



I4. Synthesis of Bio-LPG from Biomass-derived Syngas

Tomoko OGI, Masakazu NAKANISHI,
National Institute of Advanced Industrial Science and Technology
Kaoru FUJIMOTO
University of Kitakyushu

IEA Bioenergy Conference
Vienna Austria, November 13 , 2012



Introduction: About JAPAN –Energy Status, now- Primary energy sources

	2008	2009	2010	2011
hydraulic	653PJ	648PJ	699PJ	640PJ
geothermal	23PJ	24PJ	22PJ	22PJ
nuclear	2 218PJ	2 420PJ	2 576PJ	1 436PJ
coal	5 455PJ	4 409PJ	5 024PJ	4 765PJ
petroleum	11 088PJ	9 647PJ	9 964PJ	9 814PJ
natural gas	3 944PJ	3 678PJ	3 970PJ	4 431PJ
TOTAL	23 282PJ	20 827PJ	22 255PJ	21 107PJ

fossil-LPG is about 6% of imported petroleum and natural gas.

$$\left. \begin{array}{l} \text{petroleum} \\ \text{natural gas} \end{array} \right\} 14-15\text{EJ} \quad \frac{750-800\text{PJ}}{14-15\text{EJ}} \approx 6\%$$

Energy applications

	2008	2009	2010	2011
LPG	16 245kt	15 081kt	14 644kt	15 008kt
municipal gas	35 727Mm ³	32 954Mm ³	35 206Mm ³	35 109Mm ³
petroleum oil	207 670ML	193 996 ML	197 209ML	193 056ML
gasoline	57 247ML	57 447ML	58 368ML	56 864ML
diesel oil	32 246ML	32 247ML	33 057ML	44 646ML
electric power	913 138GWh	846 725GWh	901 522GWh	859 663GWh

750-800PJ

About JAPAN –Strategy- why LPG? Why bio-LPG?



New National Energy Strategy (2006)

- Petroleum oil dependency of transport energy will be brought down from 100% to 80% till 2030.
- 500,000kl liquid bio-fuel will be introduced.

Next Generation of Vehicles and Fuel Initiative (2010)

- The second generation liquid bio-fuel was nominated.

Act of Sophisticated Methods of Energy Supply Structures (2010)

- Use of non-fossil energy sources and the effective use of fossil energy resources are promoted.
- LPG suppliers are mandated to introduce bio-gas.

Basic Energy Plan for Japan (2010)

- LPG is classified to be distributed source of energy.

-----March 11, 2011---the East Japan Great Earthquake

Strategy-...Why LPG? ...then East Japan Great Earthquake



*An earthquake of unprecedented force
A tsunami of unspeakable destructiveness
....then A radioactive awakening*

Japan convulses, grieves, moves on

*At that time, from that day, LPG has been
supplied and used as local energy for
heat, cooking....*

Basic Energy Plan for Japan (2010)

- LPG is classified to be distributed source of energy.
- LPG is supplied to more than 90% area of Japan.
- ***LPG is very useful especially in disaster.***



About JAPAN –Forest- Why bio-LPG?

☞ New National Energy Strategy (2006)

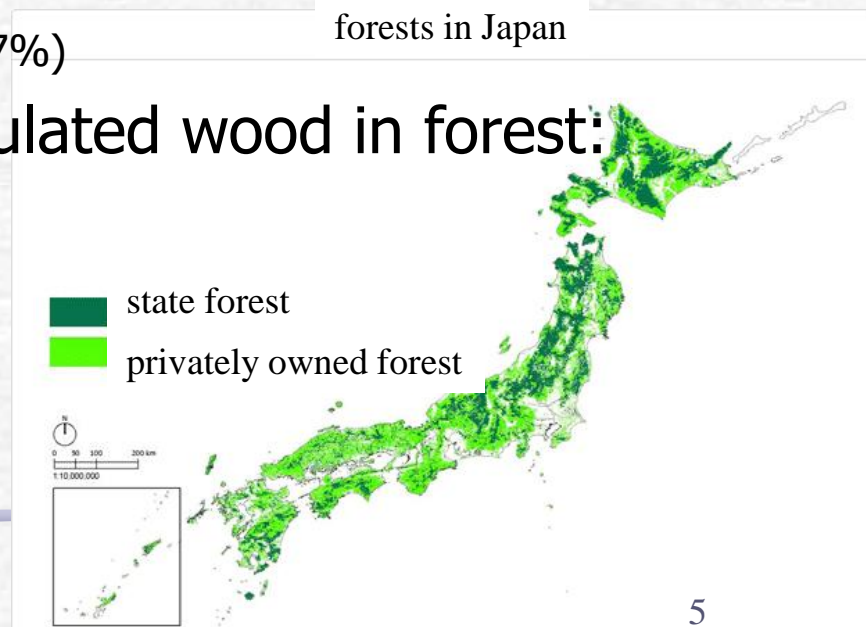
- Petroleum oil dependency of transport energy will be brought down from 100% to 80% till 2030.... → Development of substitutive Bio-fuel

☞ About 68.5% of land area of Japan is forest:

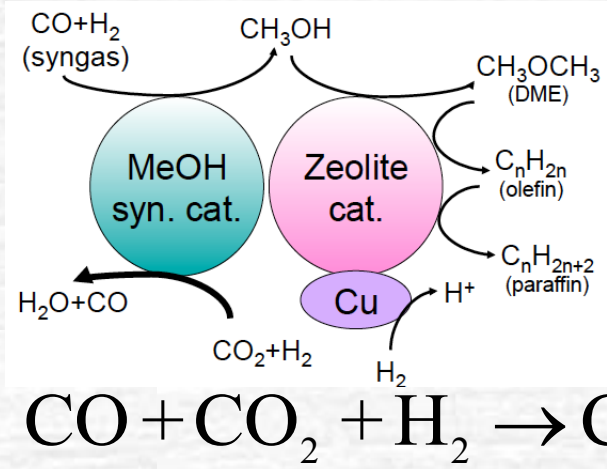
- forest area: 24 979kha
 - Total land area: 36 450kha
 - 3rd highest in the world
 - 1st: Republic of Finland (72.9%)
 - 2nd: the Kingdom of Sweden (68.7%)

- Estimated amounts of accumulated wood in forest: 4432Mm³

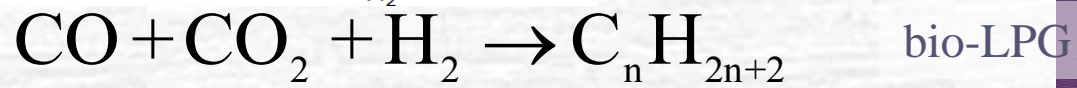
In Japan, forest residues and agricultural wastes have great potential as resources of fuel materials....(but cost for collection?)



National (NEDO) Project: Synthesis of LPG from Biomass-Derived Syngas (FY2010-2013)



direct LPG synthesis
hybrid catalyst



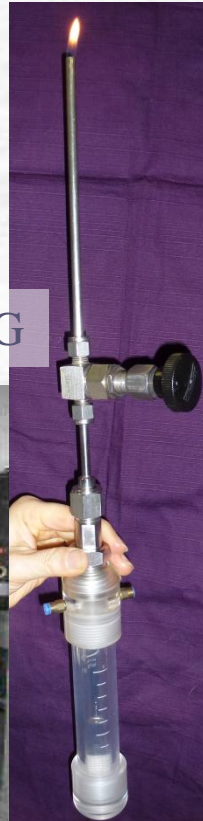
Lab-scale gasifier(AIST)



Composition control for catalytic synthesis



Lab-scale LPG synthesis reactor (Japan Synthesis Gas/University of Kitakyushu)



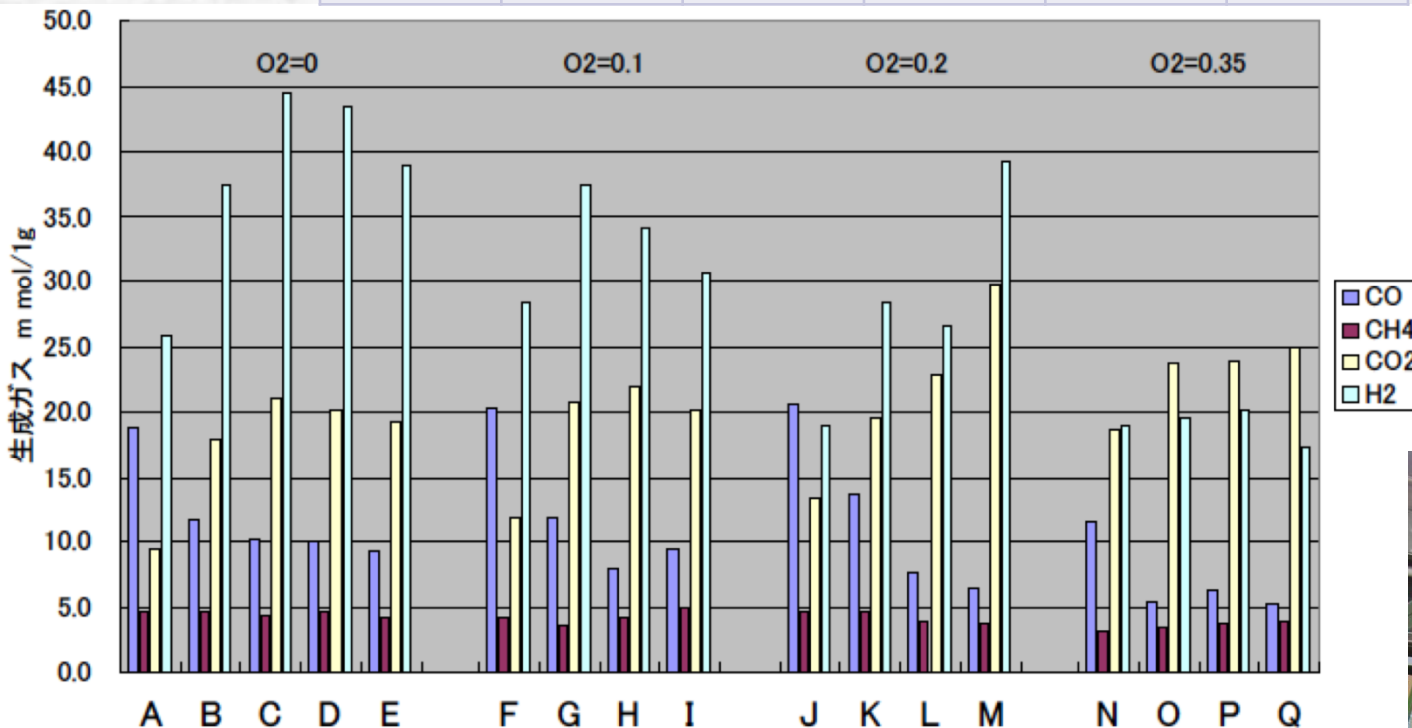
Gas properties obtained from lab-scale gasifier



C	H	O	N	S	Ash
48.6%	6.1%	45.1%	0.0%	0.0%	0.3%



Japanese cedar wood chips



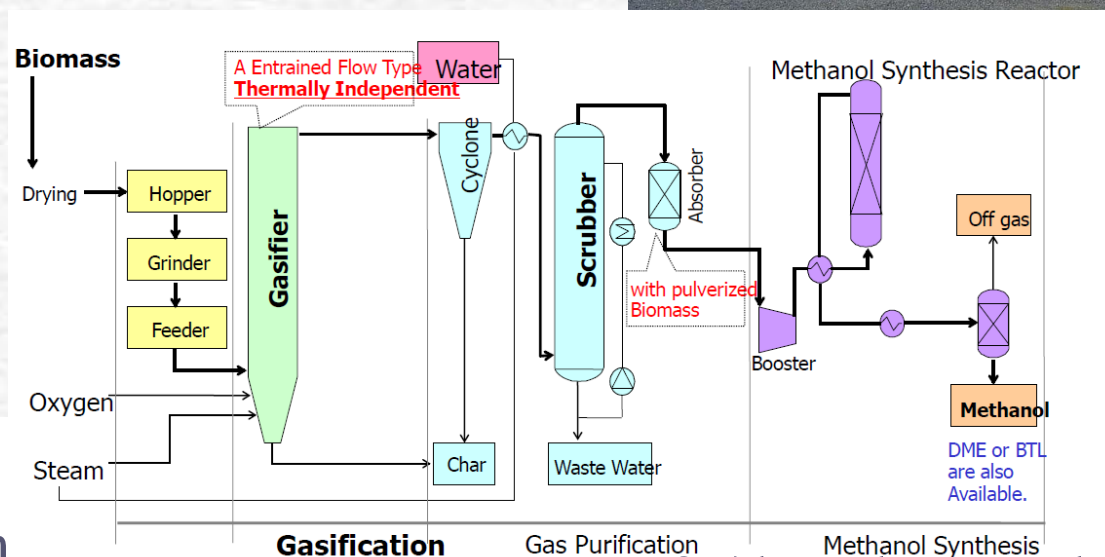
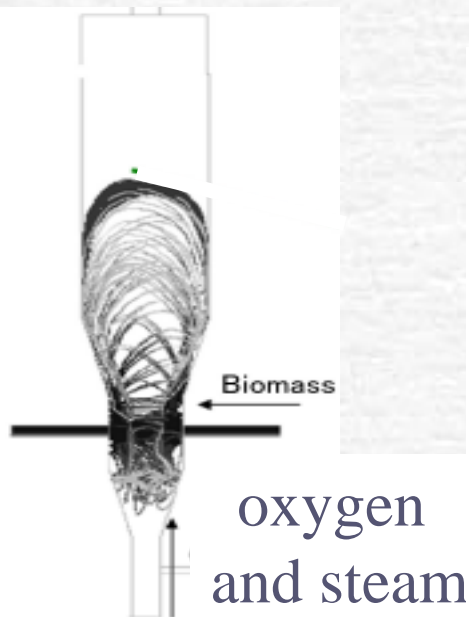
		H ₂ O/C				
		0.5	2.5	5	10	
O ₂ /C	0	A	B	C	D	
	0.1	F	G	H	I	
	0.2	J	K	L	M	
	0.35	N	O	P	Q	



Entrained-flow Biomass Gasification



- 100-200t/day-scale commercial plant (plan)
- Cold gas efficiency is 75% or higher.
- 0.1%-tar and 1%-char yields, when operating recycle systems
- H₂, CO, CO₂ contents are highly controllable by adjusting H₂O, O₂ and biomass feed rates.
 - H₂ increases when increasing H₂O.
 - H₂ and CO decrease when increasing O₂.
 - CH₄ content: 8-10%



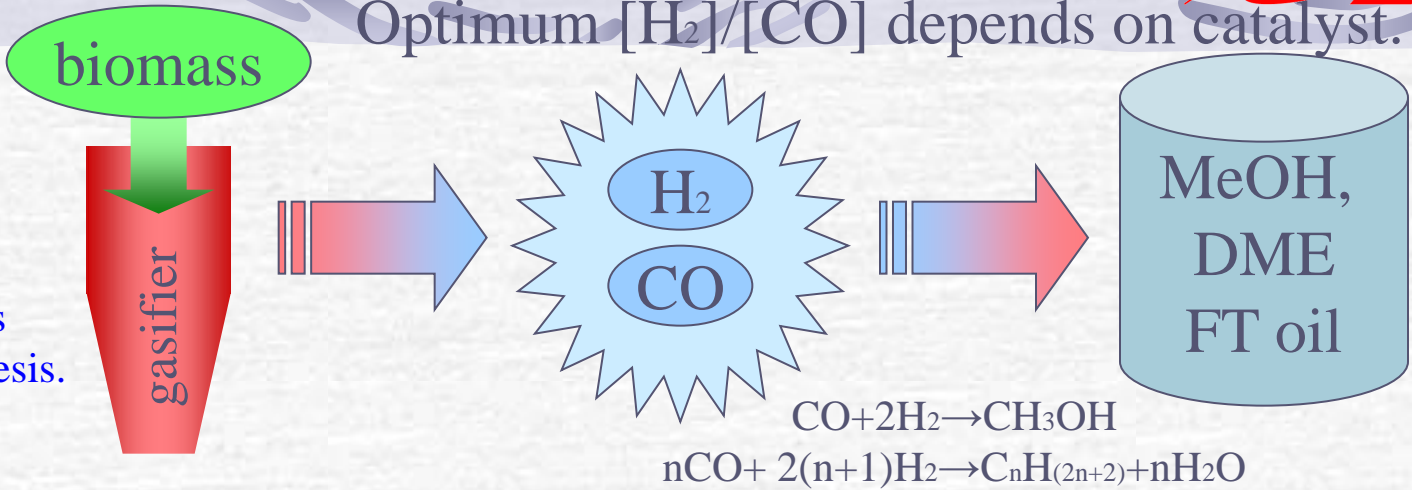
2t/d-scale test plant (MH&)



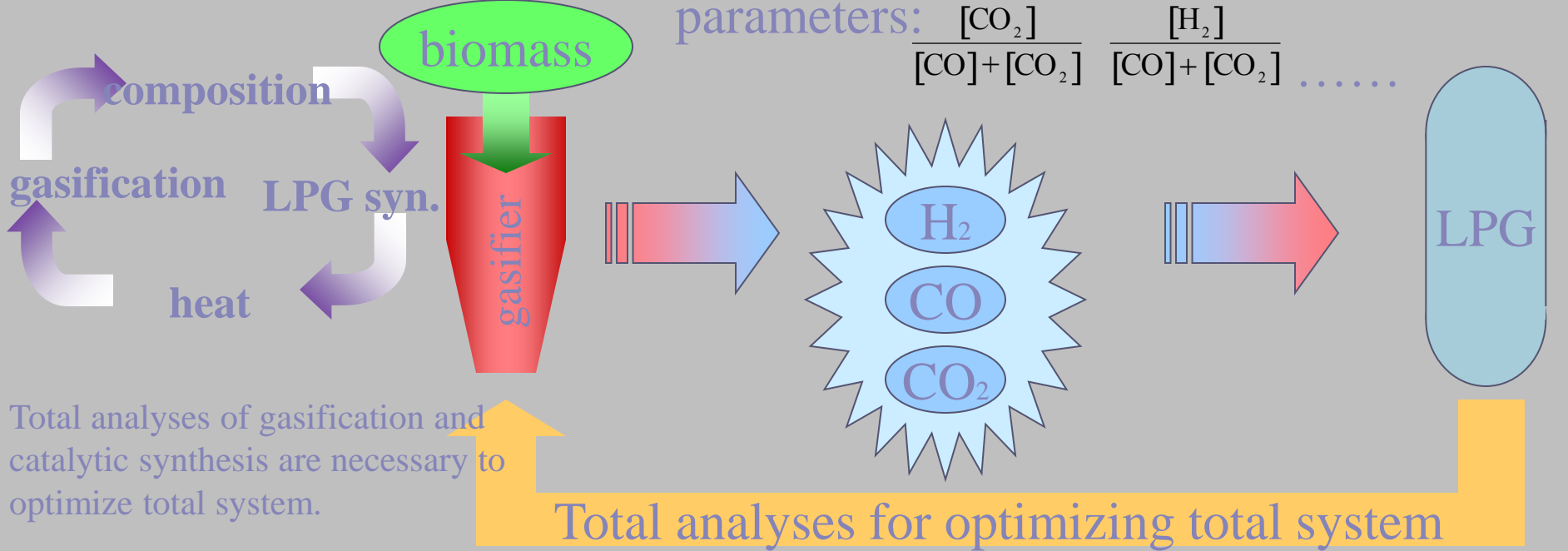
Ordinal BTL (MeOH, DME, FT oil) process

Optimum $[H_2]/[CO]$ depends on catalyst.

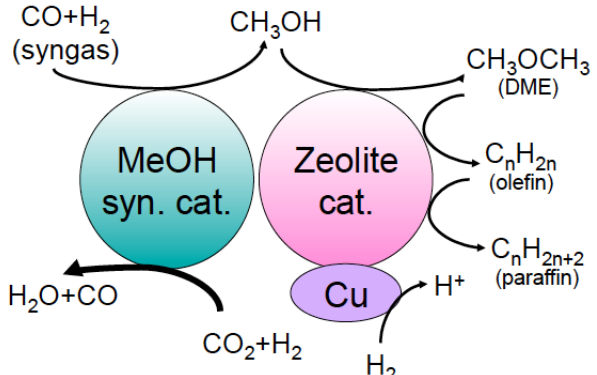
Purpose of gasification:
To obtain gas compositions
suitable for catalytic synthesis.



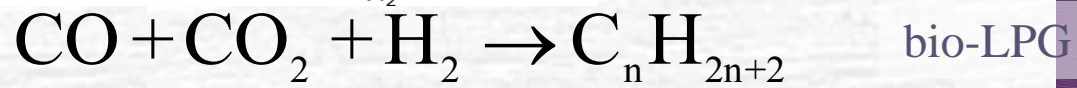
Bio LPG synthesis process



National (NEDO) Project: Synthesis of LPG from Biomass-Derived Syngas (FY2010-2013)



direct LPG synthesis
hybrid catalyst



Lab-scale gasifier(AIST)



bio-Syngas (CO , CO_2 , H_2)
Composition control for catalytic synthesis



Lab-scale LPG synthesis reactor (Japan Synthesis Gas/University of Kitakyushu)



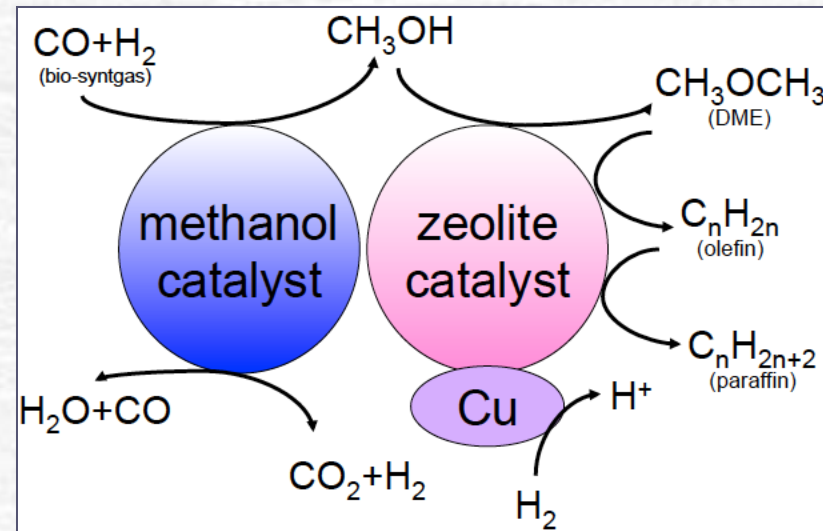
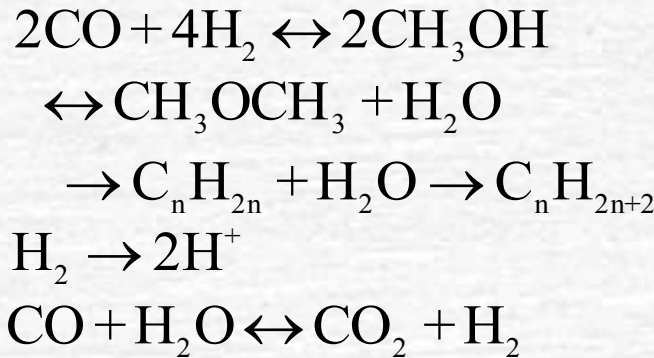
direct LPG synthesis catalyst



- a hybrid of methanol and zeolite catalysts
- One-path process
- 80%-CO conversion rate at moderate conditions
 - 260°C and 2.0MPa
- Bio-LPG is synthesized from H_2 , CO and CO_2

} Simple and low-cost reactor

- Reversible and irreversible reactions proceed simultaneously.



• Synthesis properties depend on H_2 , CO, CO_2 contents.

• optimum conditions:

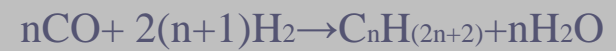
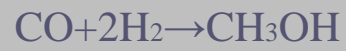
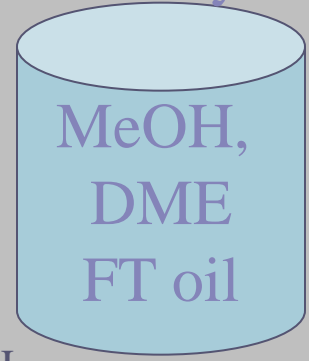
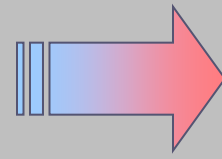
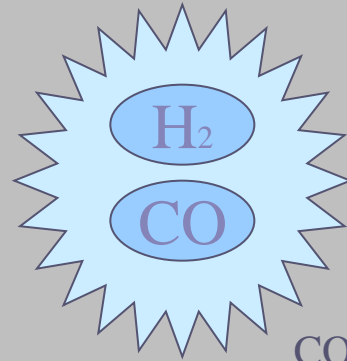
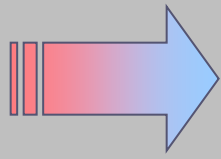
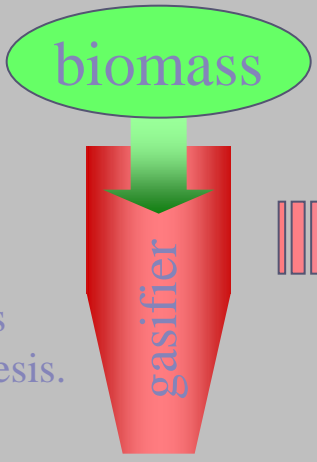
$$\frac{[H_2]}{[CO] + [CO_2]} \geq 2.5 \quad 0.1 \leq \frac{[CO_2]}{[CO] + [CO_2]} \leq 0.4$$



Ordinal BTL (MeOH, DME, FT oil) process

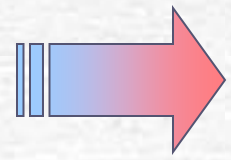
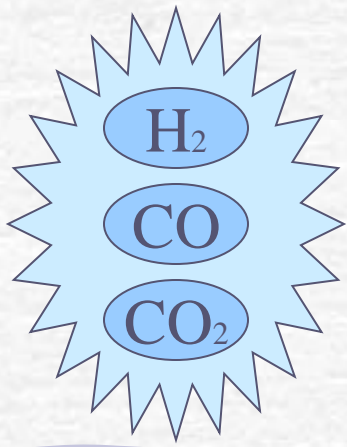
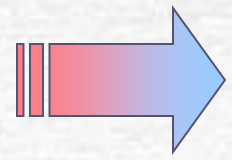
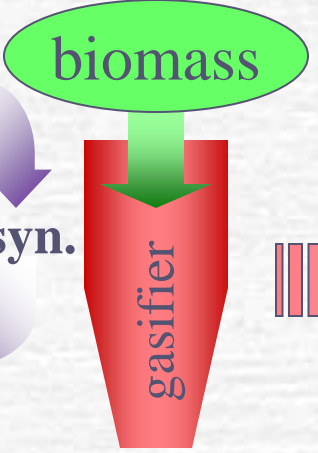
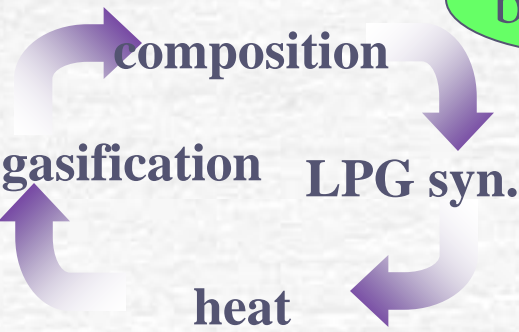
Optimum [H₂]/[CO] depends on catalyst.

Purpose of gasification:
To obtain gas compositions
suitable for catalytic synthesis.



Bio LPG synthesis process

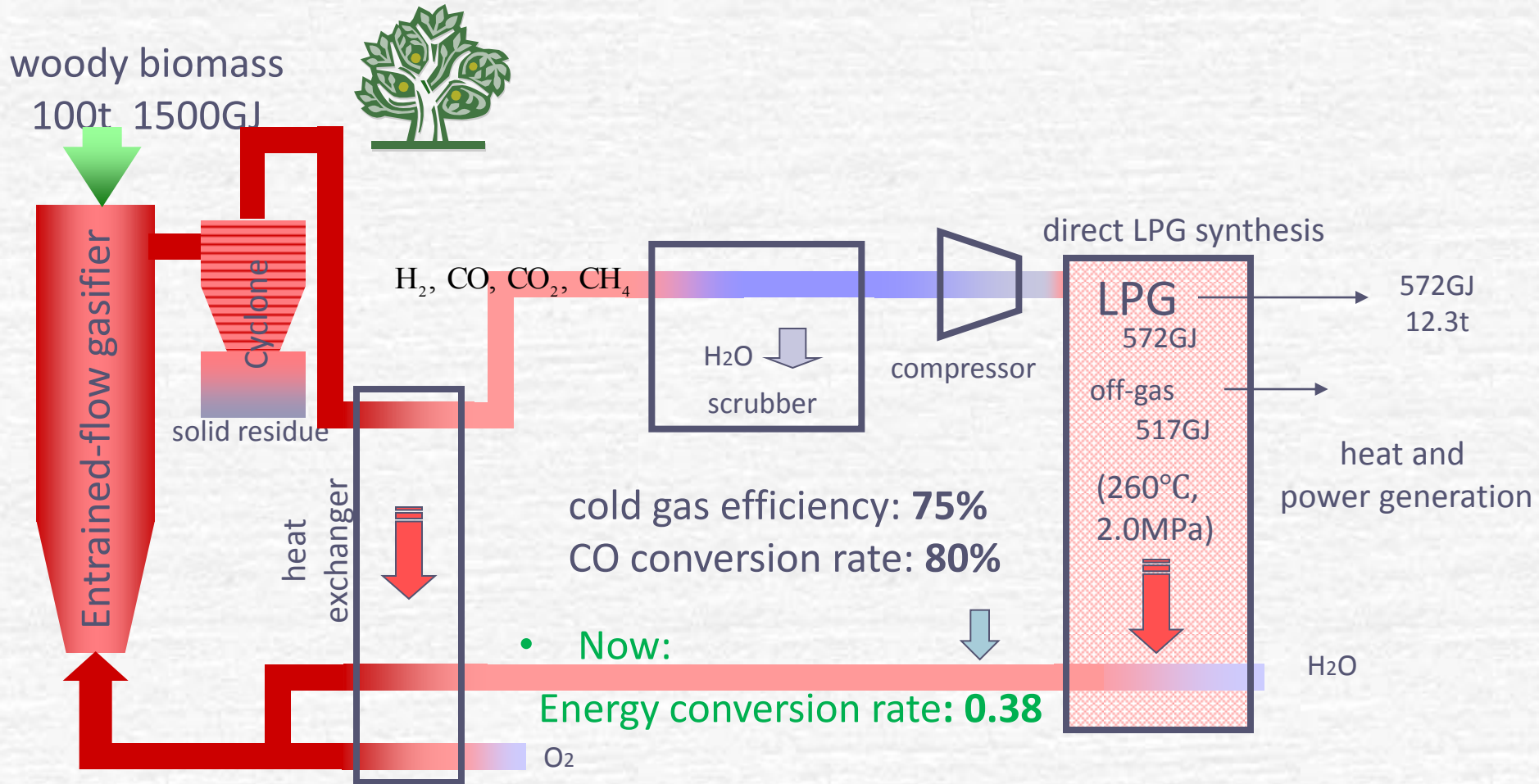
parameters: $\frac{[\text{CO}_2]}{[\text{CO}] + [\text{CO}_2]}$ $\frac{[\text{H}_2]}{[\text{CO}] + [\text{CO}_2]}$



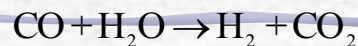
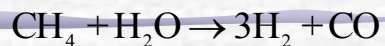
Total analyses of gasification and
catalytic synthesis are necessary to
optimize total system.

Total analyses for optimizing total system

Integrated system of gasification and LPG synthesis



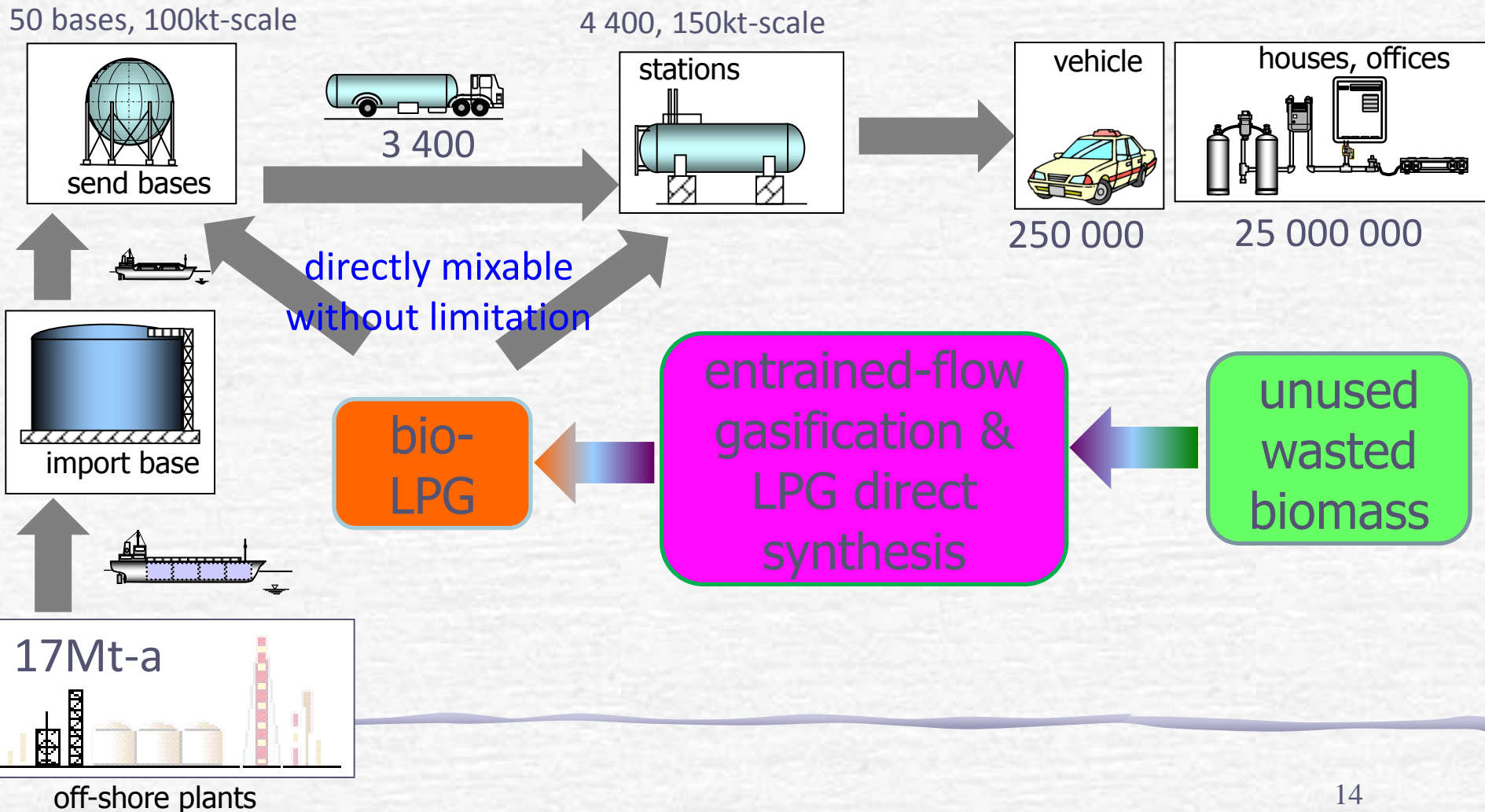
*Catalytic gas reforming (CH₄ reforming & Shift reaction)



LPG supply chain and bio-LPG plant



- Bio-LPG is suitable for present LPG supply chain.
- Bio-LPG is directly mixable without limitation.



Conclusion



System of woody biomass gasification(entrained-flow)-direct catalytic LPG synthesis is a promising process

high cold efficiency gasification & high CO conversion to direct LPG synthesis
(but subjects to be solved.....)

Bio-LPG is suitable for Japanese LPG supply chain and for local product for local consumption.

- Bio-LPG of about 5.8kt is producible at a 100t/d-scale plant in each year.
 - 750 million Japanese yen (>9 million US dollar).
 - 130 000 yen/t-LPG.
- Bio-LPG of about 288kt is producible if 50 LPG bases have own plants.
 - about 2% of LPG consumption.



Subjects of the further study

- catalytic gas reforming to increase H₂ and CO
- low cost ultra clean-up
- to acquire engineering data of LPG synthesis reactor
- effective recovery of heat

Acknowledgement



- This research has been performed and funded by NEDO biomass project.
- Authors thank to Dr. Takeno and Dr. Matsumoto (Mitsubishi Heavy Industries Ltd.,)for variable discussion about biomass gasification.



Thank you for your
kind attention!!

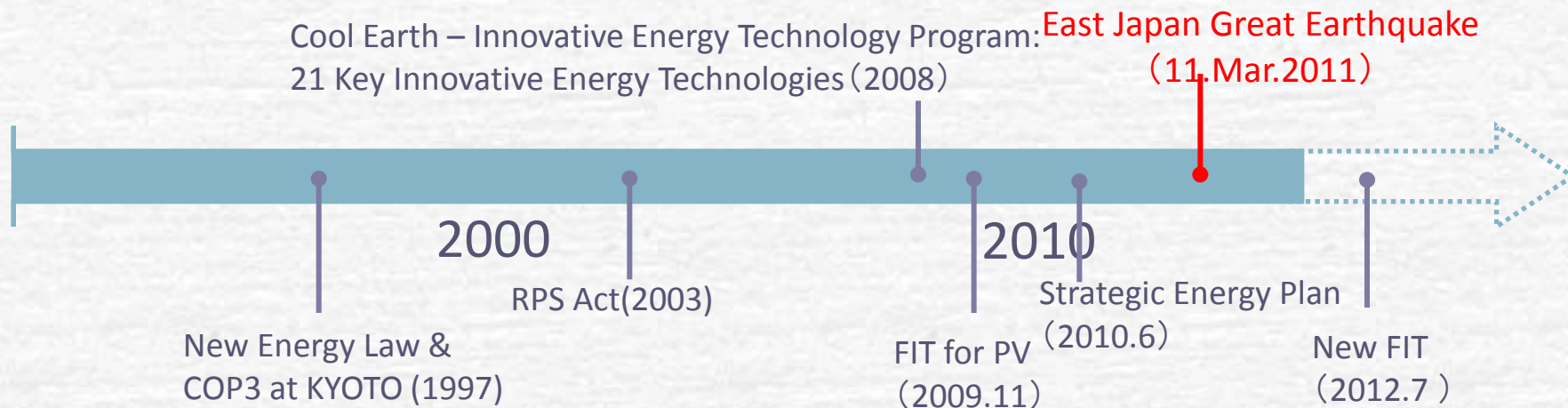


Before 3.11/ After 3.11discontinuity

Before 3.11

Review of Basic Energy Plan (June 2010) ---

**--Enforcement of introduction of Nuclear and renewable energy
Nuclear (& hydraulic, “zero emission” energy) ⇒ ~50%**



After 3.11

Again, Review of Basic Energy Plan ---

**--Enforcement of introduction of renewable energy,
especially solar and wind ?**

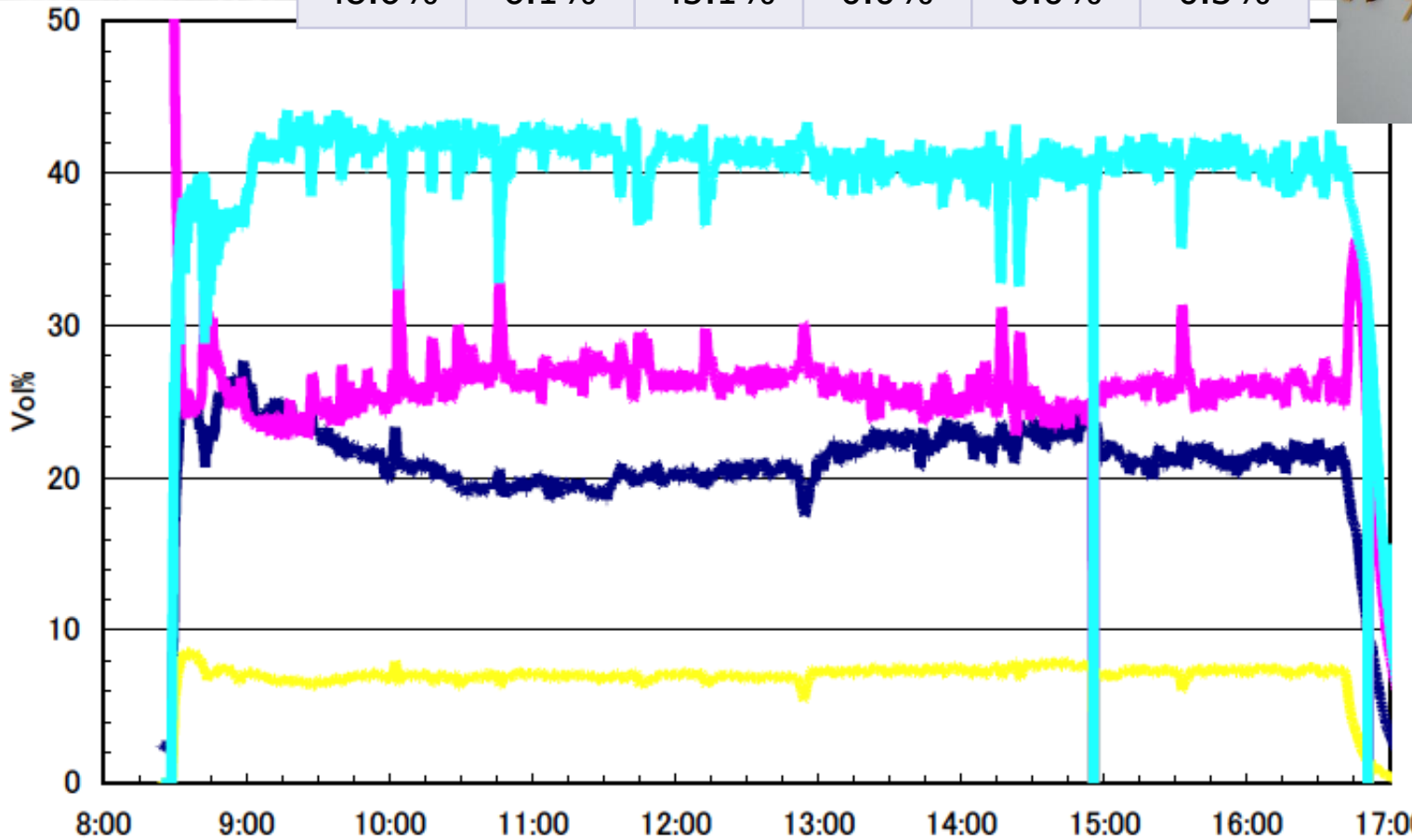
⇒ now under discussion...

Gas compositions of lab-scale gasifier



Japanese cedar wood chips

C	H	O	N	S	Ash
48.6%	6.1%	45.1%	0.0%	0.0%	0.3%





Integrated system of gasification and LPG synthesis

Because amount of recoverable/usable heat is limited, H₂O and O₂ feed rates (gasification agent) are limited.



woody biomass
100t 1500GJ



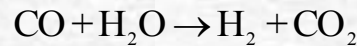
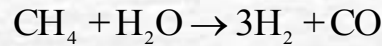
Practical gas compositions

$$\frac{[H_2]}{[CO]+[CO_2]} \leq 1.5 \quad \frac{[CO_2]}{[CO]+[CO_2]} \geq 0.5$$

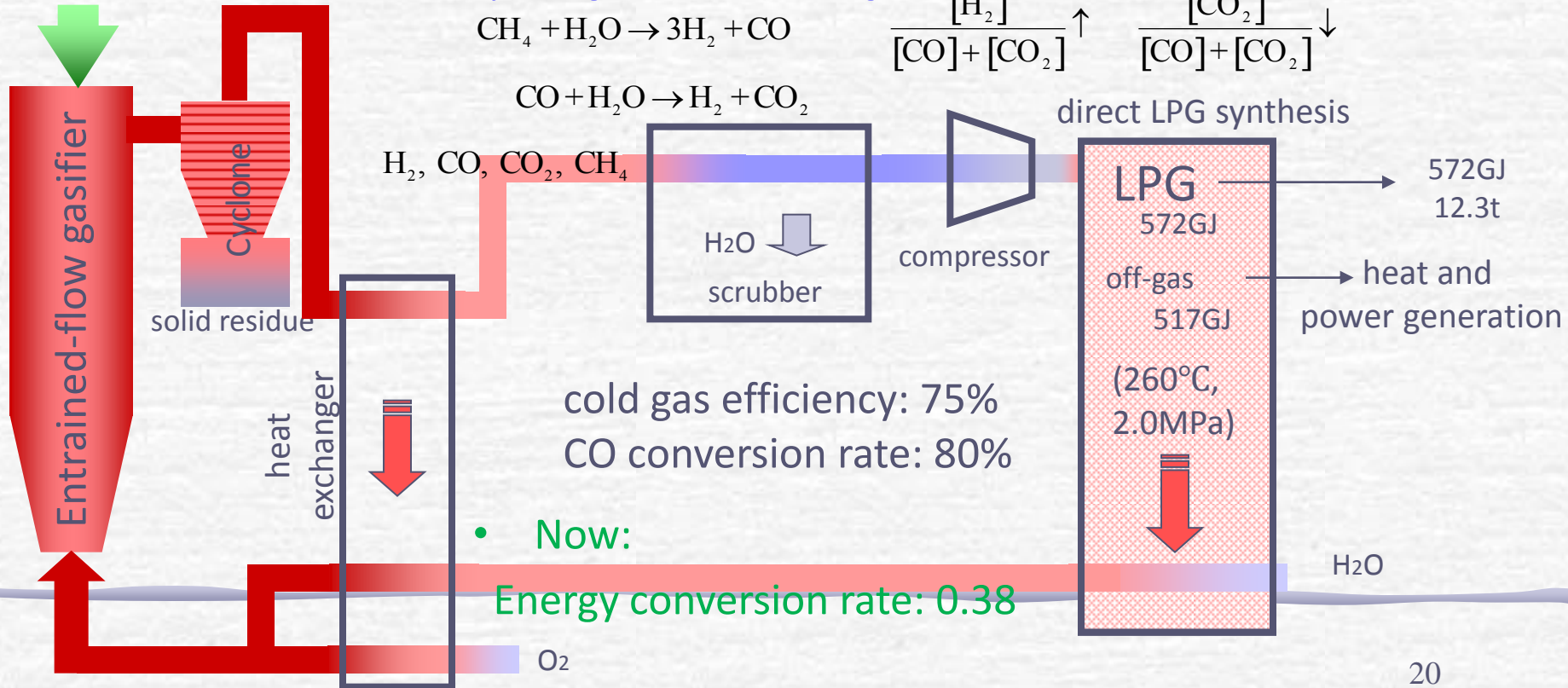
Optimum gas compositions for LPG synthesis

$$\frac{[H_2]}{[CO]+[CO_2]} \geq 2.5 \quad 0.1 \leq \frac{[CO_2]}{[CO]+[CO_2]} \leq 0.4$$

Catalytic gas reforming



$$\frac{[H_2]}{[CO]+[CO_2]} \uparrow \quad \frac{[CO_2]}{[CO]+[CO_2]} \downarrow$$



cold gas efficiency: 75%
CO conversion rate: 80%

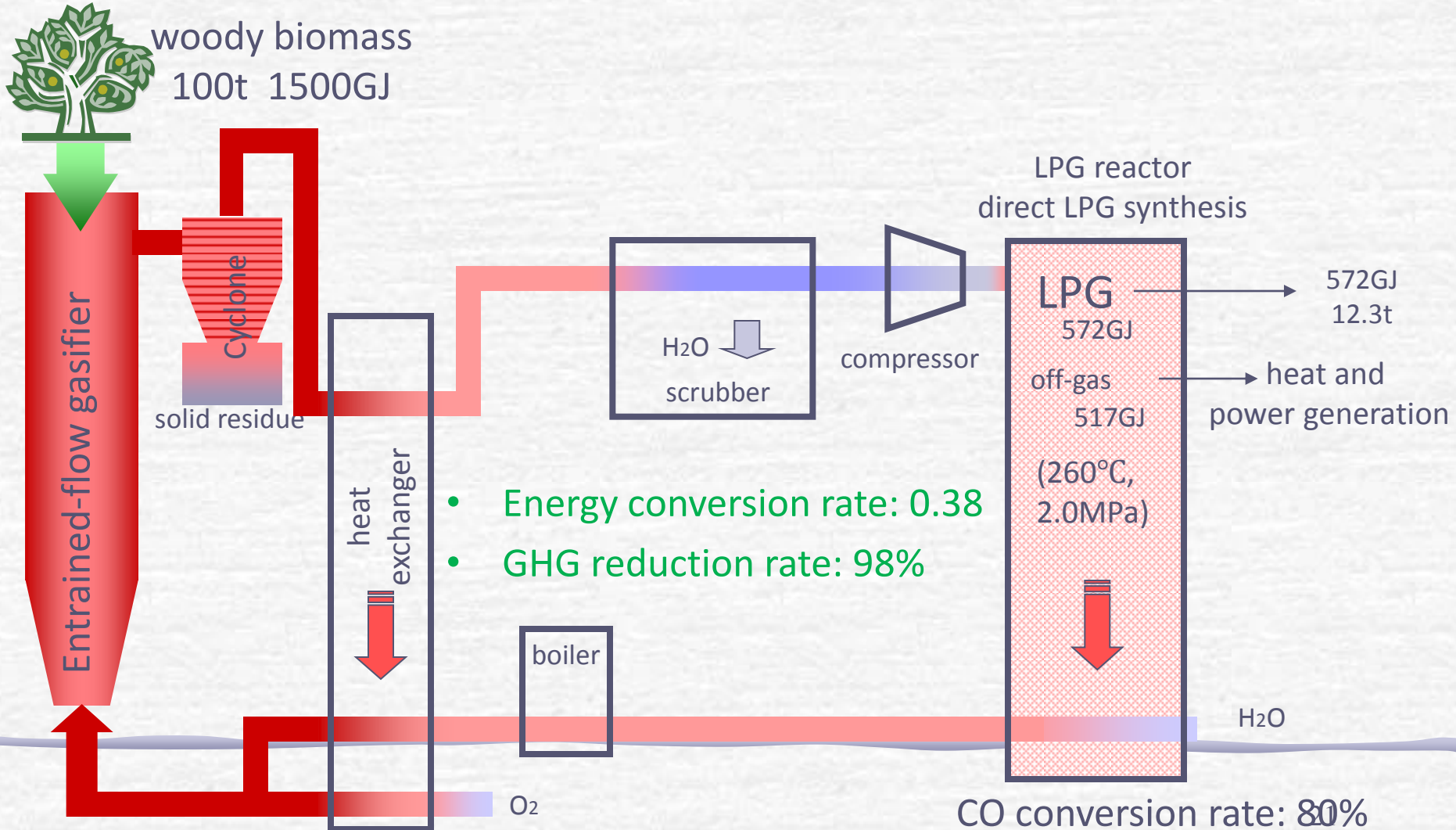
- Now:
Energy conversion rate: 0.38

Integrated system of gasification and LPG synthesis

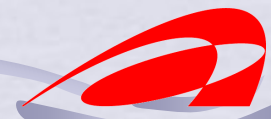
Entrained-flow Biomass gasification

Direct LPG Synthesis

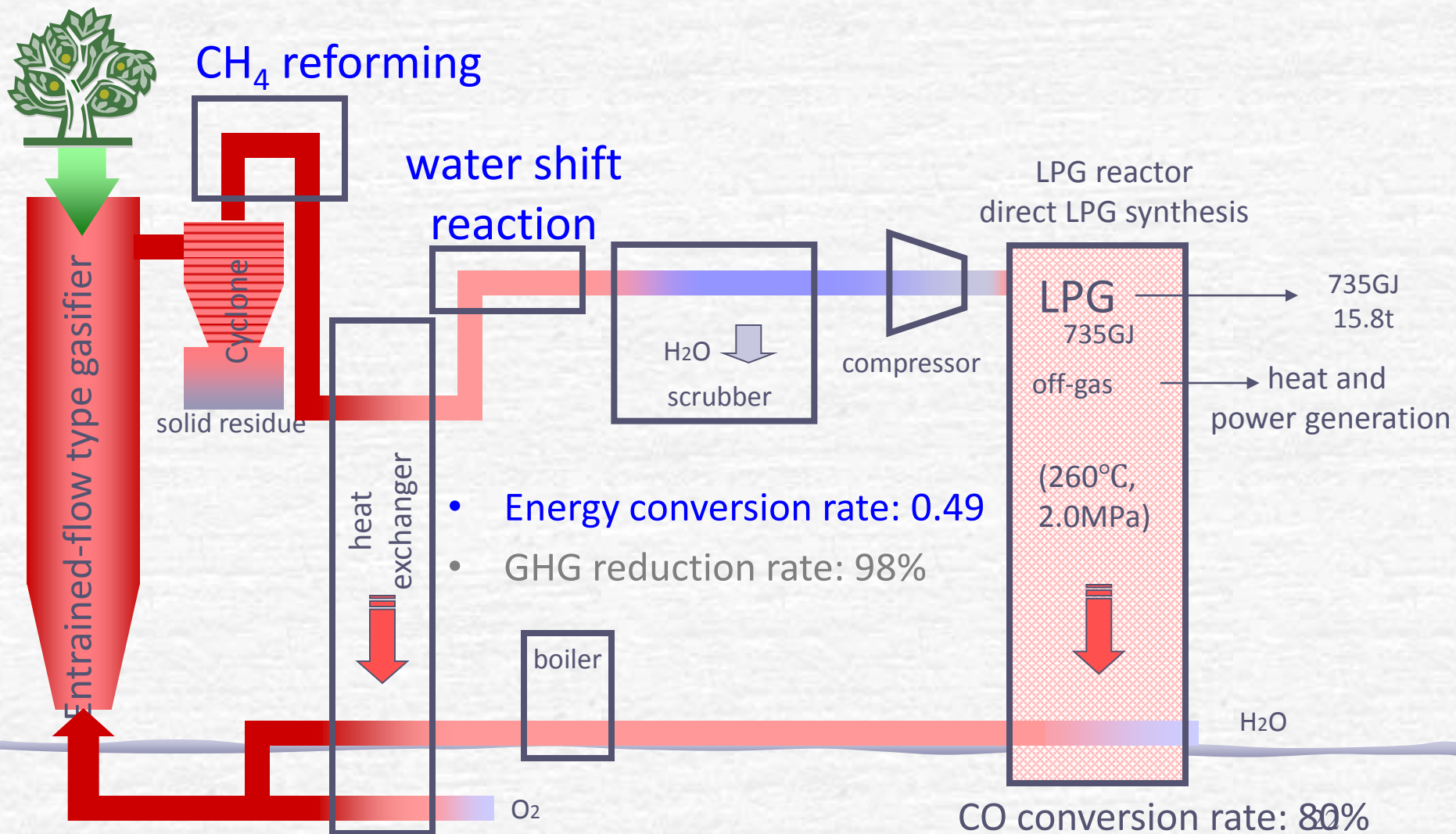
simple, low cost
and high efficiency



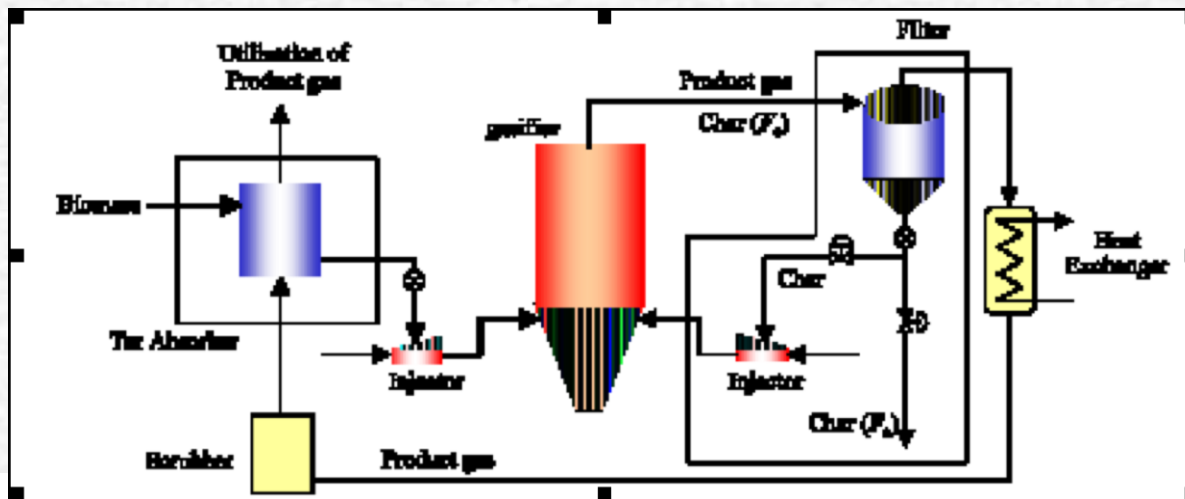
Targets of last 2-year study



- to improve energy conversion rate to 0.49 &
- to acquire engineering data for a commercial plant



Development of High-Efficiency Biomass Gasification Process by Removal and Recycle of By-product

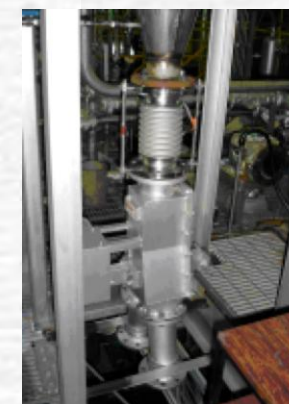


filter



Char yield: 5% -> 1%
 Tar yield: 1% -> 0.1%

Char recycle system and continuous tar absorber combined with biomass feeding devices



branched damper