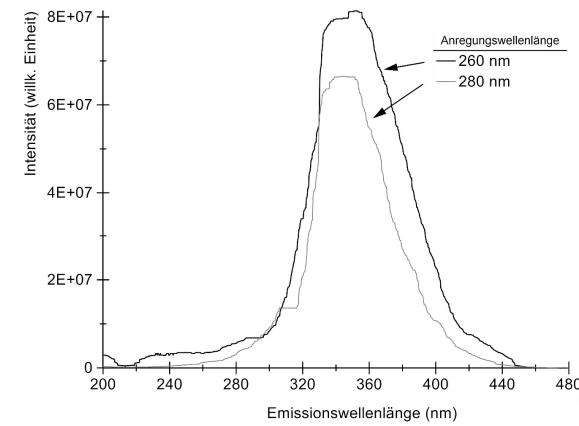
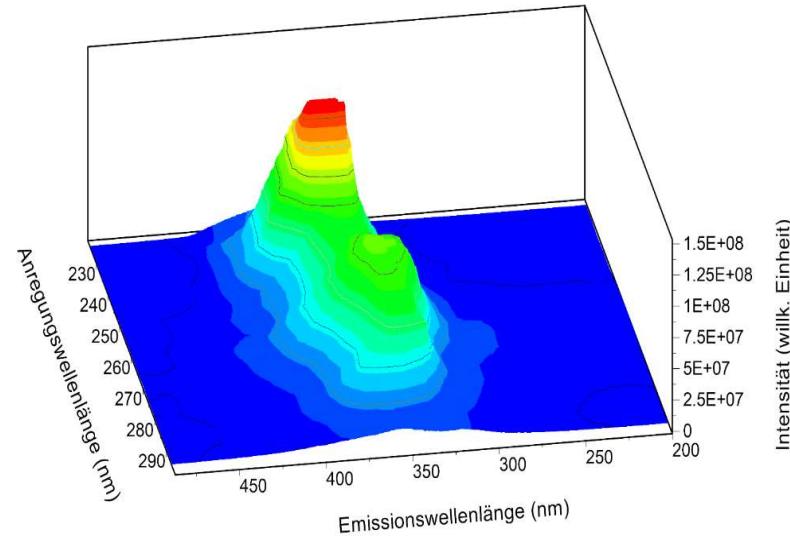
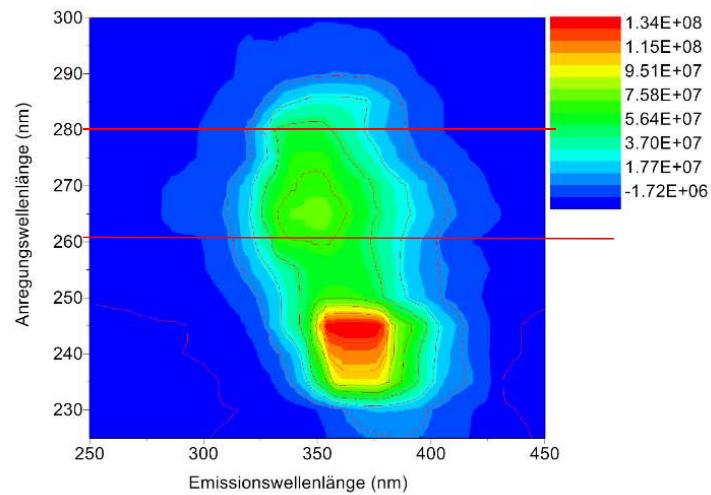
) O X R U H V F H Q F H G H F D \ F X U Y H V R I 3 \$ + Ø X L R Q U R \ \ F F O H R Q K F H [D Q H F D \ F X U Y H R I D Q W K U D I
R E W D L Q H G D W . D I W H U H [F L W D W L R Q

PAH	Excitation wavelength (nm)	Lifetime in cyclohexane (ns)
Naphthalene	270	17.9
Phenanthrene	253	13.9
Anthracene	253	3.9
Pyrene	272	16.9
Fluoranthene	260	30.9

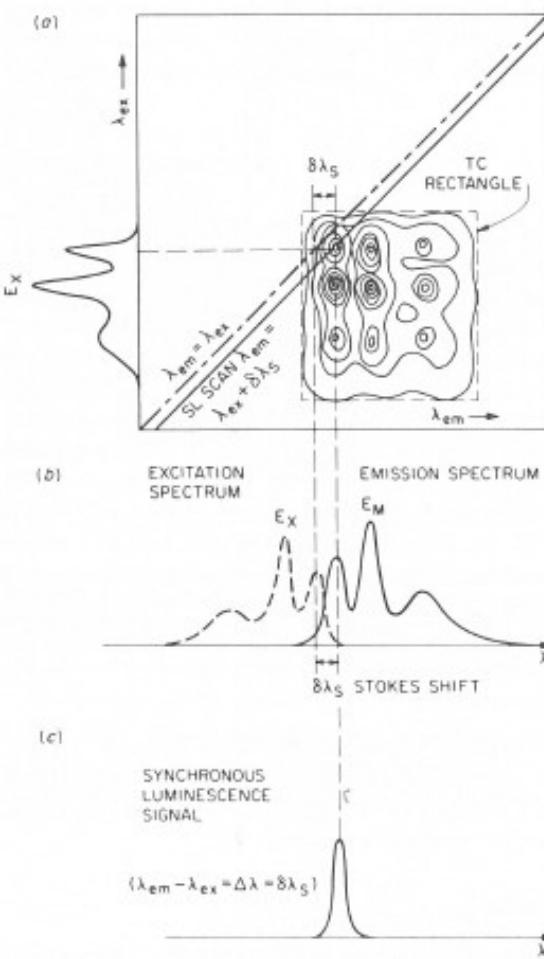
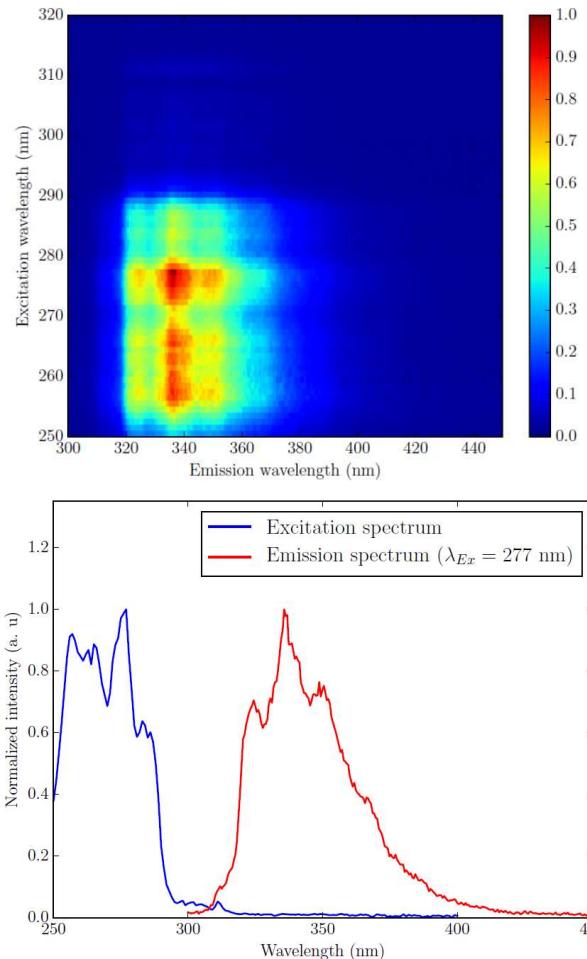
([S H U L P H Q W D O I O X R U H V F H Q F H O L I H W L P H V R I 3 \$ + V L Q F \ F O R K H [D Q H

Excitation-Emission matrices – EEM's

- EEM of phenanthrene
- Are EEM's a choice for mapping and fingerprinting?

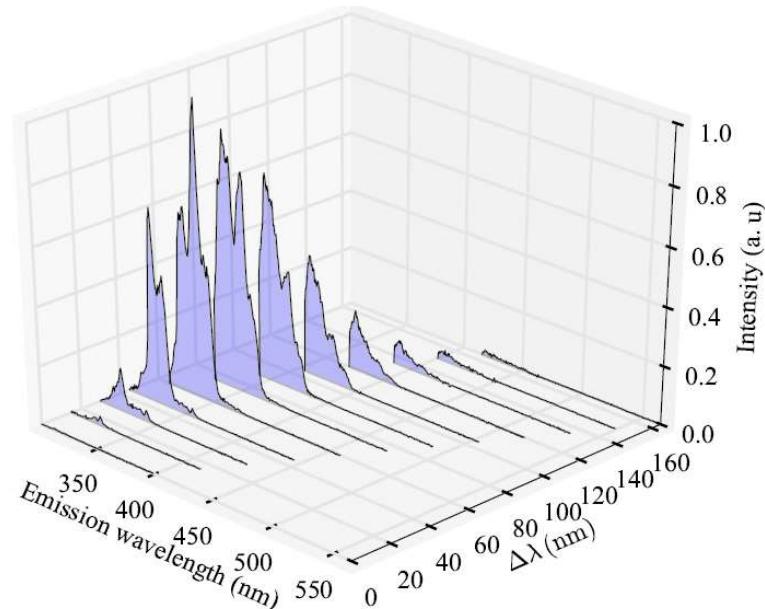
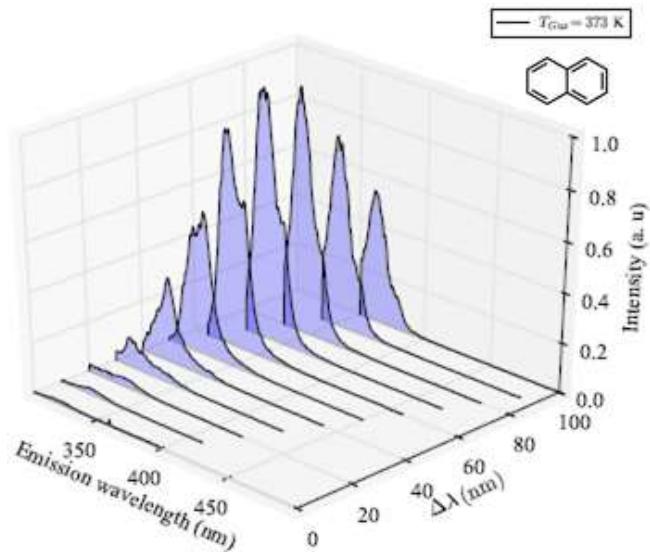


Excitation- Emission matrizes obtained with wavelength tunable OPO-laser

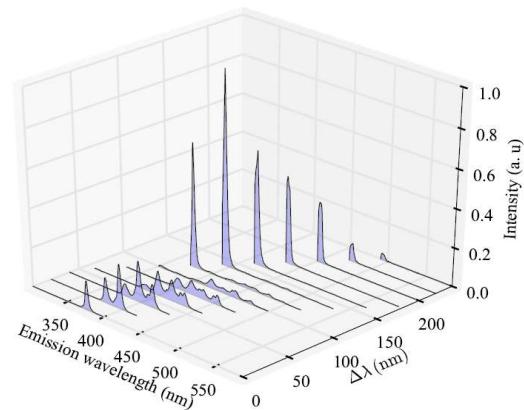
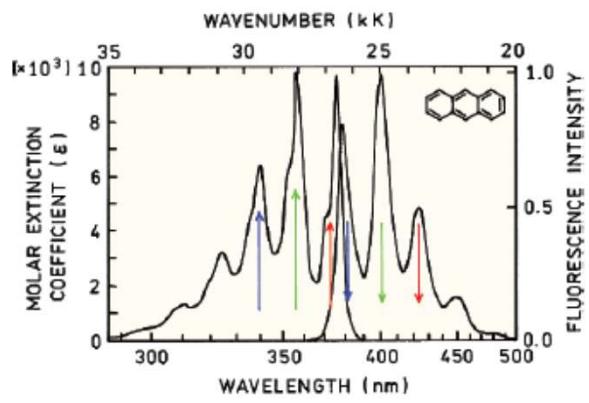


[FLWDWLRLQ (PLVVLRLQ PDWUL[XESKHLPSWLUFWR IDWKGH UHODWLRLQVKLSV EHWZHHQ D ((
H[FLWDWLRLQ DQG HPLVVLRLQ VSHFWUDPBLRZLURQSVSHFWUD DQG F V\QFKURQRXV IOXR
SKHQDQWKUHQH LQ F\ORKH[DQHTRUD FRQFHQWUDWLRLQ
* E Wehrly, 'Modern fluorescence spectroscopy 4', Plenum Press, New York, 1981
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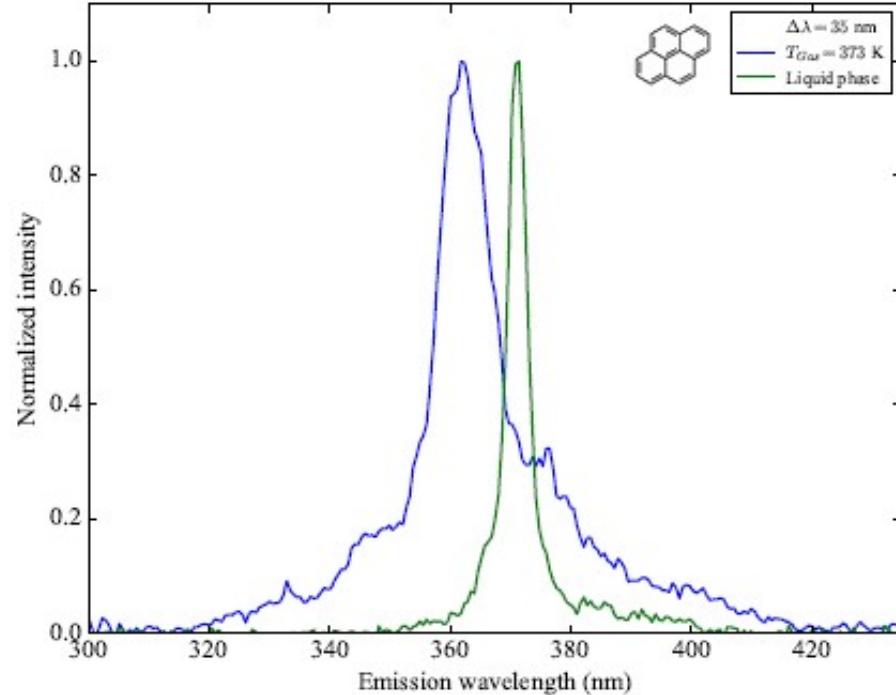
Synchronous fluorescence



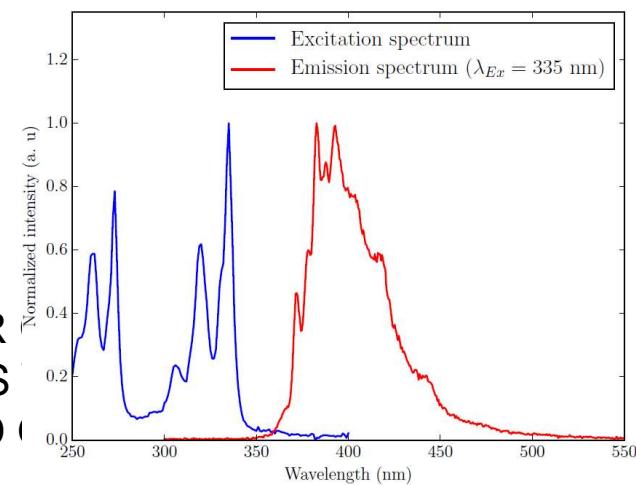
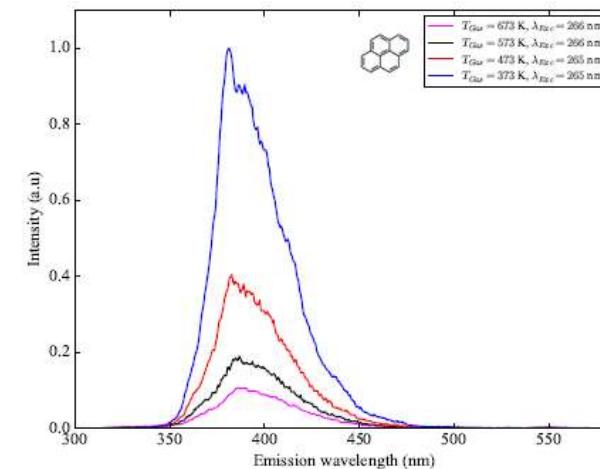
& R P S D U L V R Q R I W K H V \ Q F K U R Q R X V I O X R U H V F H Q F H V S H F
O H I W D Q G O L T X L G & K D V H D W Y D U \ L Q J



Synchronous fluorescence



& R P S D U L V R Q R I W K H V \ Q F K U R
 I O X R U H V F H Q F H V S H F W U D R I S
 J D V S K D V H D W D V S W F L I L F Y D I



J D V S K D V H D Q G O L T X L G S K D V H H P L V V L R

Towards industrial application for tar monitoring

5 R E X V W D Q G U H O L D E O H Avoiding contact of windows with sample gas flow by windows flushing with inert gas in specially constructed measurement cell

\$ V P X F K D Q G D V O R Q J D V Heated cell and main gas paths within SR V V L E O H X Q P D Q Q H G R S a h l o v e n M i k e R c o n s t r u c t i o n

/ L W W O H P D L Q W H Q D Q F H No moving parts; use of ejector pump;
U H T X L U H P H Q W V E \ W H F K Q r i e f e r s d o n With combustion of the off
Q R W D Q D O \ W L F H [S H U W V gas (no H₂, no CO, no tar left over – safety issue!)

/ H V V H [S H Q V L Y H D Q G Z L W A Applying UV diodes instead of laser
U R E X V W F R P S R Q H Q W V for excitation

(O L P L Q D W L Q J I X U W K H U L D e h o r n e d Q b o d e d Q) b o p t i c s
I D F W R U V compartment for LED and for spectrometer; possibly addition of dilution unit for measuring at lower gas temperature

Field deployable set-ups



@CHALMERS University of Technology,
Gothenburg, SE



@ GoBiGas plant, Gothenburg, SE



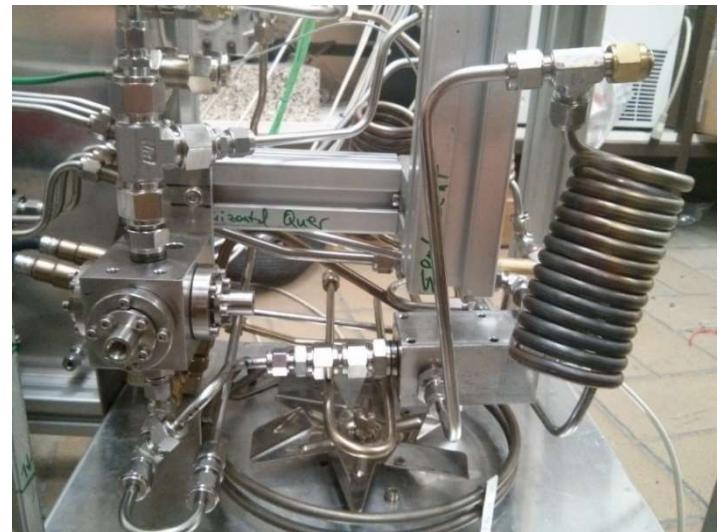
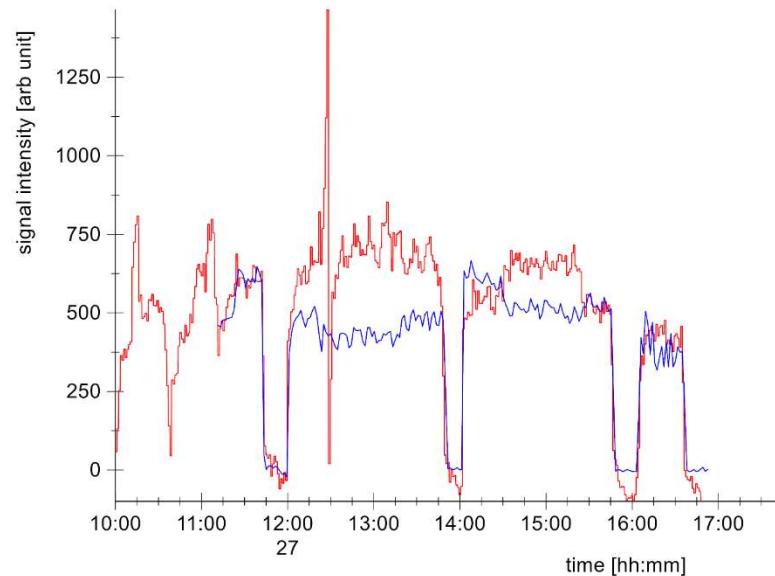
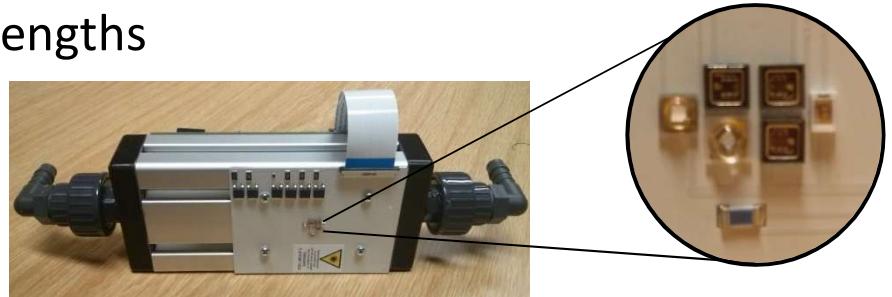
@PSI, Villigen, CH



Comparison with
FID device from
Ratfisch
Analysensysteme
and University
Stuttgart

Field deployable set-ups

UV LED diode array with different wavelengths
265, 280, 300, 340, 380, 455 nm



Comparison with FID device (blue curve) from
Ratfisch Analysensysteme and University Stuttgart

Acknowledgement

Special thanks to my former team members at TU Berlin Dr. Thomas Mouton, Halgurd Taher, Sandra Walther, Dr. Shaimaa Mahdi and Dr. Eva M. Brüning for their work and their patience for working out several spectroscopic options of individual tar species detection in the hot process gas.

The fundamental work was part of my M X Q ICP
U H V H D U F K J U R X S ³¹ : (*2017 & 2018)



Federal Ministry
of Education
and Research



Federal Ministry of Education and Research via PtJ:
FKZ: 03SF0442

In the % L R 3 U R * 5 H 6 V 2014-2017 Demonstration units applying the main design principles were tested and further adapted in field measurements at CHALMERS University and the GoBiGas plant.

Federal Ministry of Food and Agriculture via FNR:
FKZ: 22401814.



Federal Ministry
of Food
and Agriculture



Fachagentur Nachwachsende Rohstoffe e.V.



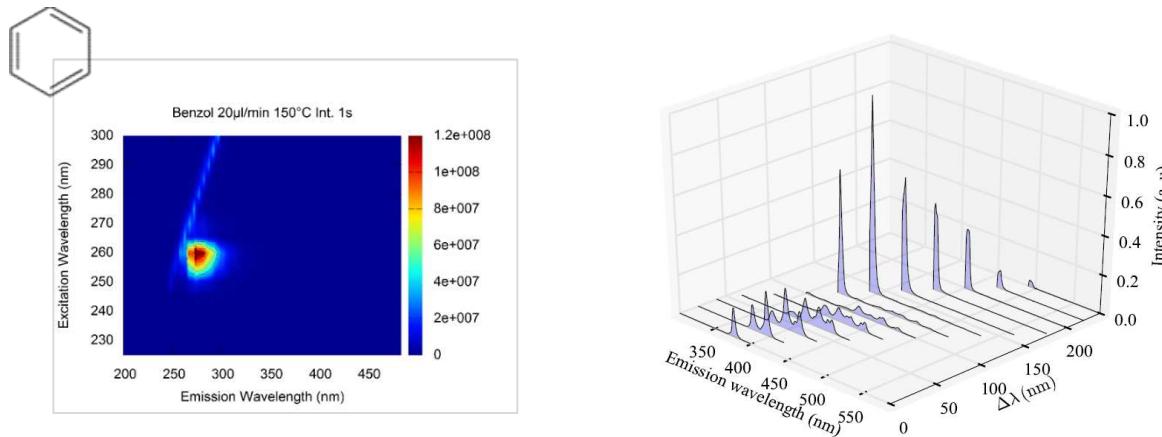
The journey continues...

Our solution:

- An optical instrument based on UV-fluorescence that measures by avoiding the direct contact with the gas.
- Special heated and tempered sections for process gas, utility gases and for sensitive optical and electronic parts of the device.
- Generating under-pressure for gas transport from the sampling point and combustion of the effluents by non moving solid parts, strongly reducing maintenance requirements!
- Industry-standard instrument control for easy adaptability in process environments including sensing device and auxiliary sampling system.
- Contact and latest developments: condensables.com



Thank you for your interest
and for your attention!



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